

Approcci combinatoriali alla scoperta di nuovi materiali

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Fondazione Bruno Kessler
Povo - Trento

Outline

Scienza combinatoriale

Tavola di Mendeleev e possibili combinazioni

Approcci sperimentali

Nuovi materiali e nuove interfacce



MANY AT A TIME Program head Andreas Marzinik (front to back) and lab specialists Raphael Gattlen and Urs Rindisbacher of Novartis Pharma AG, Basel, Switzerland, pipette coupling reagent into 96-well reaction blocks.

COMBINATORIAL CHEMISTRY

Chemical & Engineering News,
August 2001

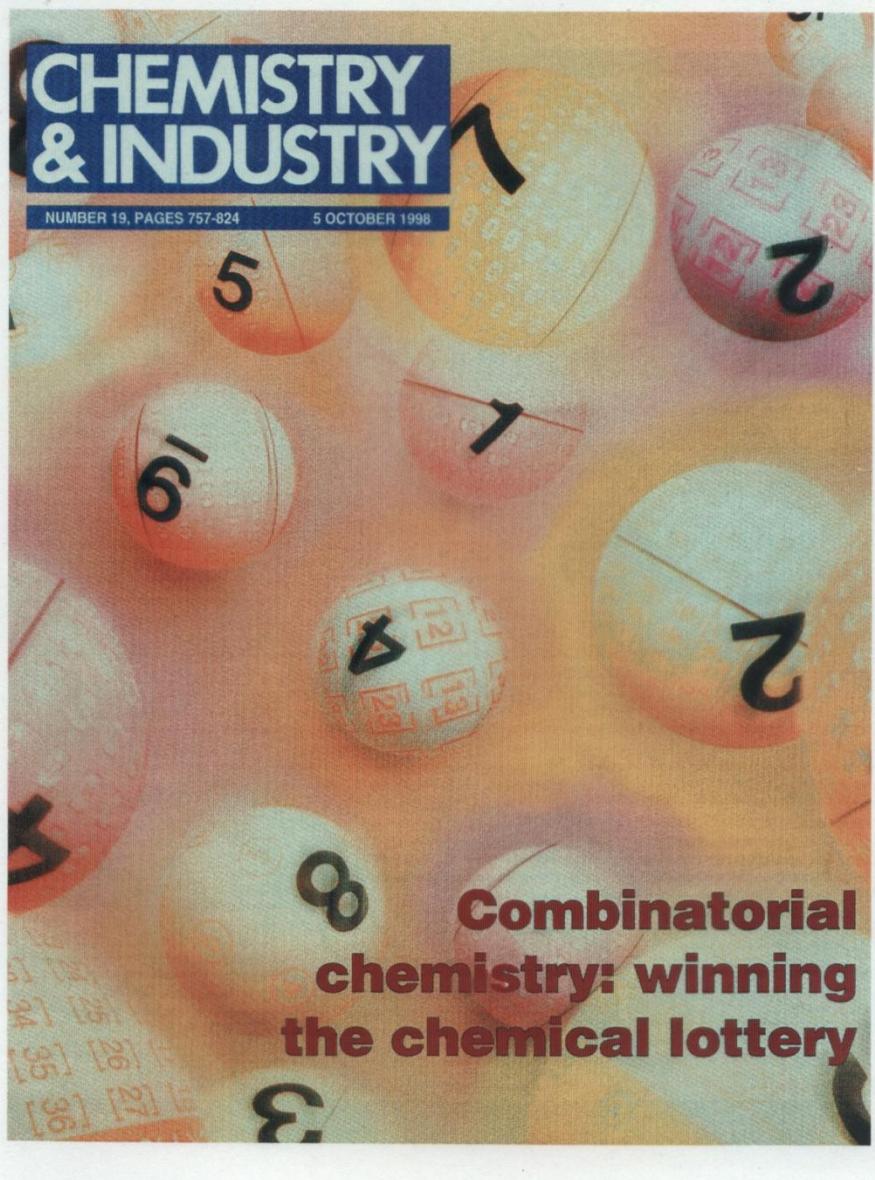
A Combinatorial Approach to Materials Discovery

X.-D. Xiang,* Xiaodong Sun, Gabriel Briceño, Yulin Lou,
Kai-An Wang, Hauyee Chang, William G. Wallace-Freedman,
Sung-Wei Chen, Peter G. Schultz*

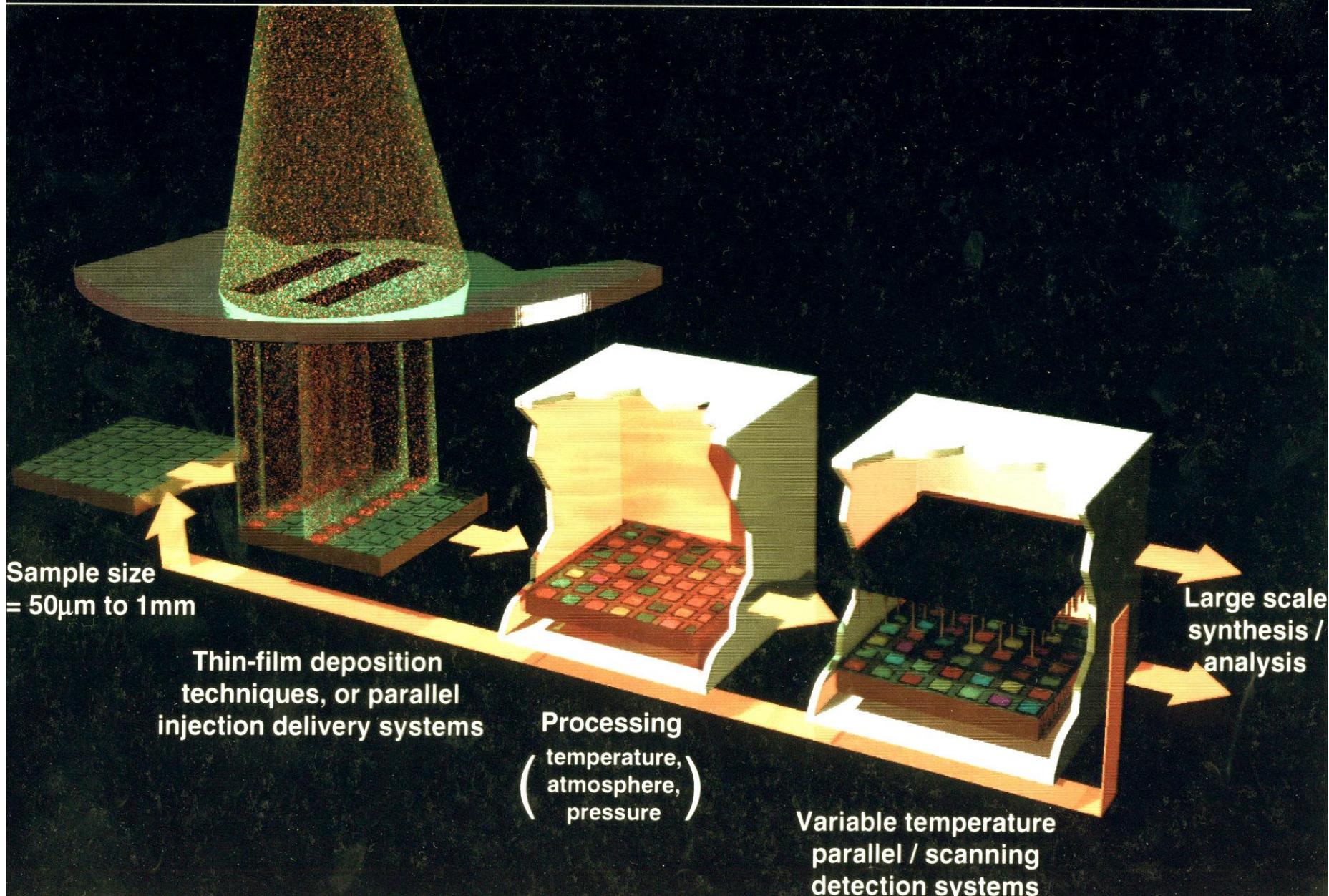
A method that combines thin film deposition and physical masking techniques has been used for the parallel synthesis of spatially addressable libraries of solid-state materials. Arrays containing different combinations, stoichiometries, and deposition sequences of BaCO₃, Bi₂O₃, CaO, CuO, PbO, SrCO₃, and Y₂O₃ were generated with a series of binary masks. The arrays were sintered and BiSrCaCuO and YBaCuO superconducting films were identified. Samples as small as 200 micrometers by 200 micrometers in size were generated, corresponding to library densities of 10,000 sites per square inch. The ability to generate and screen combinatorial libraries of solid-state compounds, when coupled with theory and empirical observations, may significantly increase the rate at which novel electronic, magnetic, and optical materials are discovered and theoretical predictions tested.

Science 268, 1738 (95)

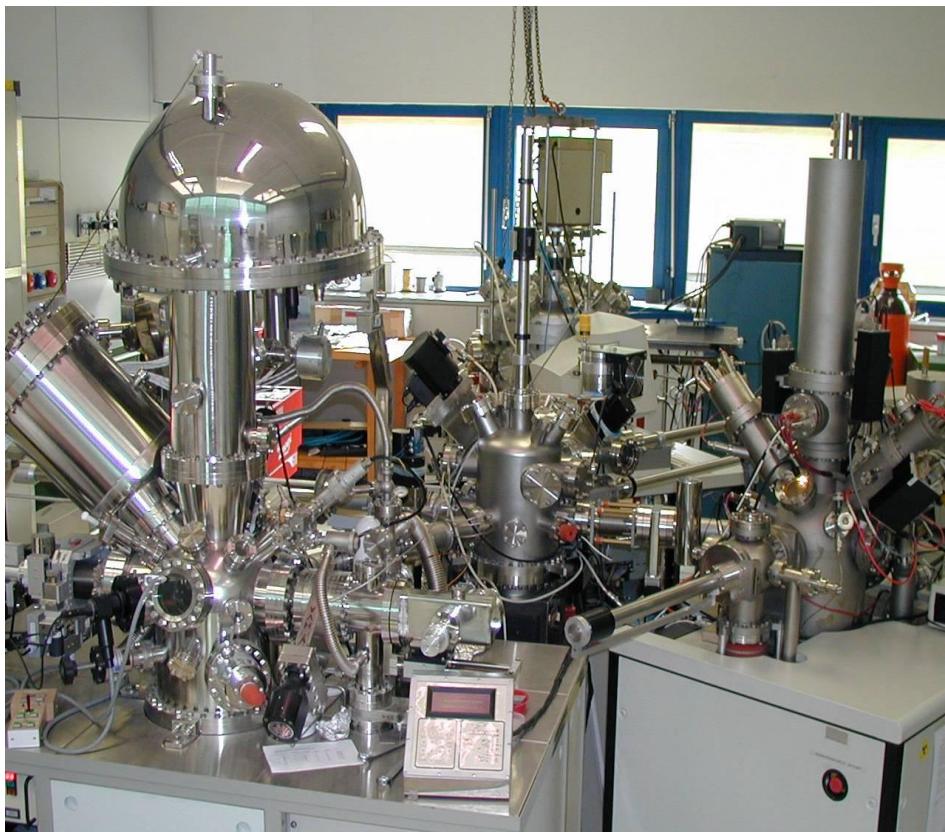
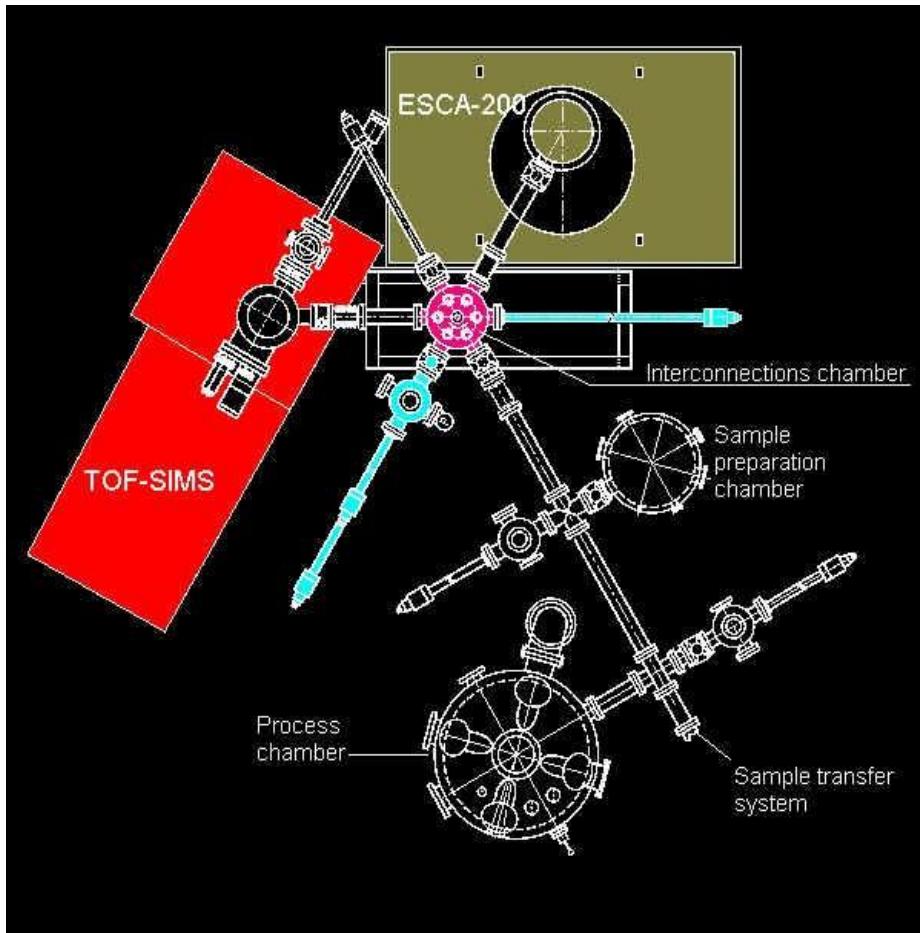




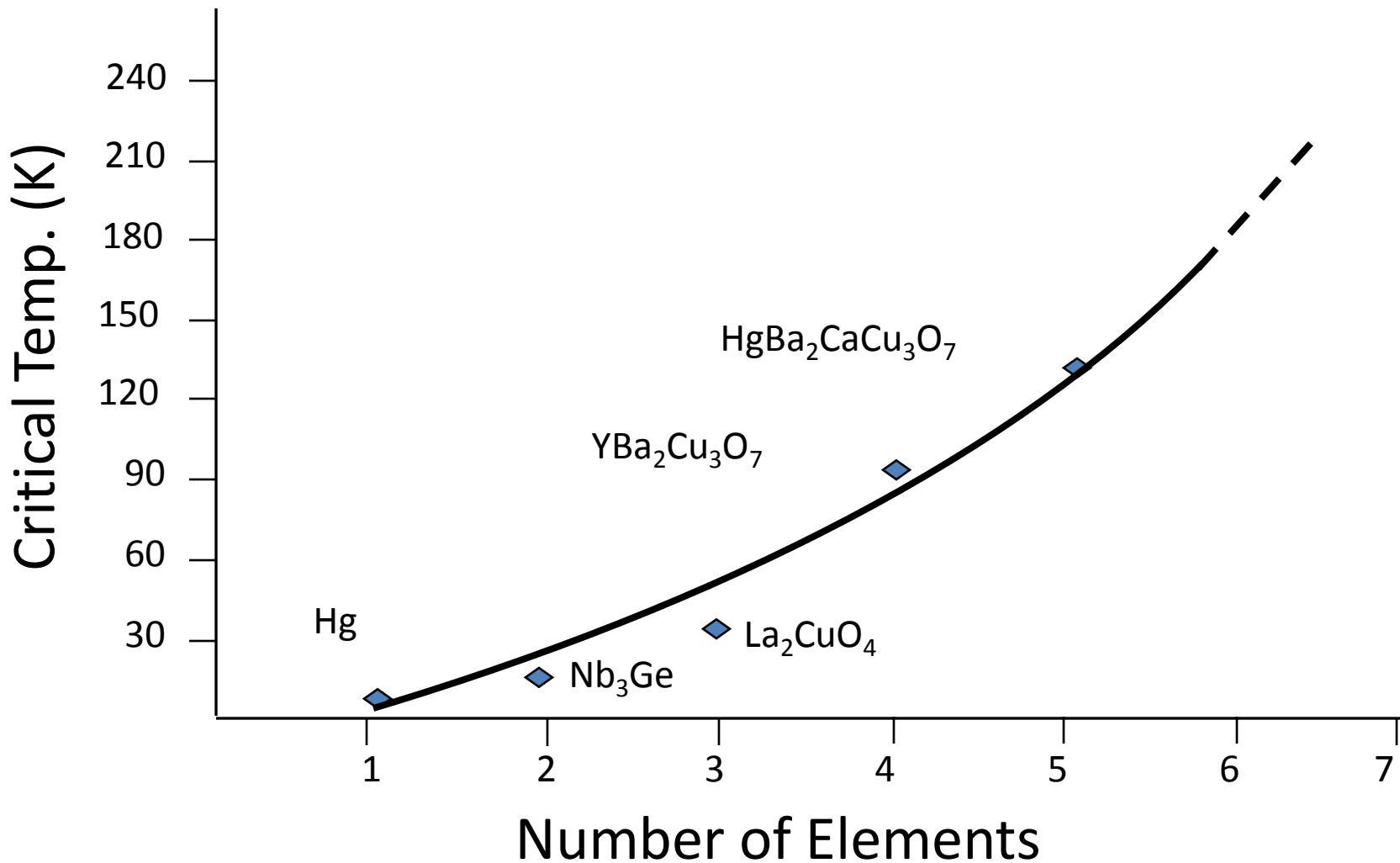
COMBINATORIAL APPROACH TO MATERIALS



Sintesi di materiali tramite approccio combinatoriale in FCS adattando e utilizzando l'apparato già esistente CLUSTER LAB



Correlation between materials complexity and physical properties



How many different compounds are there?

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
H																	He
Li	Be																
Na	Mg																
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Und	Und	Jnh	Uns	Und	Ne	Jun								
87	88	89	104	105	106	107	108	109	110								
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu				
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr				
90	91	92	93	94	95	96	97	98	99	100	101	102	103				

Take 60
“useful” elements.

There are about
30,000 known inorganic
compounds.

Binary compounds have the form AB. (e. g., MgF, SiC, ZnO,.....)
60 x 60 x different combinations: **most are known.**

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Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Und	Und	Und	Unh	Uns	Und	Une	Jun							
87	88	89	104	105	106	107	108	109	110								
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu				
58	59	60	61	62	63	64	65	66	67	68	69	70	71				
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr				
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Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
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60 x 60 x 60 x different combinations: **~3 % of all possible known**

Quaternary compounds have the form ABCD. (YBaCuO, AlNiCOFe,...)
60 x 60 x 60 x 60 x different combinations: **0.01% of all possible known**

How many different compounds are there?

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
H																	He
Li	Be																
Na	Mg																
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Und	Jnd	Jnh	Uns	Und	Jne	Jun								
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60 x 60 x different combinations: **most are known**.

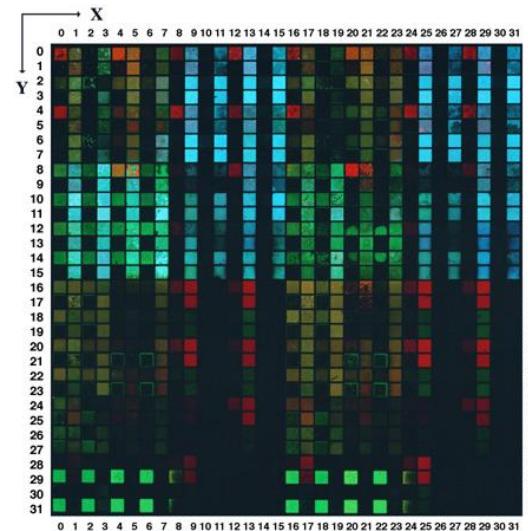
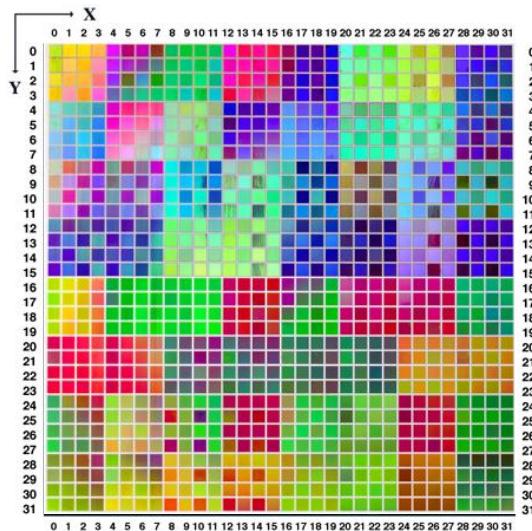
Ternary compounds have the form ABC. (BaTiO, NiMnGa, HgCdTe,...)

60 x 60 x 60 x different combinations: **~3 % of all possible known**

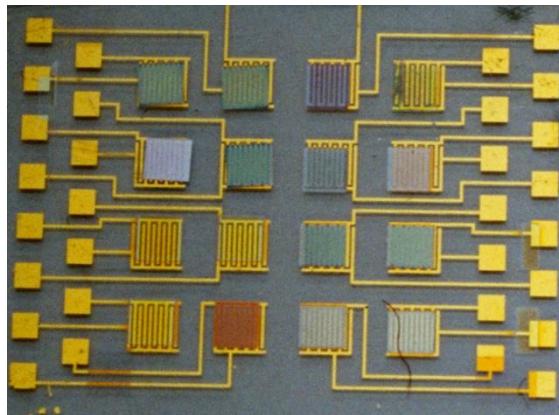
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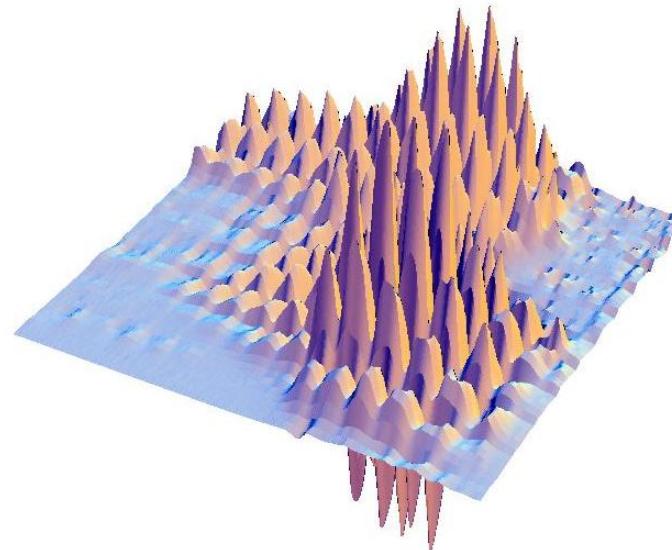
Combinatorial libraries of inorganic materials



Luminescent
materials libraries,
Science **279**,
1712 (1998)



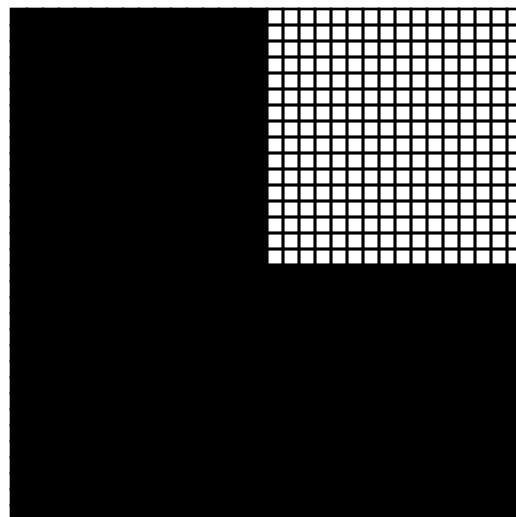
Semiconductor gas sensor library,
“electronic nose”,
Appl. Phys. Lett. **83**, 1255 (2003)



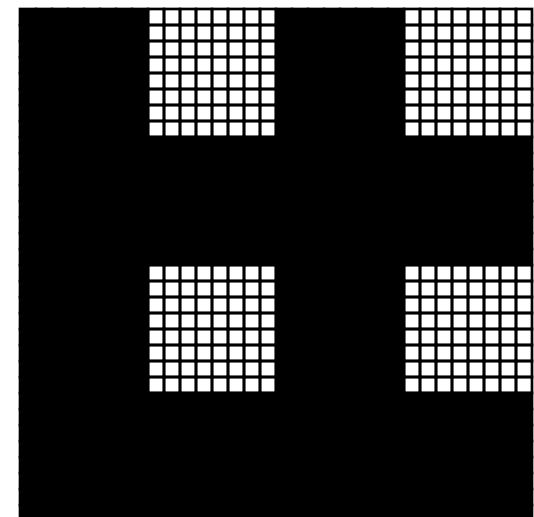
Magnetic shape memory alloy library,
Nature Materials **2**, 180 (2003)

Quaternary Masks

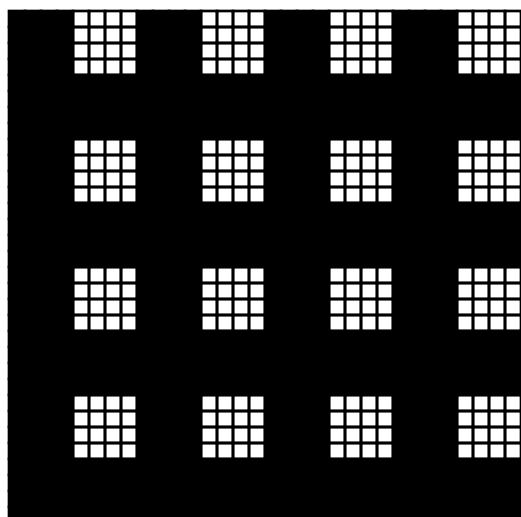
A



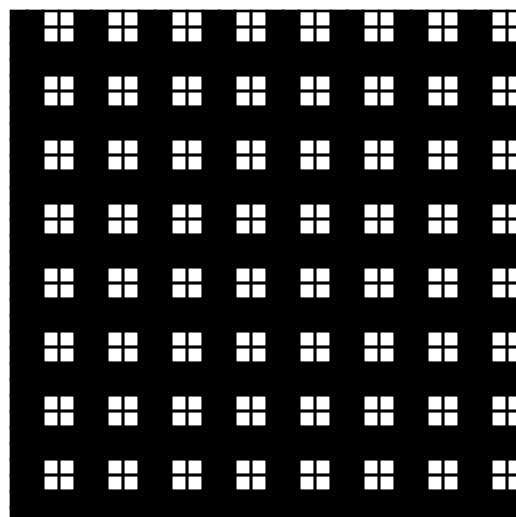
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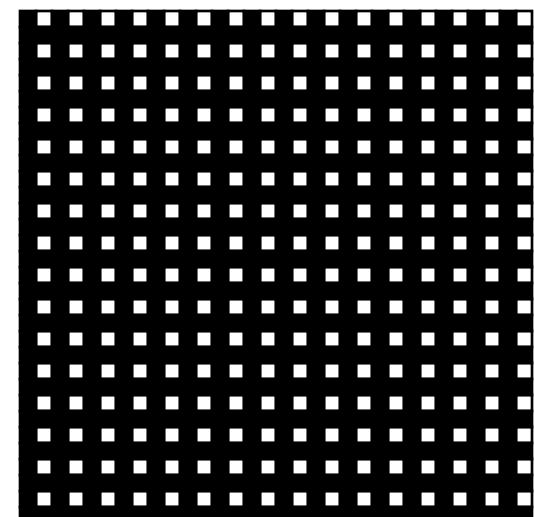
C



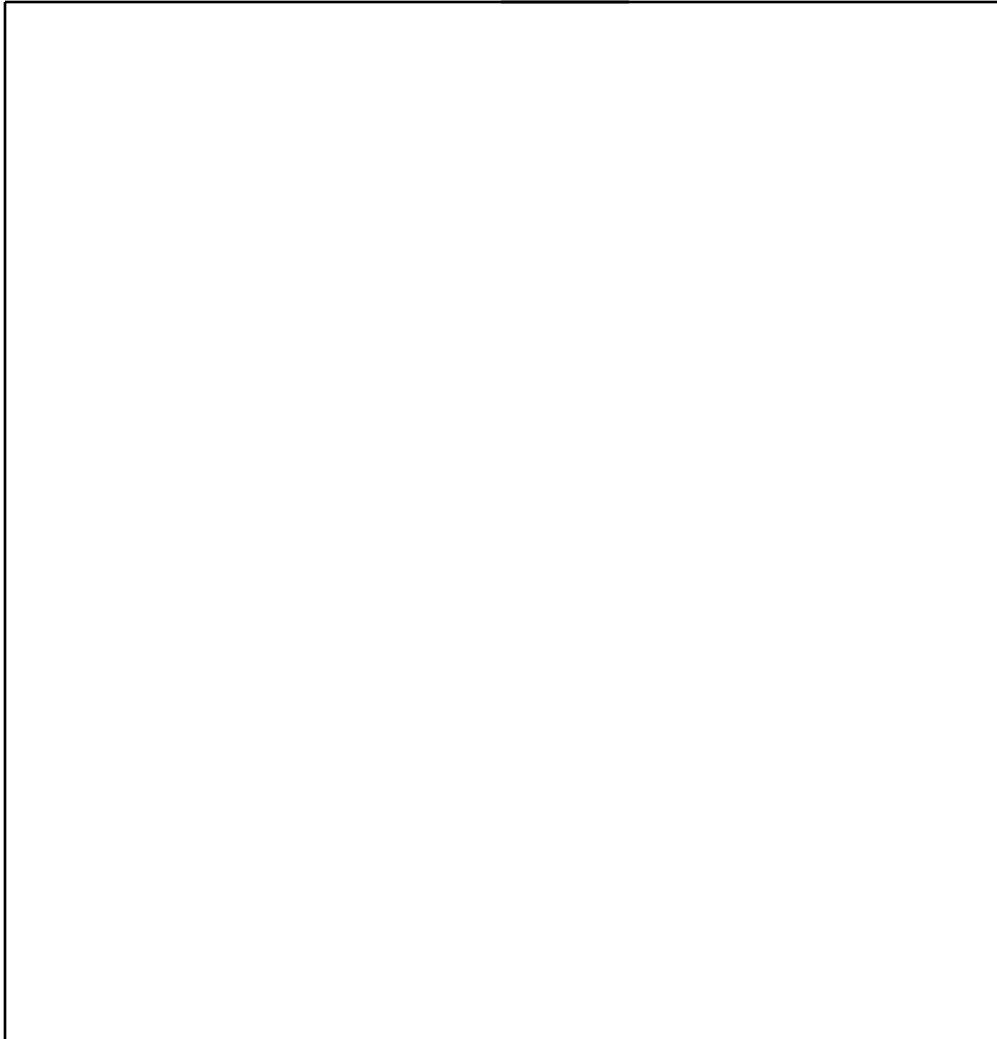
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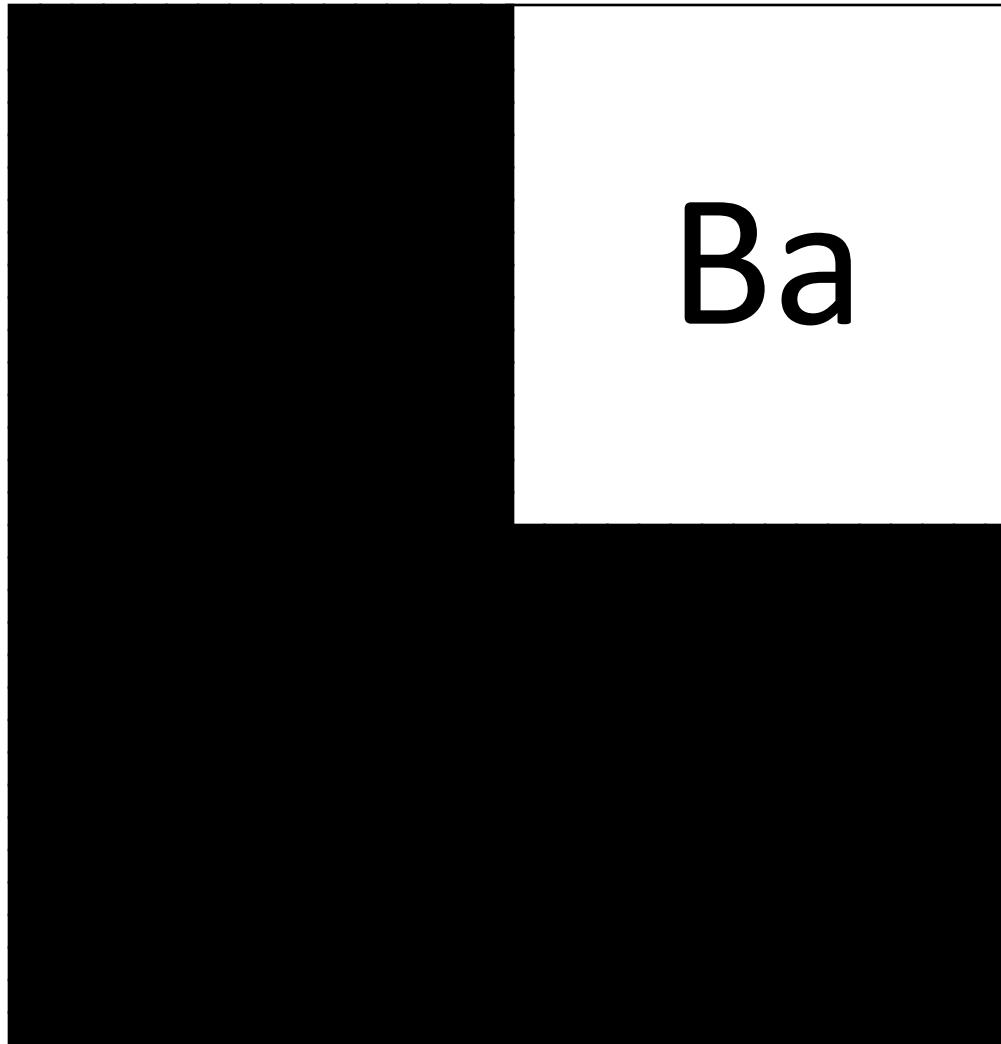
E



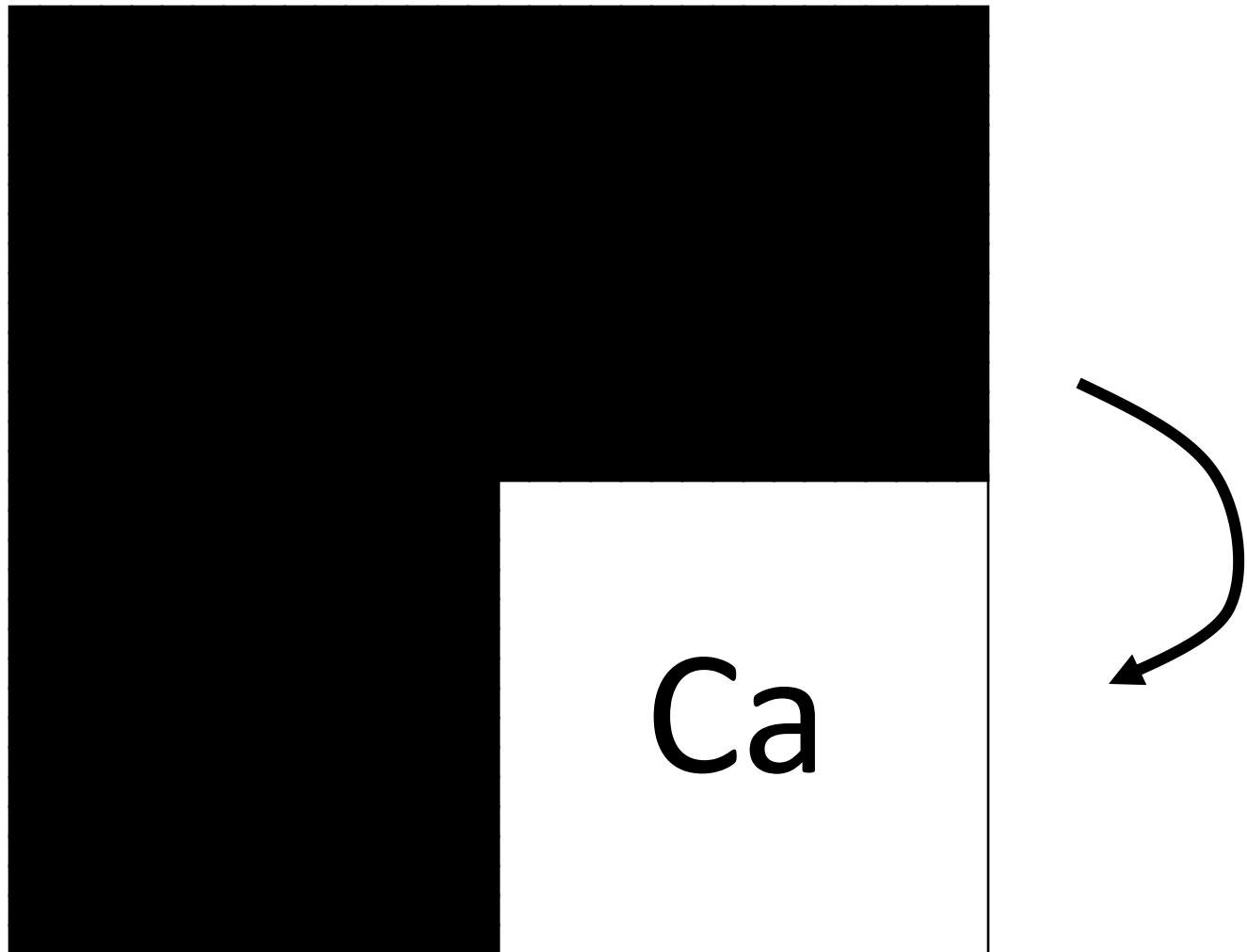
Quaternary Masking



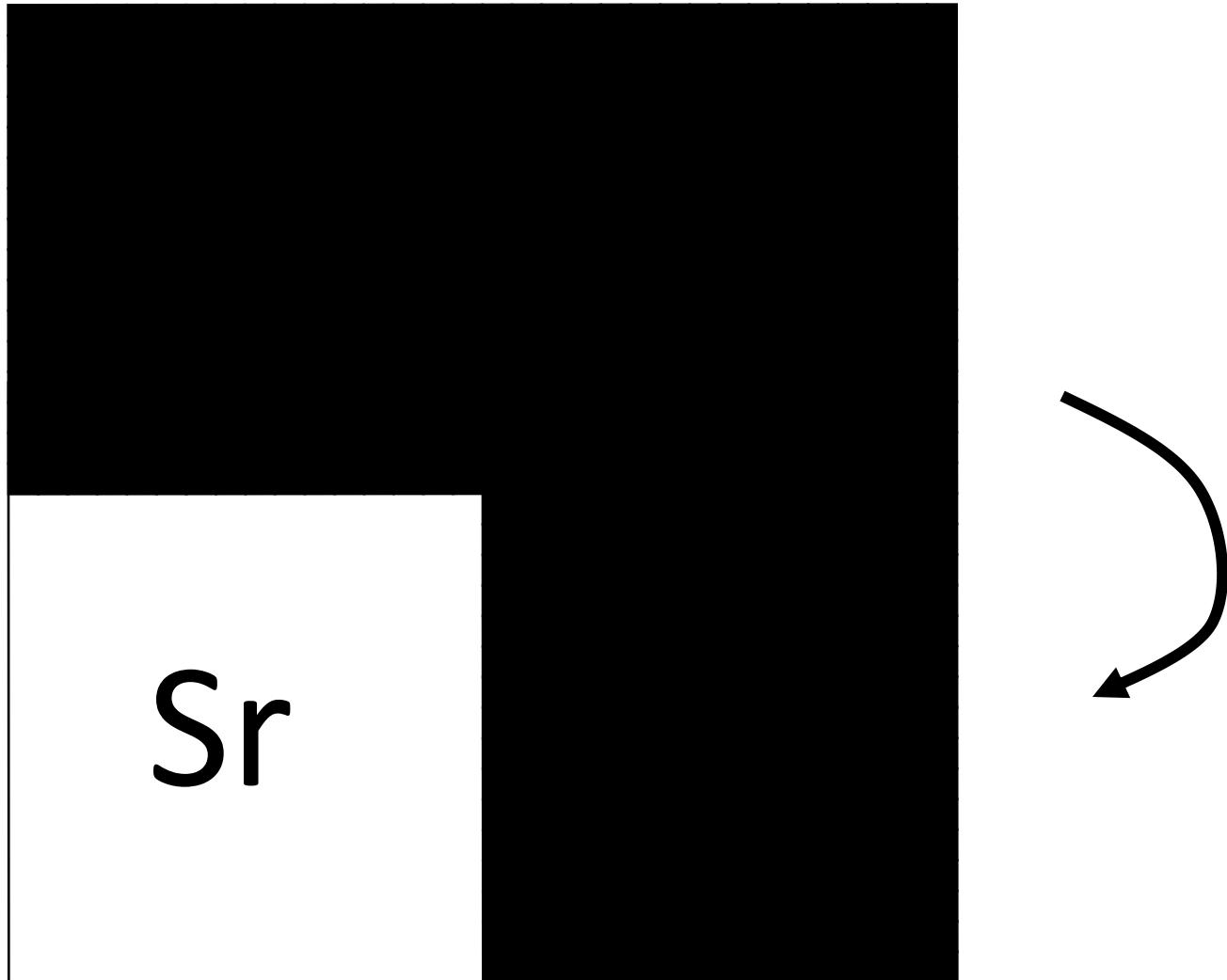
Quaternary Masking: 1st mask, 1st position



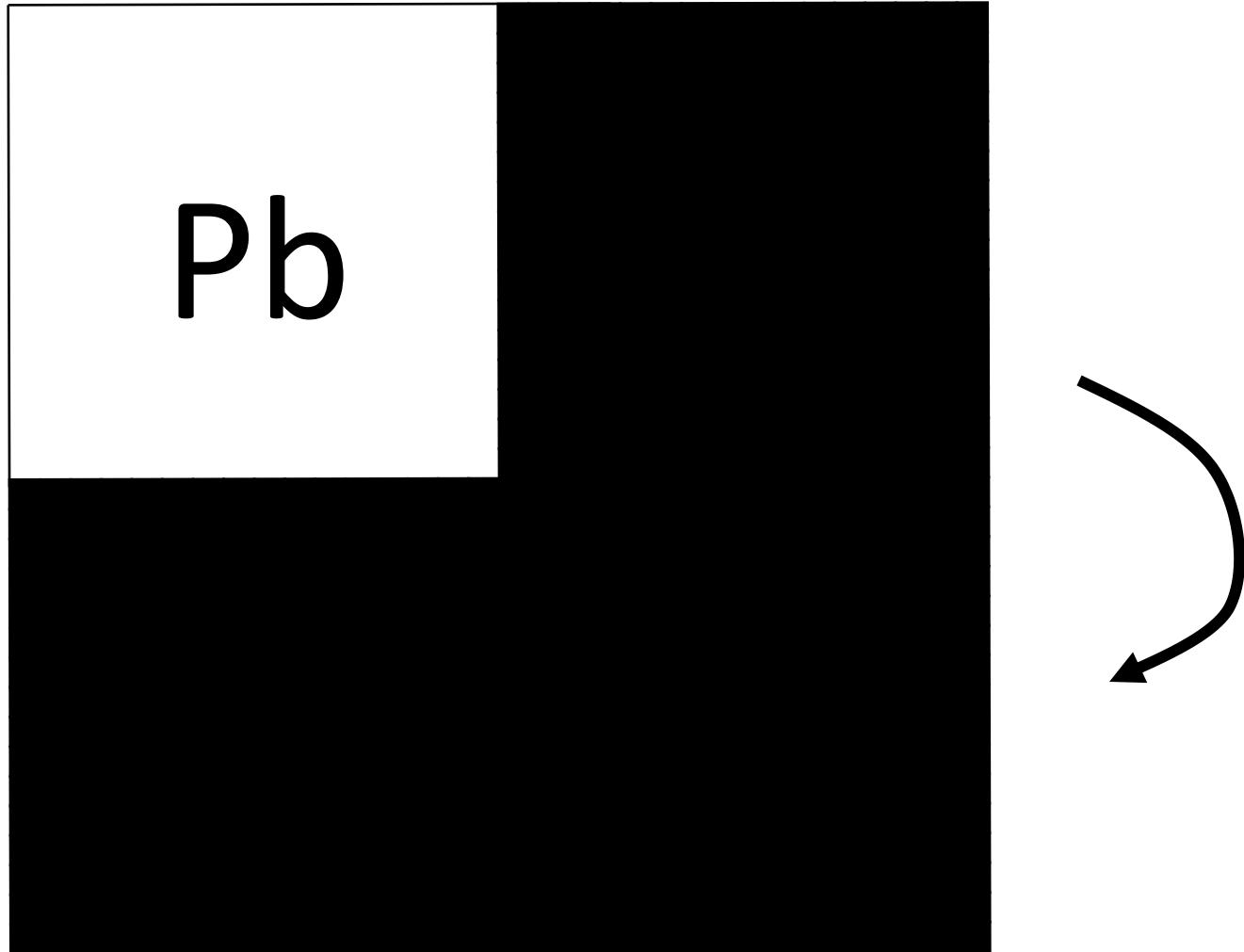
Quaternary Masking: 1st mask, 2nd position



Quaternary Masking: 1st mask, 3rd position



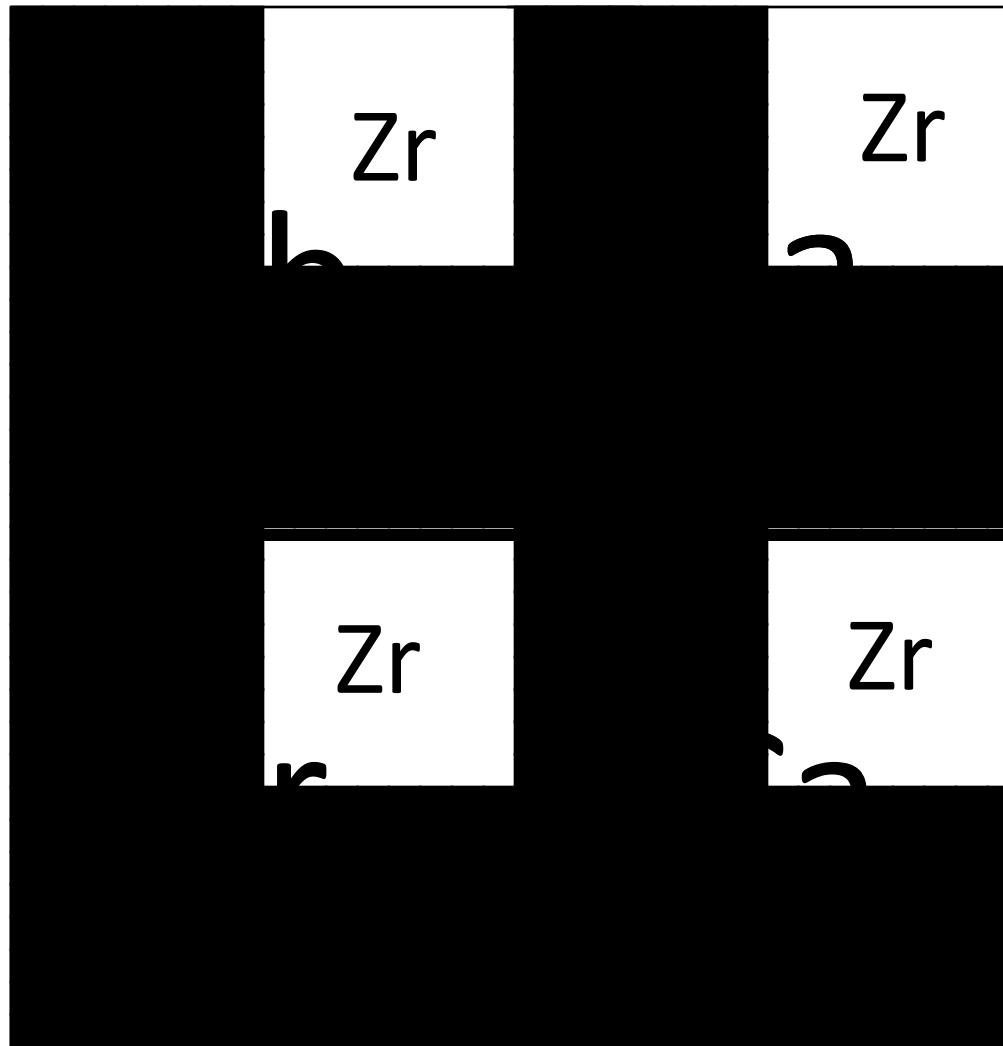
Quaternary Masking: 1st mask, 4th position



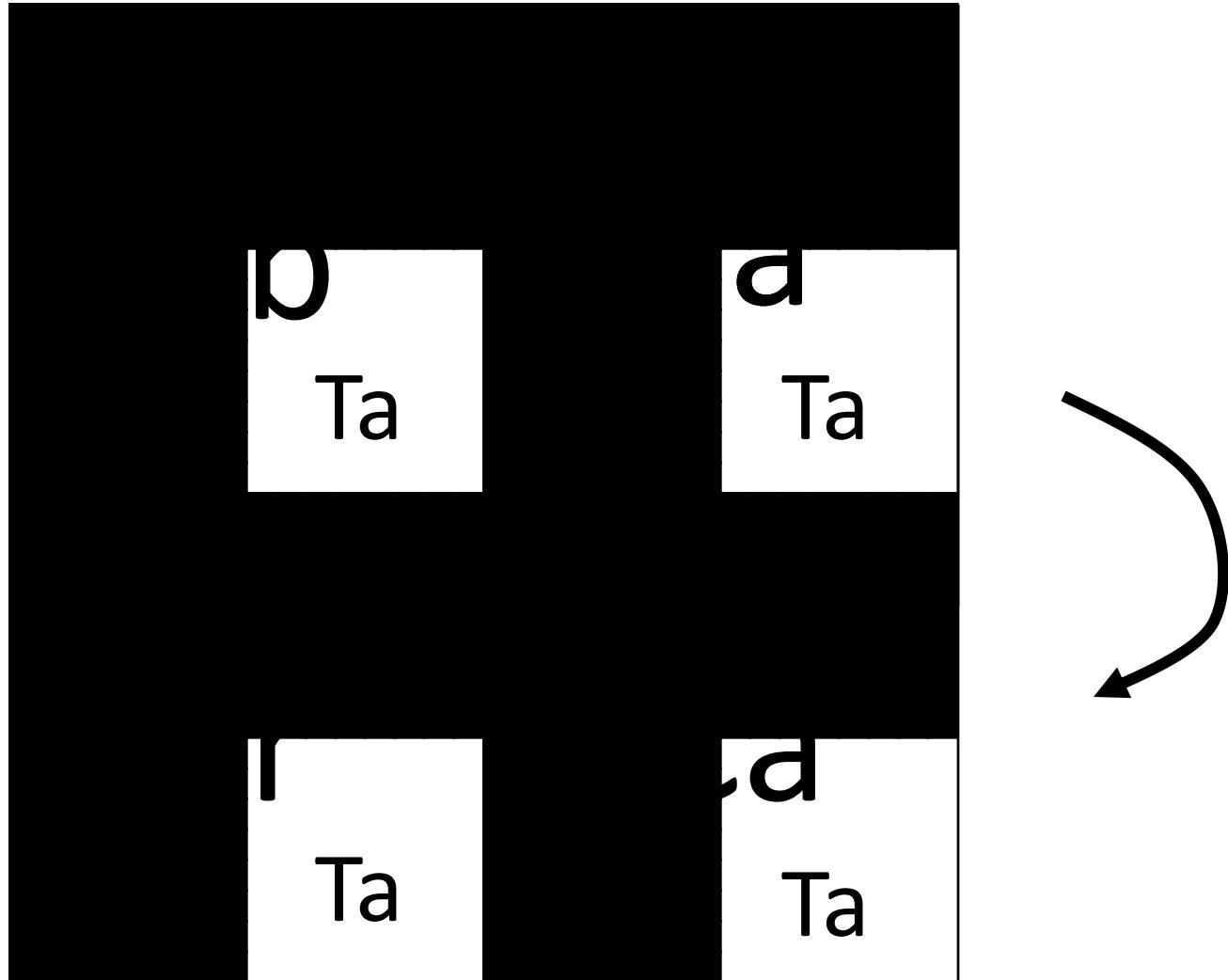
Quaternary Masking: after 1st mask

Pb	Ba
Sr	Ca

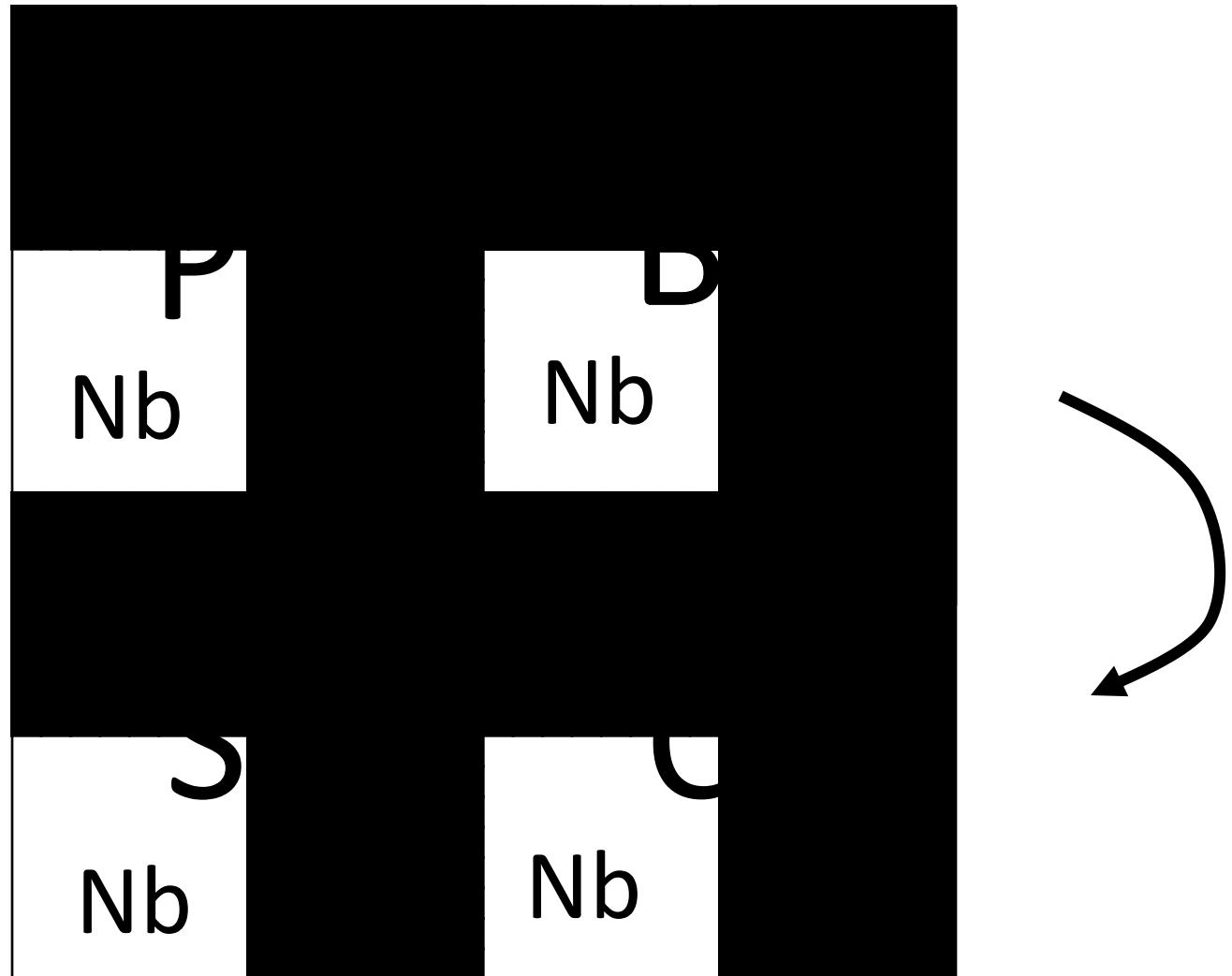
Quaternary Masking: 2nd mask, 1st position



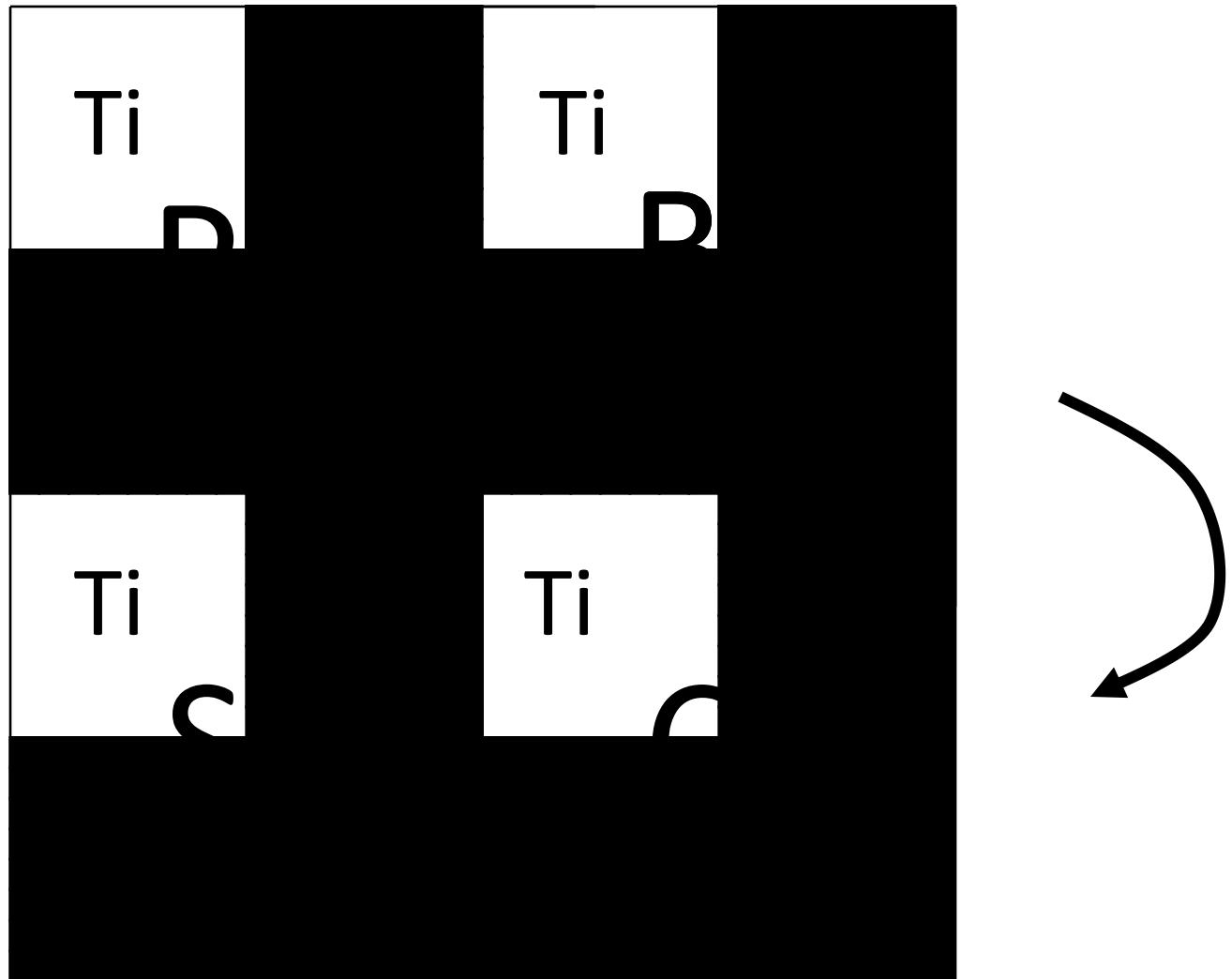
Quaternary Masking: 2nd mask, 2nd position



Quaternary Masking: 2nd mask, 3rd position



Quaternary Masking: 2nd mask, 4th position



PbTiO_3	PbZrO_3	BaTiO_3	BaZrO_3
PbNb_2O_6	PbTa_2O_6	BaNb_2O_6	BaTa_2O_6
SrTiO_3	SrZrO_3	CaTiO_3	CaZrO_3
SrNb_2O_6	SrTa_2O_6	CaNb_2O_6	CaTa_2O_6

depositions: $4 \times n$

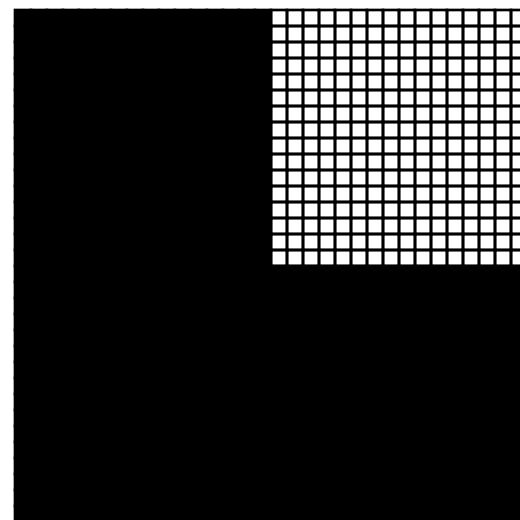
A

combinations: 4^n

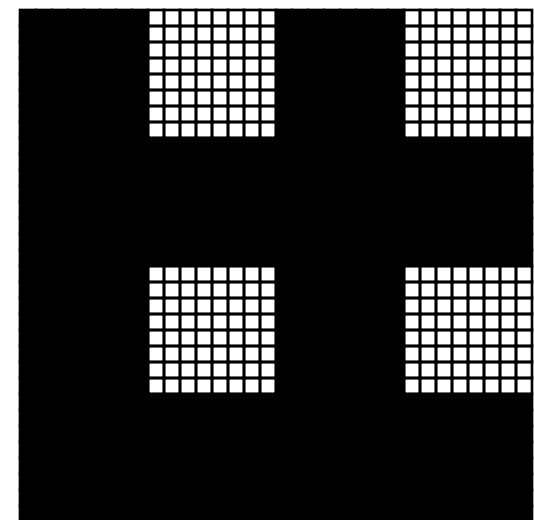
5 masks:

$4 \times 5 = 20$ depo's

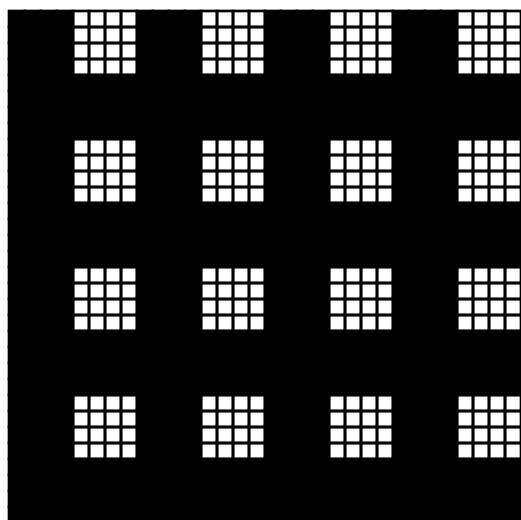
$4^5 = 1024$ samples



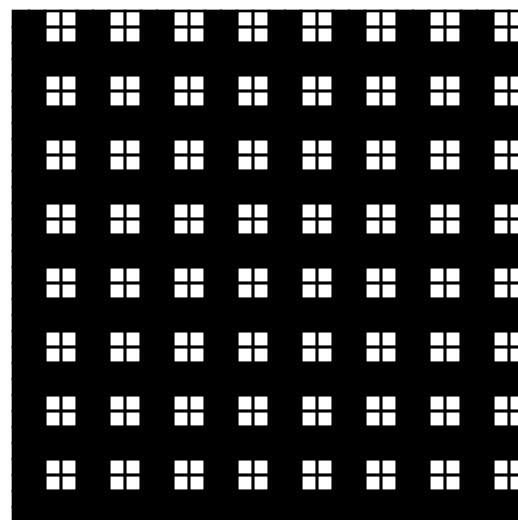
B



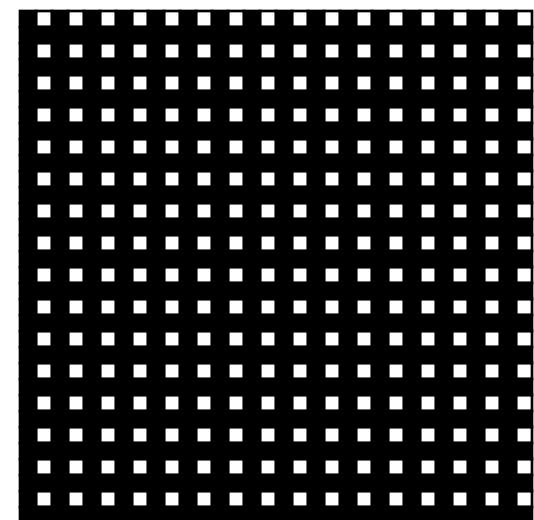
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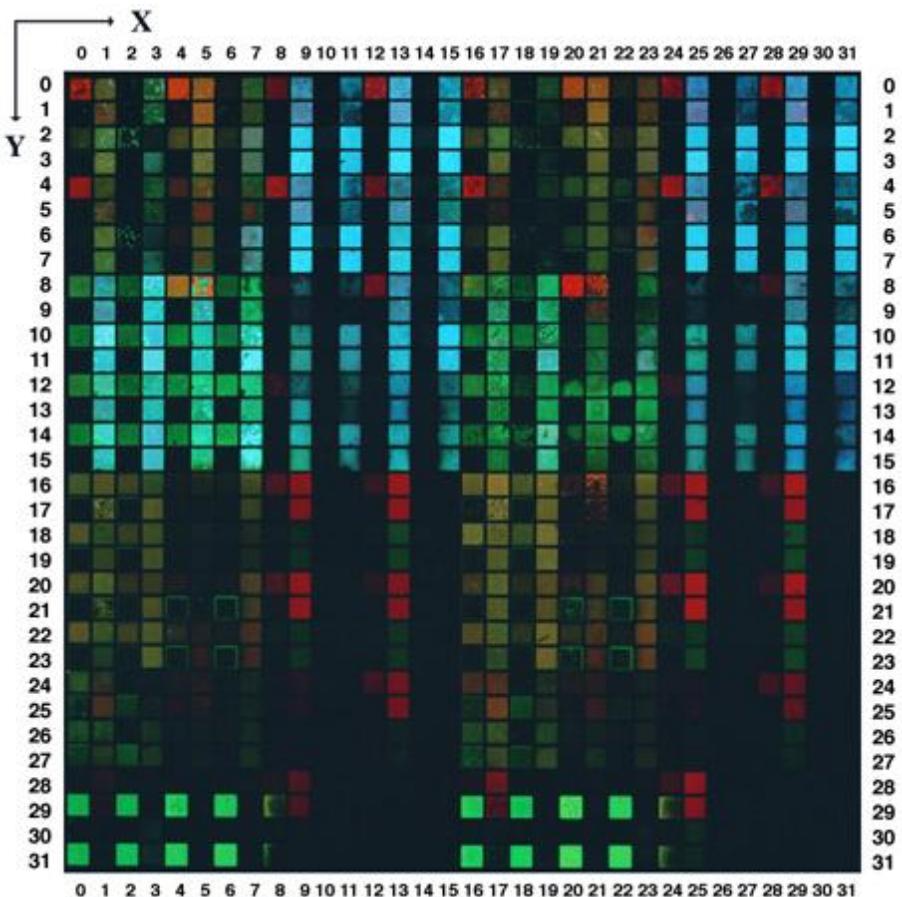
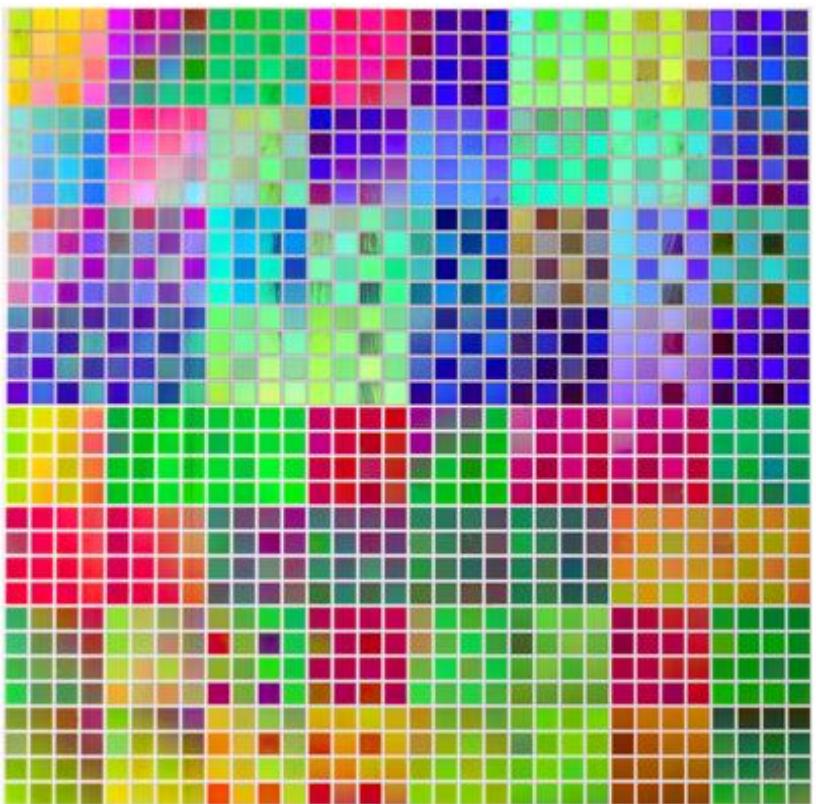


D



E

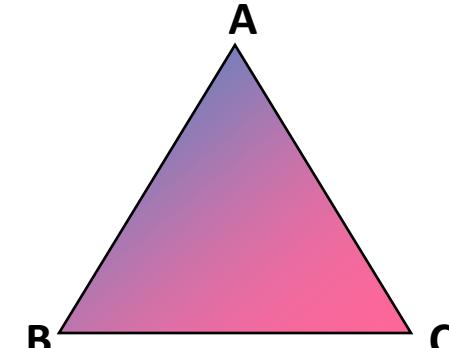
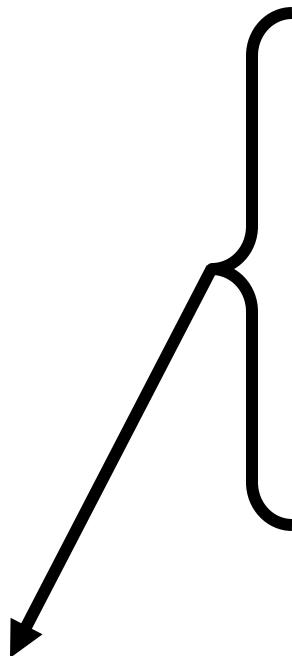
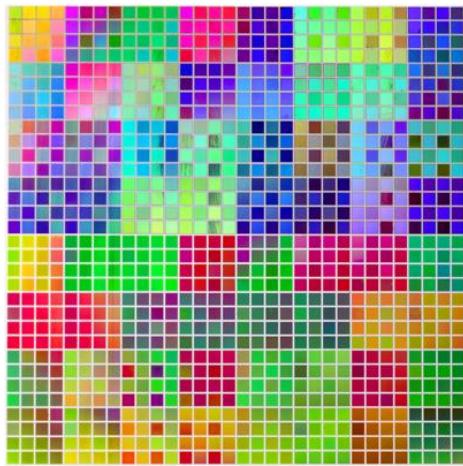




(Right) Luminescent image of the same library after thermally processed under UV excitation.

Various combinatorial experimental designs

discrete libraries vs composition spreads

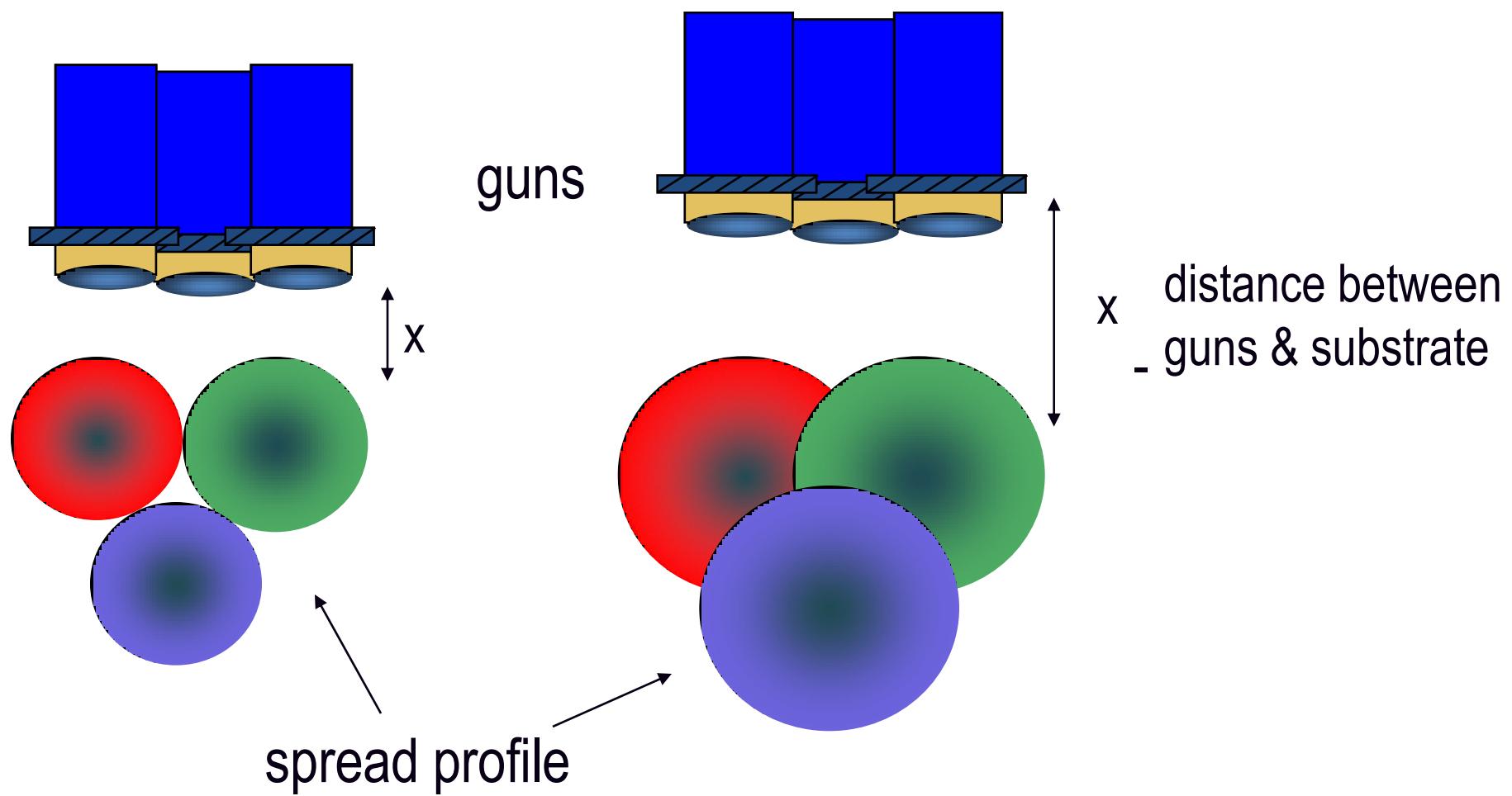


- Composition spreads allow continuous mapping of physical properties and phase boundaries
- Run to run variation in ordinary experiments is removed

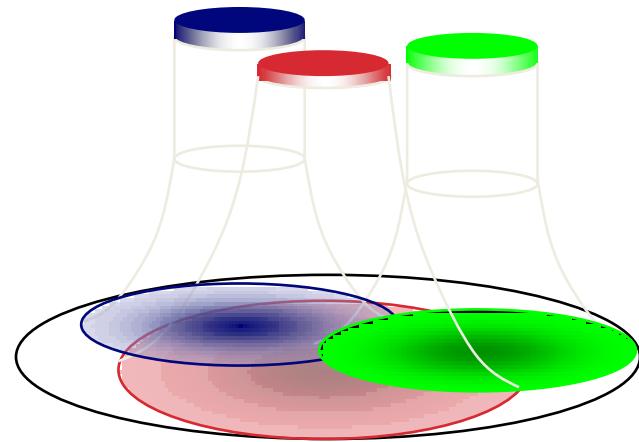
Combinatorial UHV

Co-sputtering ($P_{\text{base}} \sim 1 \times 10^{-9}$ Torr)

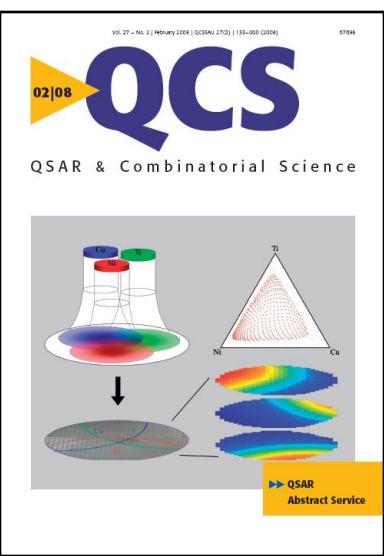
CCS-Continuos Composition Spread Approach



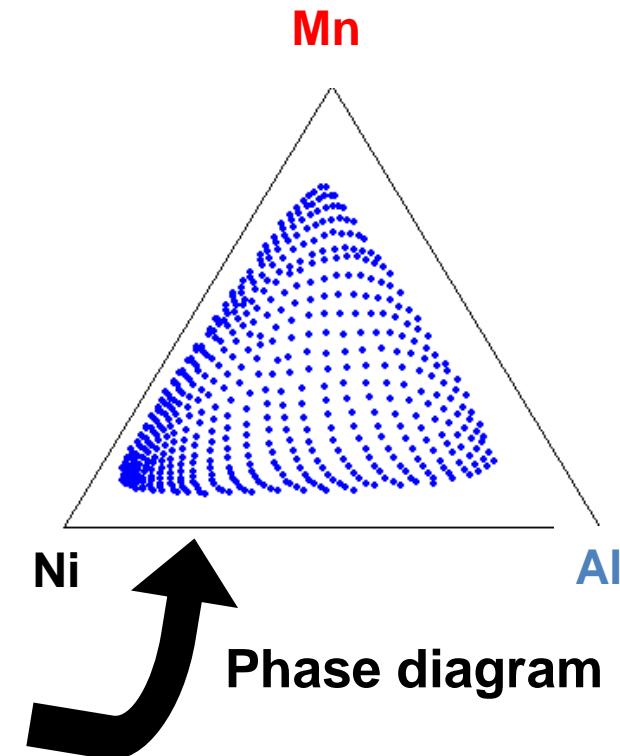
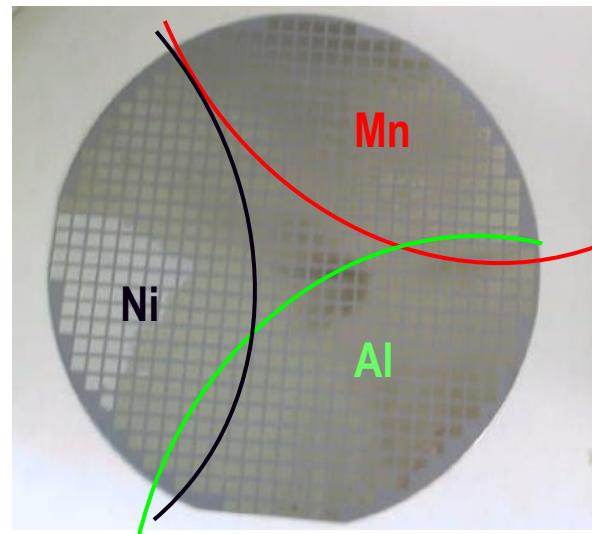
Composition Spreads of Ternary Metallic Alloy Systems



Co-sputtering scheme



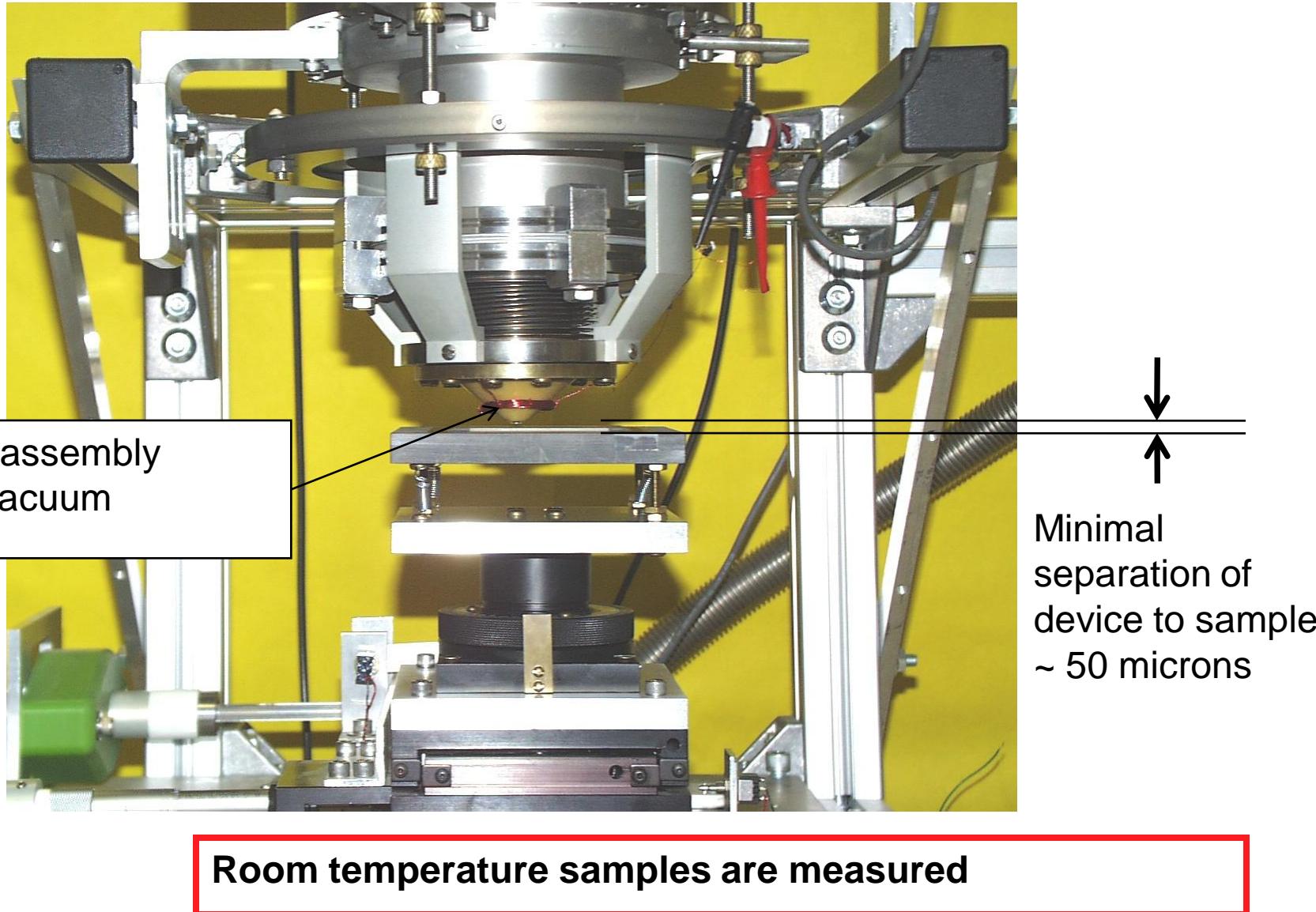
3" spread wafer



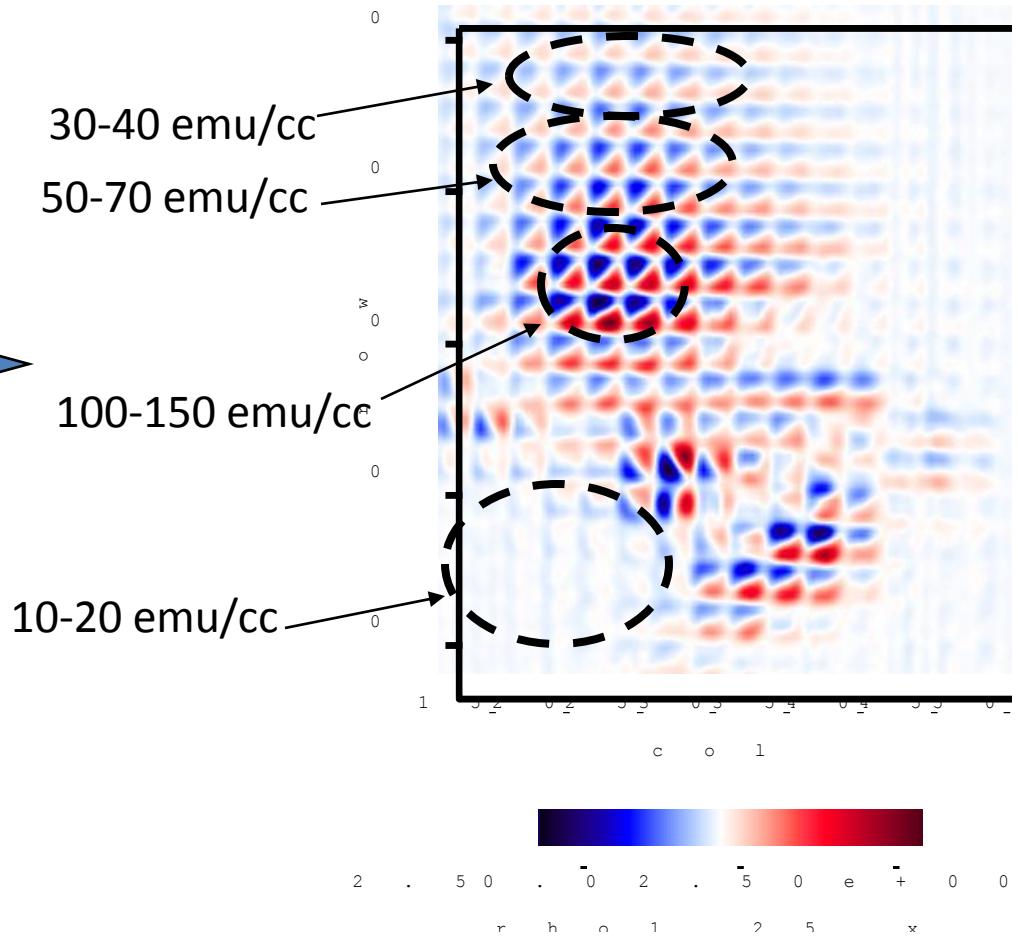
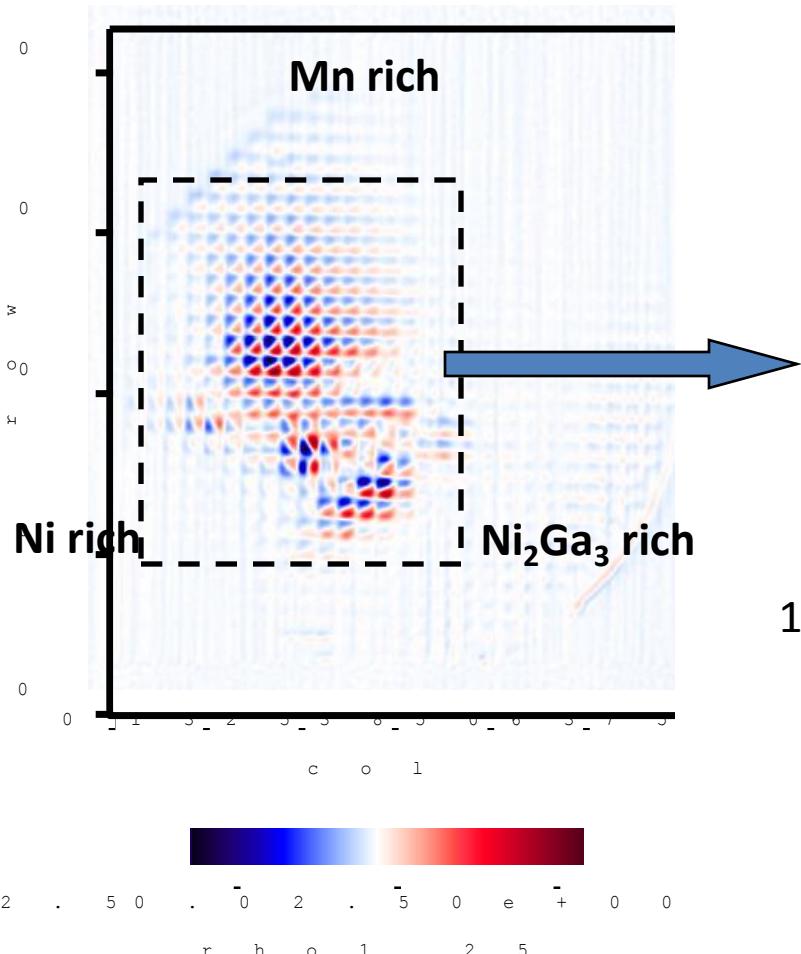
Phase diagram

Composition is mapped using an electron probe (WDS)

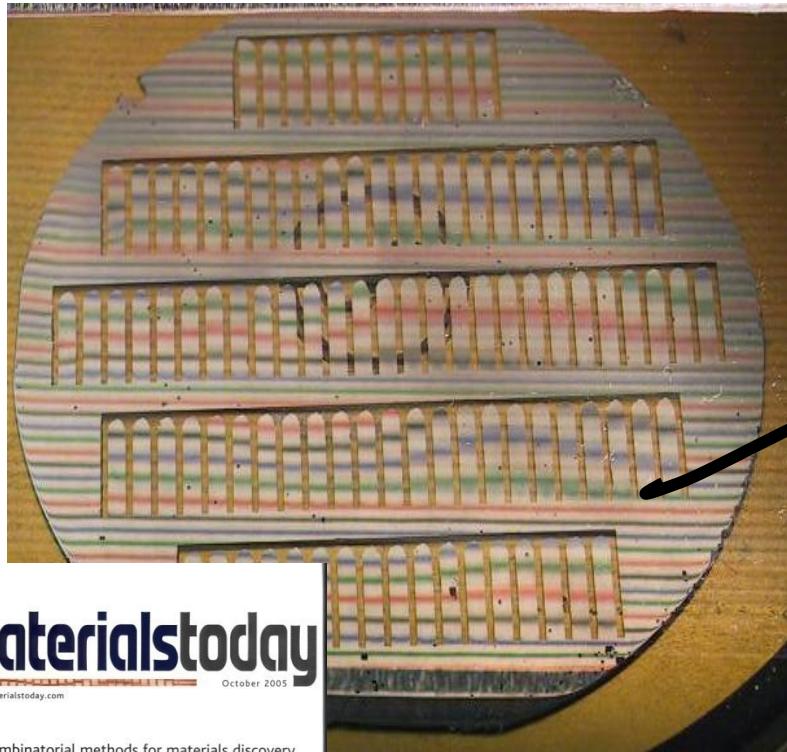
Scanning SQUID microscope based on $\text{YBa}_2\text{Cu}_3\text{O}_7$ thin film (F. Wellstood, UMD)



Scanning SQUID image of a Ni-Mn-Ga spread wafer (room temperature)

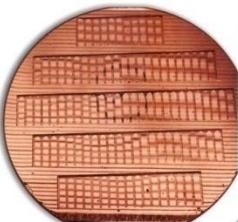


Combinatorial search of ferromagnetic shape memory alloys

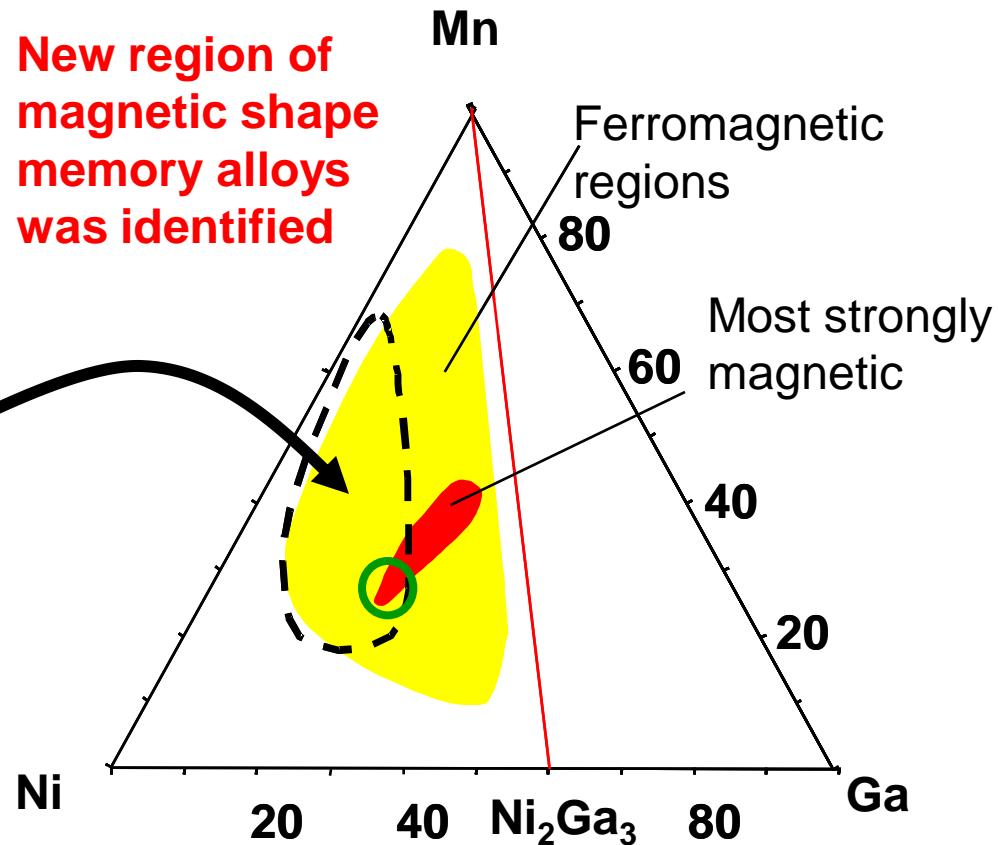


materialstoday
October 2005

Combinatorial methods for materials discovery
High-throughput mapping of properties
The arrival of materials informatics

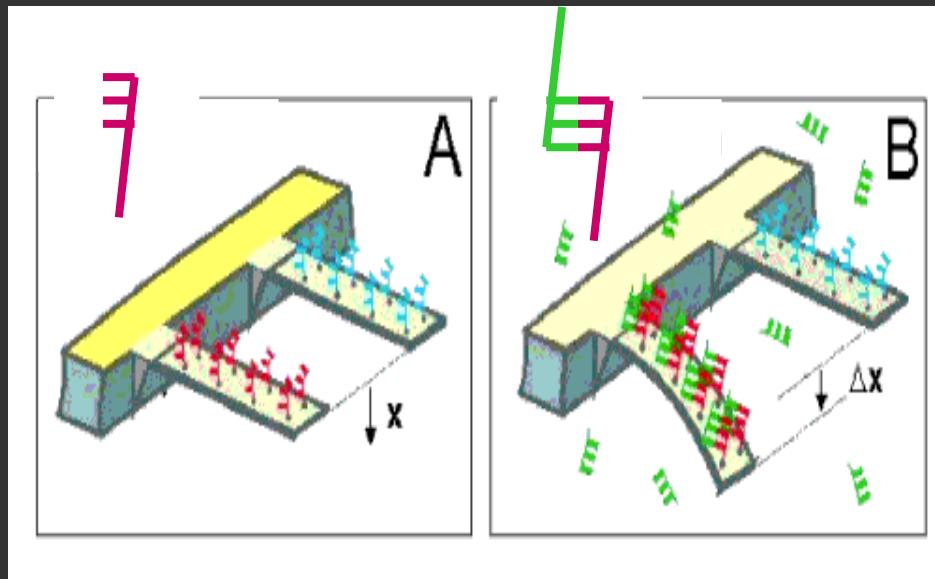


Composition gradient fabricated on micromachined cantilever arrays



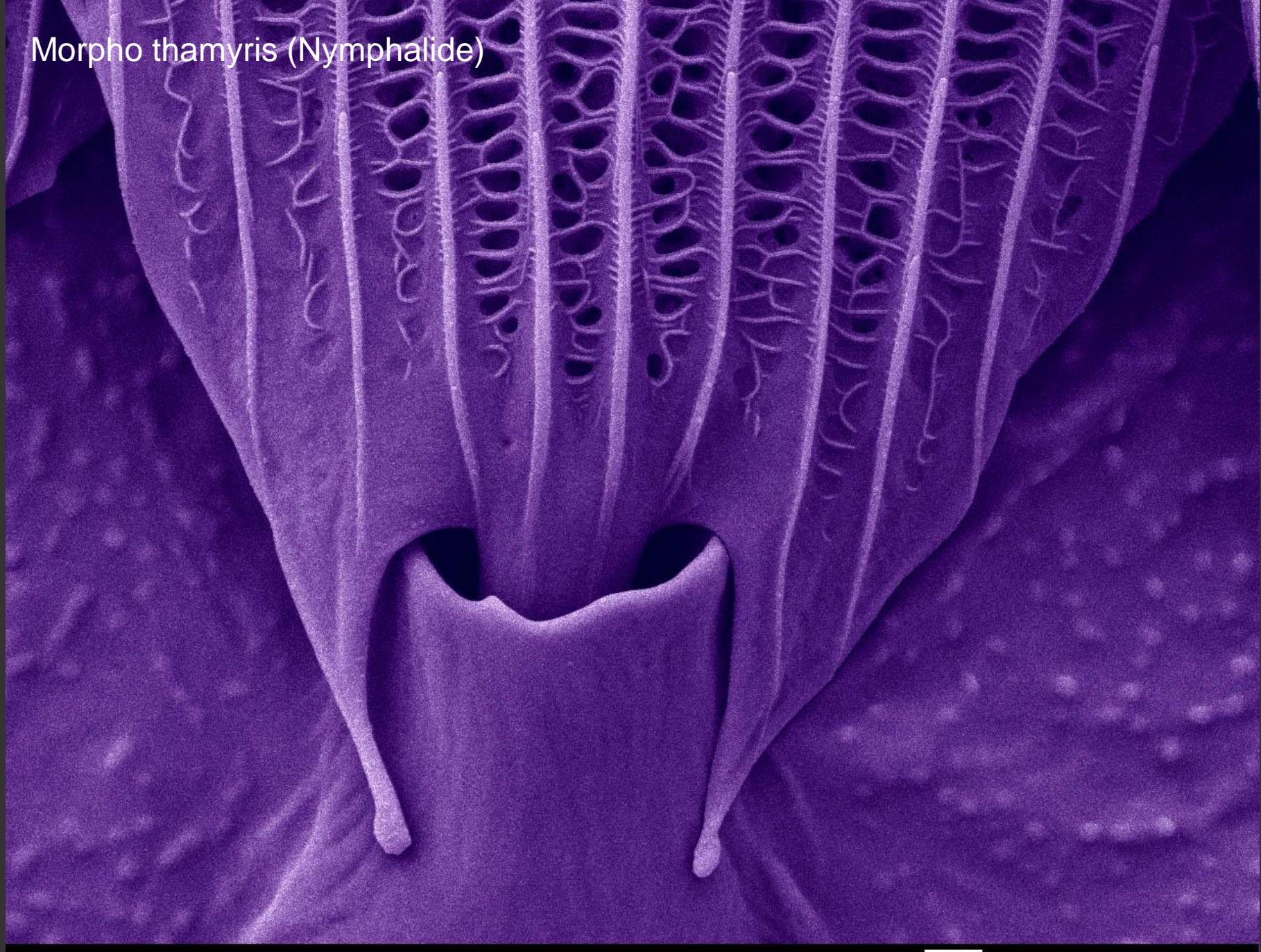
Nature Materials 2, 180 (2003)

“Sistema di monitoraggio rapido del gruppo sanguigno e per la rivelazione di reazioni immunoematologiche”



macro → micro → nano

Morpho thamyris (Nymphalide)



NONE

LEI

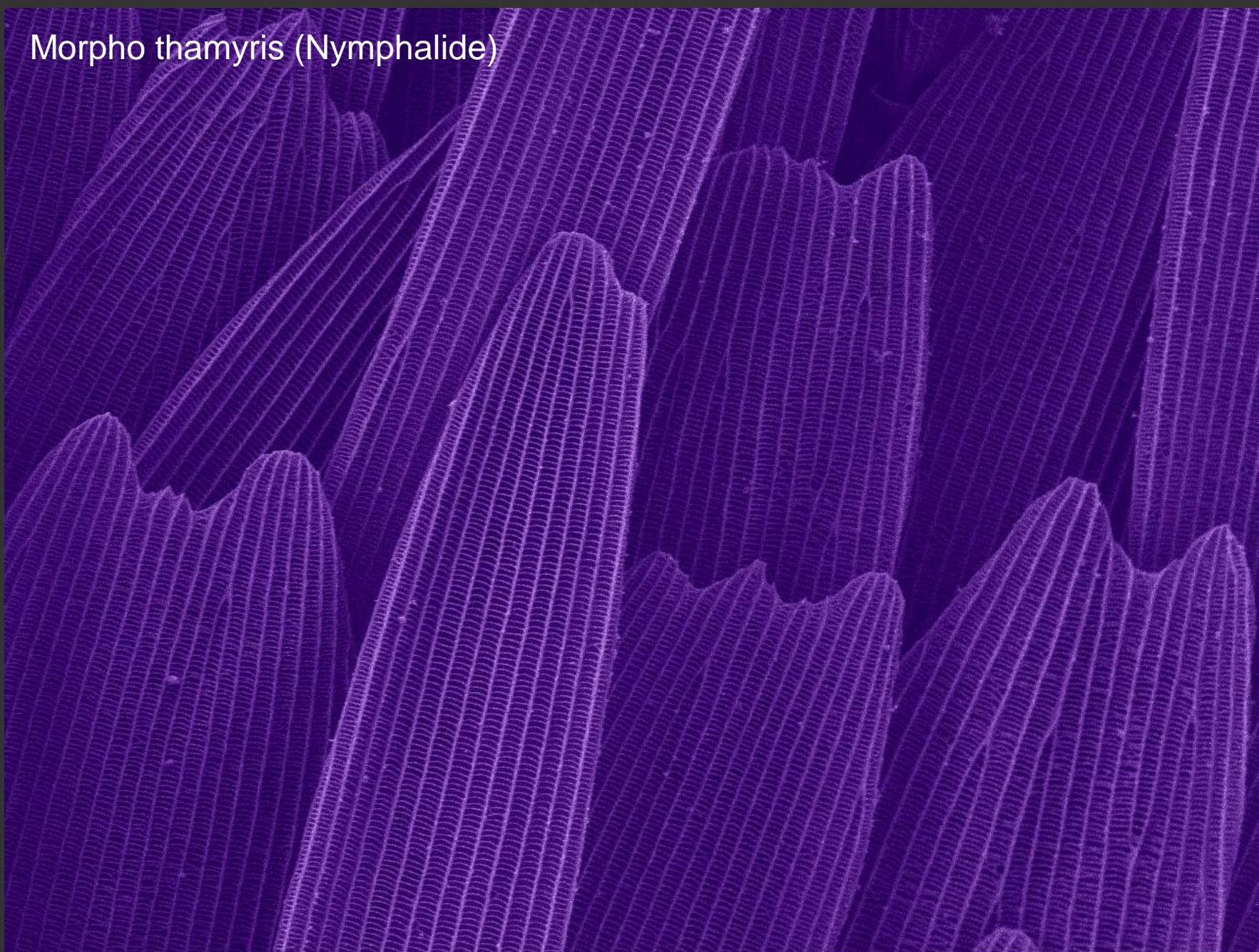
5.0kV

X5,500

1μm

WD 7.4mm

Morpho thamyris (Nymphalide)



NONE

LEI

5.0kV

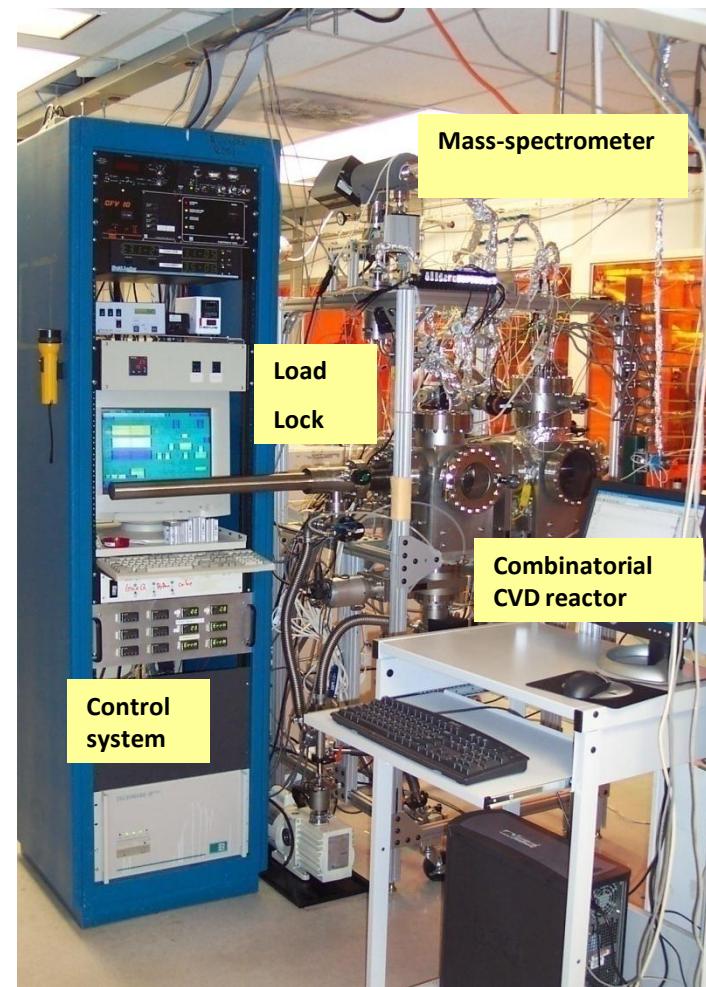
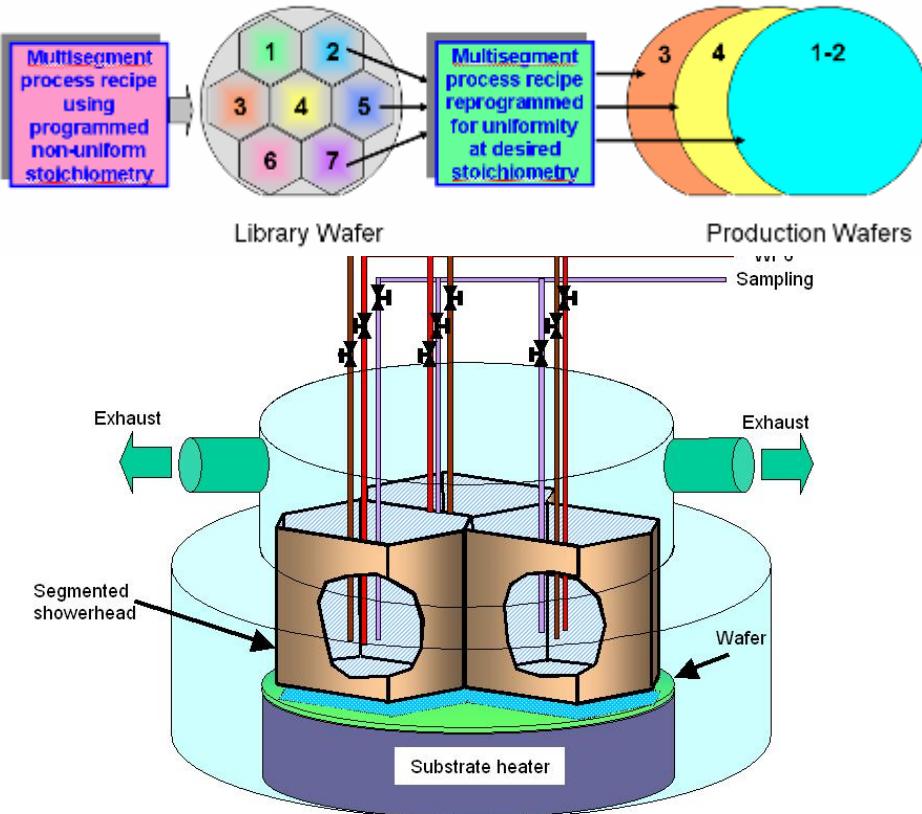
X1,000

10 μ m

WD 7.2mm

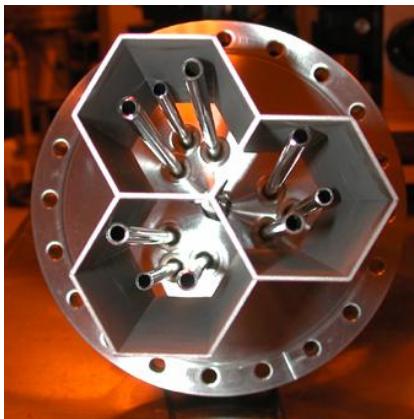
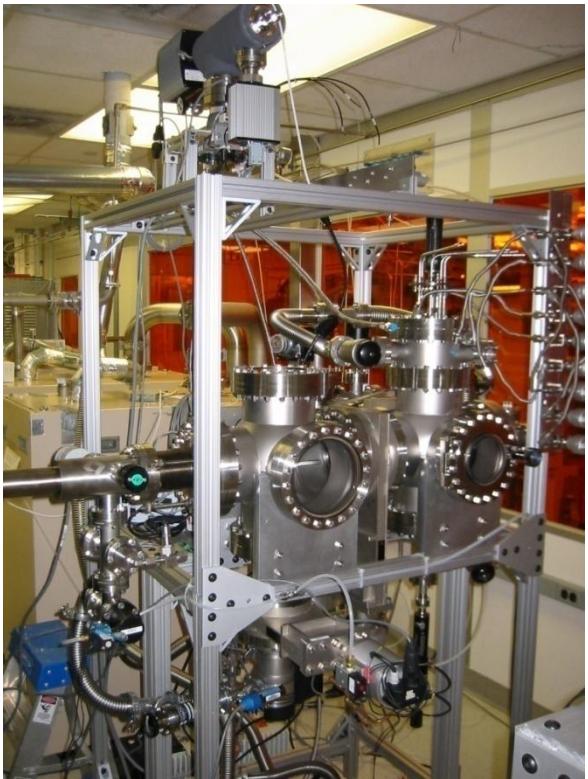
Spatially Programmable Equipment Design for Combinatorial CVD

New paradigm for intelligent design and control of semiconductor process equipment



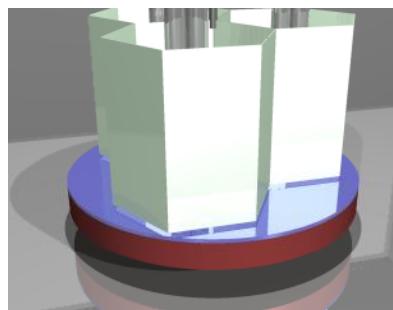
G. W. Rubloff, R. Adomaitis, et. al.

Combinatorial CVD Reactor

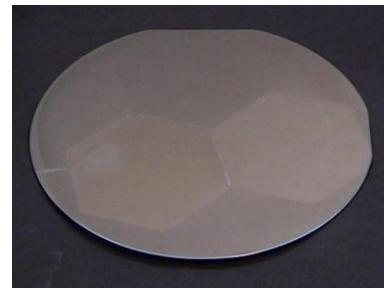
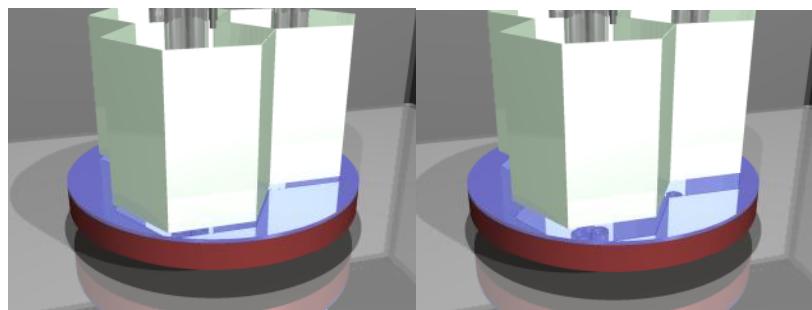


Segment-wafer spacing

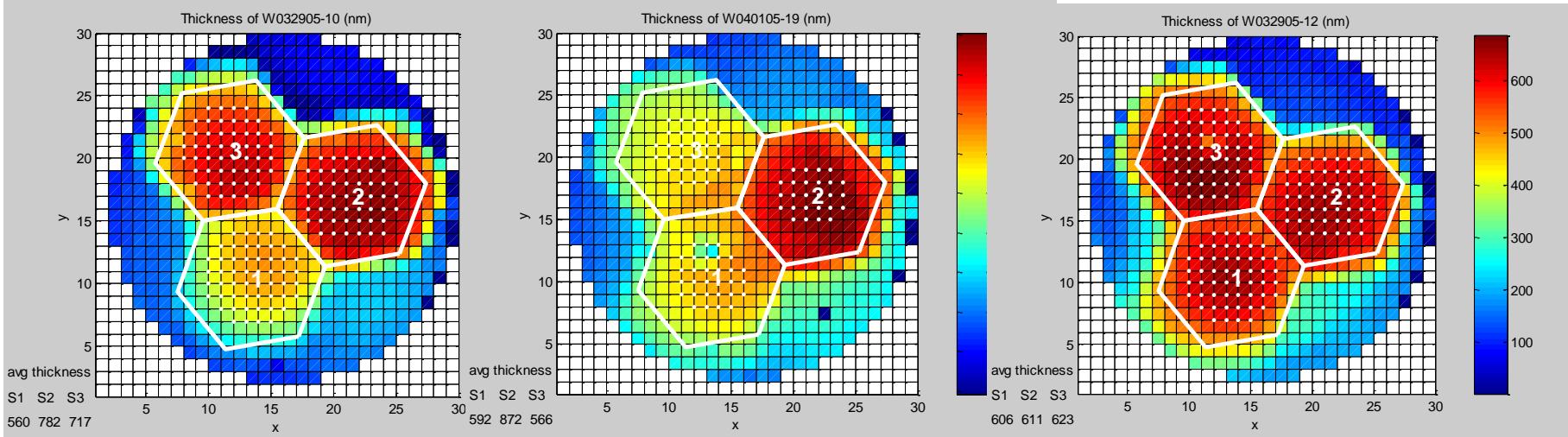
1mm



3mm



Combi-CVD Programmability



W032905_10	S1	S2	S3
WF6	6	9	12
H2	24	36	48
Ar	30	15	0

W040105_19	S1	S2	S3
WF6	9	12	6
H2	36	48	24
Ar	15	0	30

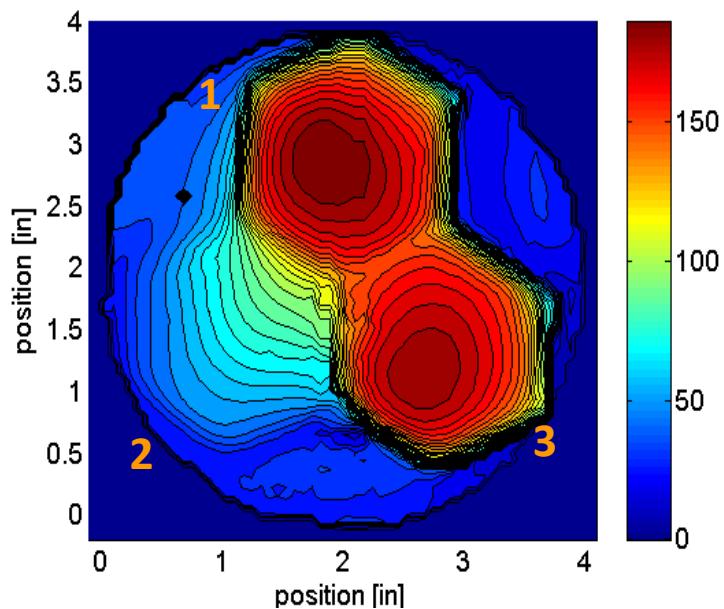
W032905_12	S1	S2	S3
WF6	12	6	9
H2	48	24	36
Ar	0	30	15

- Spatially programmable CVD enables combinatorial studies
- Post-process mapping of thickness/rate
- Composition mapping (IRST microcombi project)

Combinatorial W CVD from WF₆/H₂

Reactant flow distributions

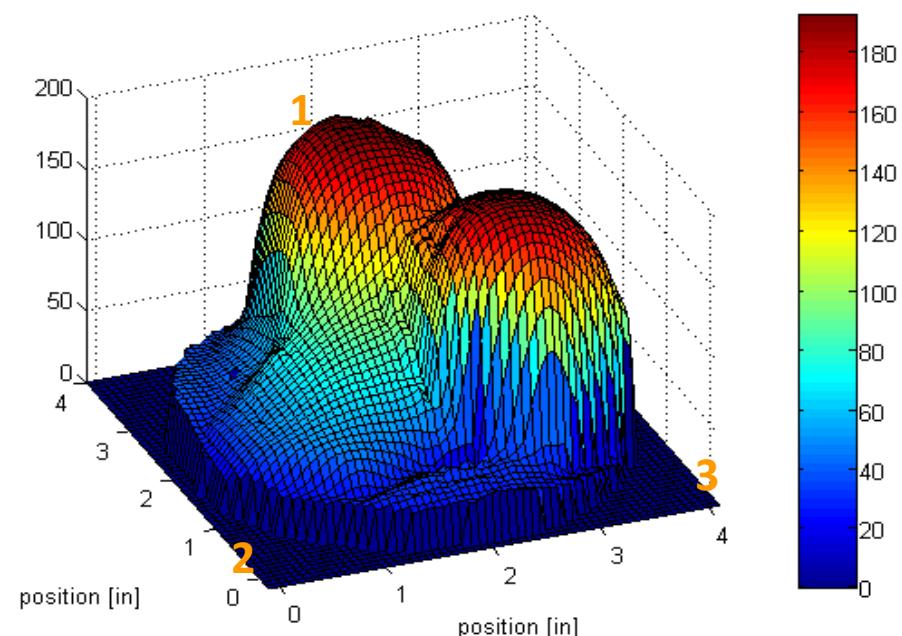
	Seg 1.	Seg 2.	Seg 3.	
Ar	0	60	30	[sccm]
WF6	12	0	6	[sccm]
H2	48	0	24	[sccm]
Total	60	60	60	[sccm]



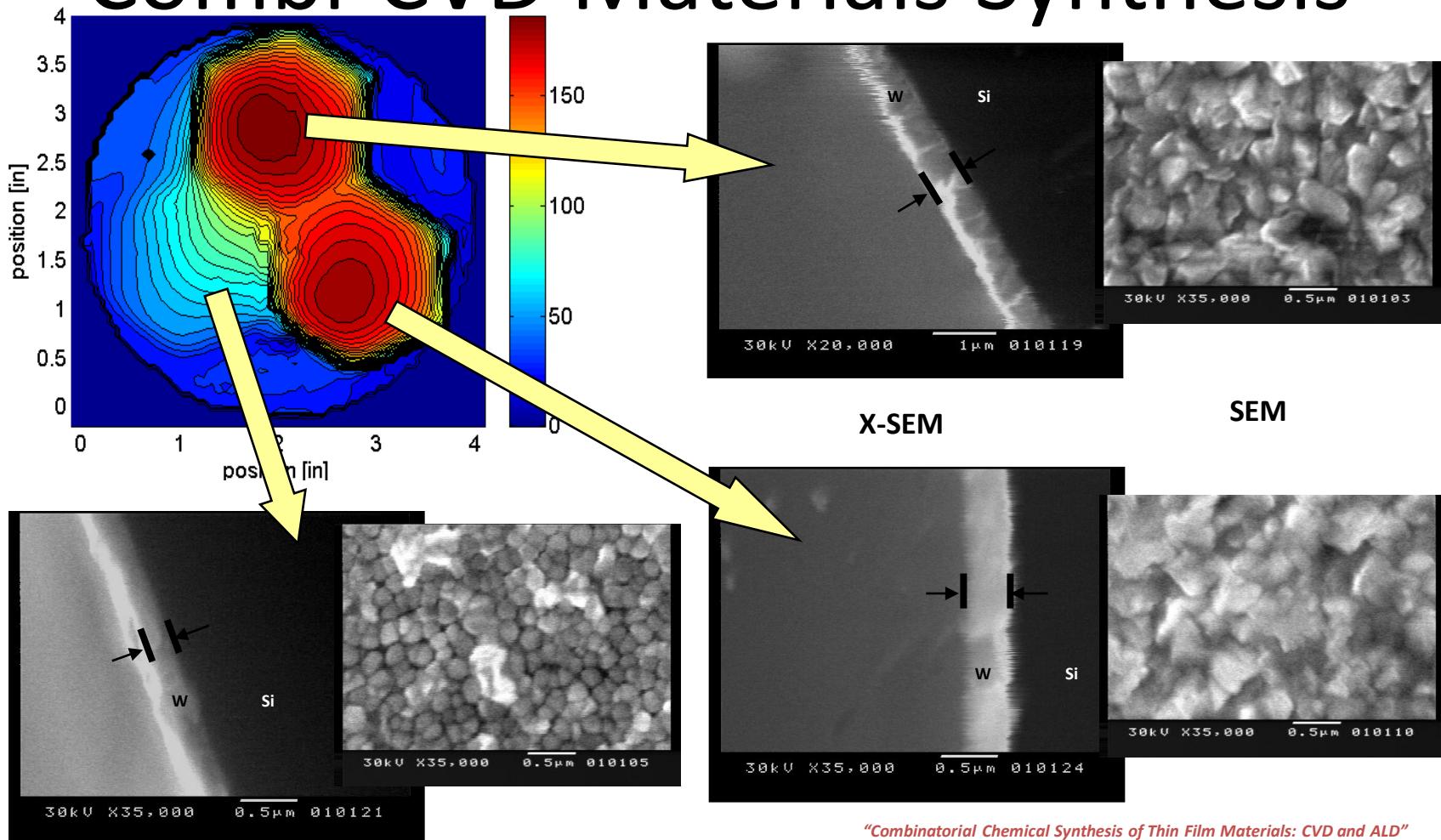
Film thickness distributions

$$\text{Growth rate} \propto [P_{H_2}]^{1/2}$$

Heater T : 400 °C
Chamber P : 1 torr
Gap : 1mm
Process time : 10 min



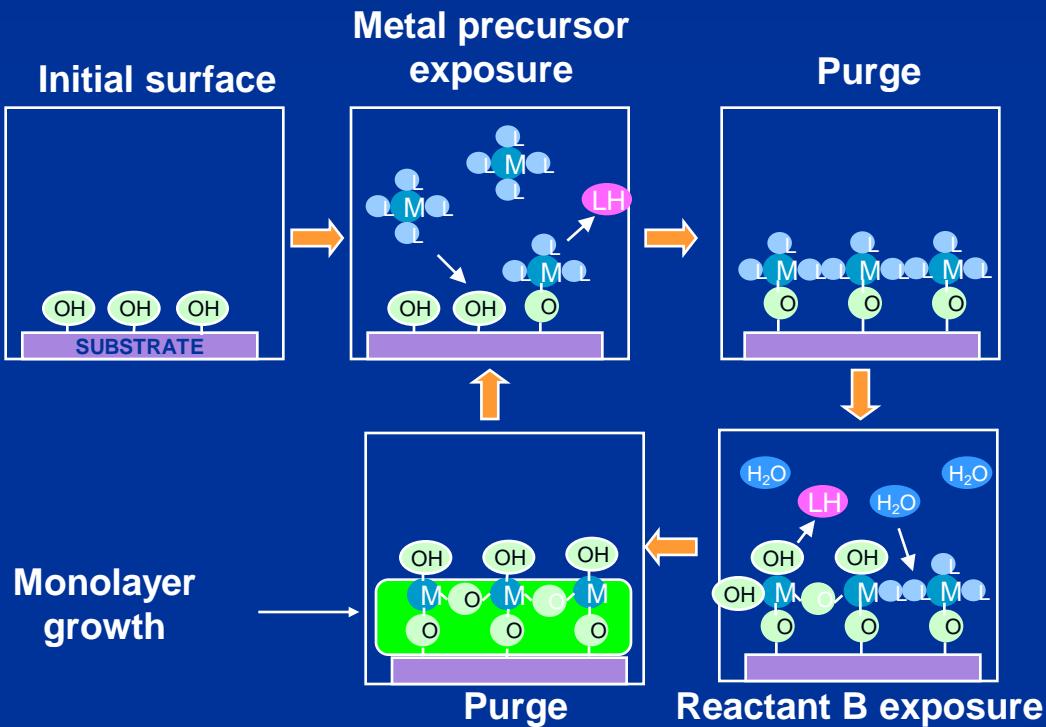
Combi-CVD Materials Synthesis



"Combinatorial Chemical Synthesis of Thin Film Materials: CVD and ALD"
G.W.Rubloff, R.A.Adomaitis, L.Henn-Lecordier and M.Anderle
Invited to 4th International Workshop on Combinatorial Materials
Science and Technology, December 4-6, 2006, San Juan, Puerto Rico

Atomic Layer Deposition

*a simple picture
an ideal process*



BUT...

Nucleation & surface condition dependence

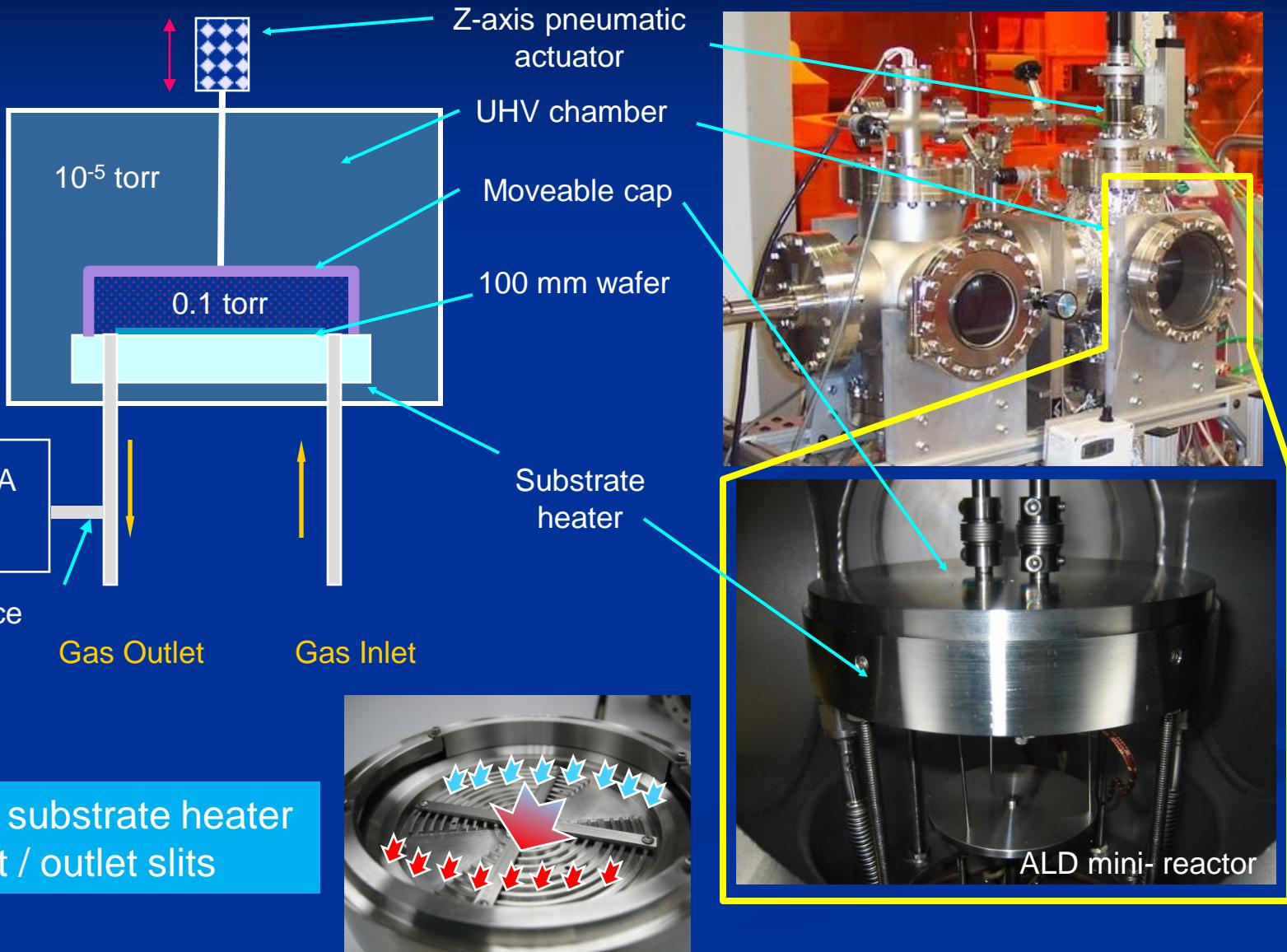
Temperature-dependent growth

Dose dependencies

Incomplete layer adsorption & reaction

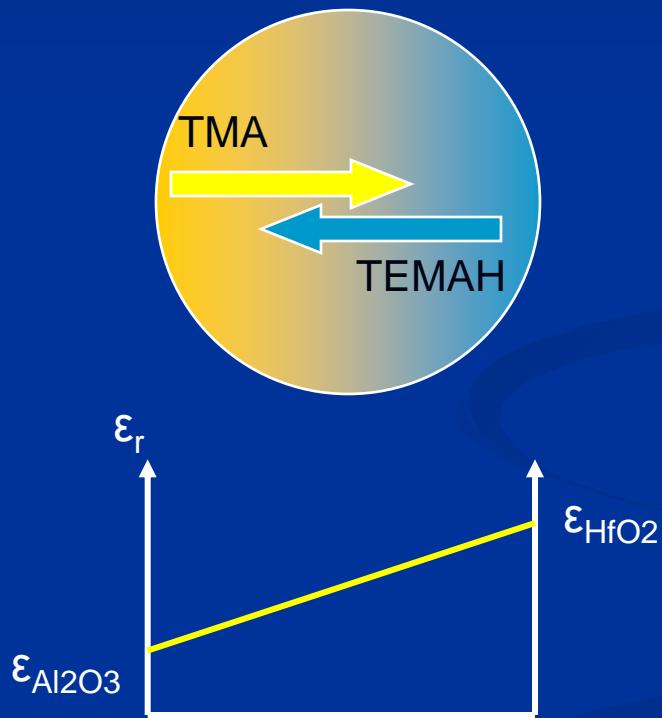
Multilayer adsorption & reaction

ALD reactor design



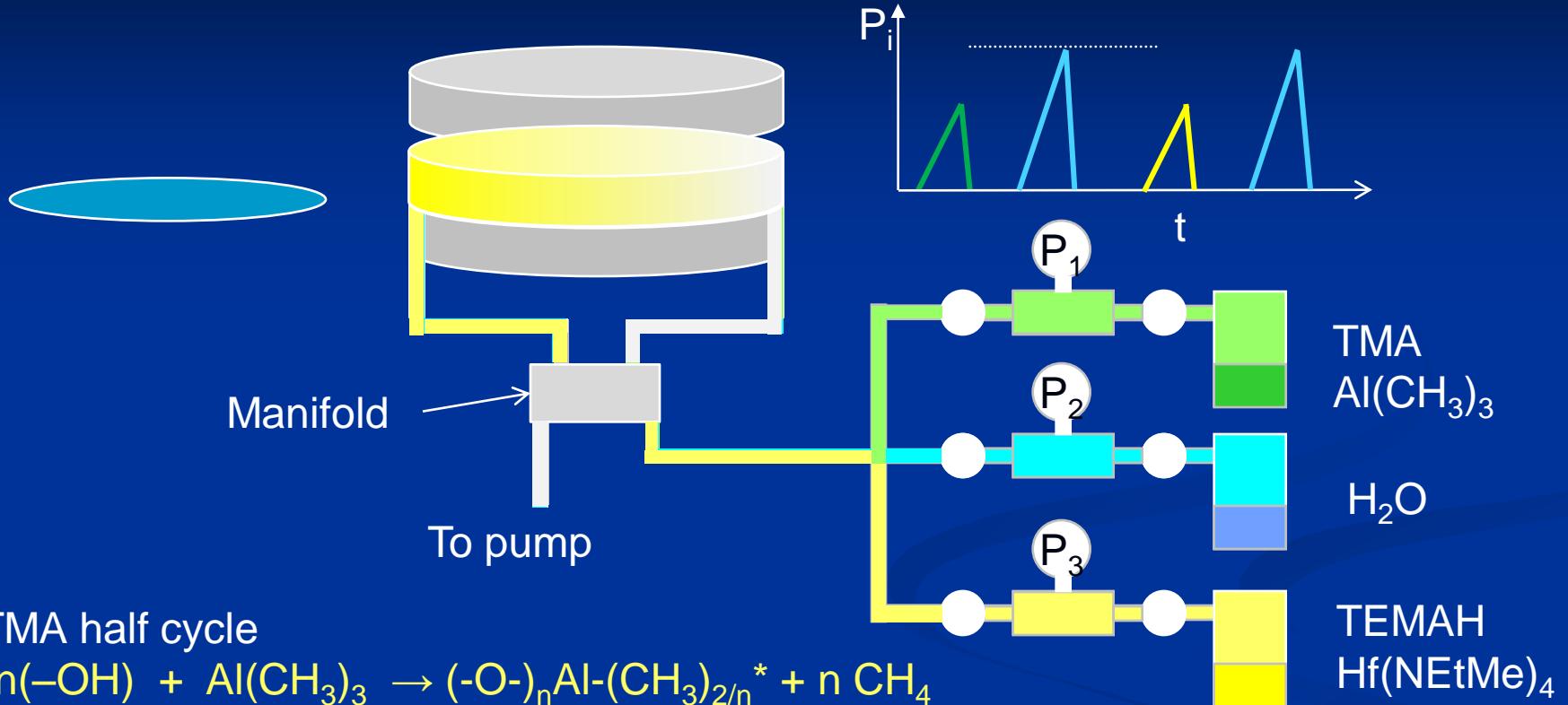
Our approach to combi ALD

Final objective:
Compositional gradient for ternary systems requires thickness gradient across wafer



Tuning of materials properties via
compositional gradient

Process flow sequence for combi Hf-Al-O films



TMA half cycle



H_2O half cycle



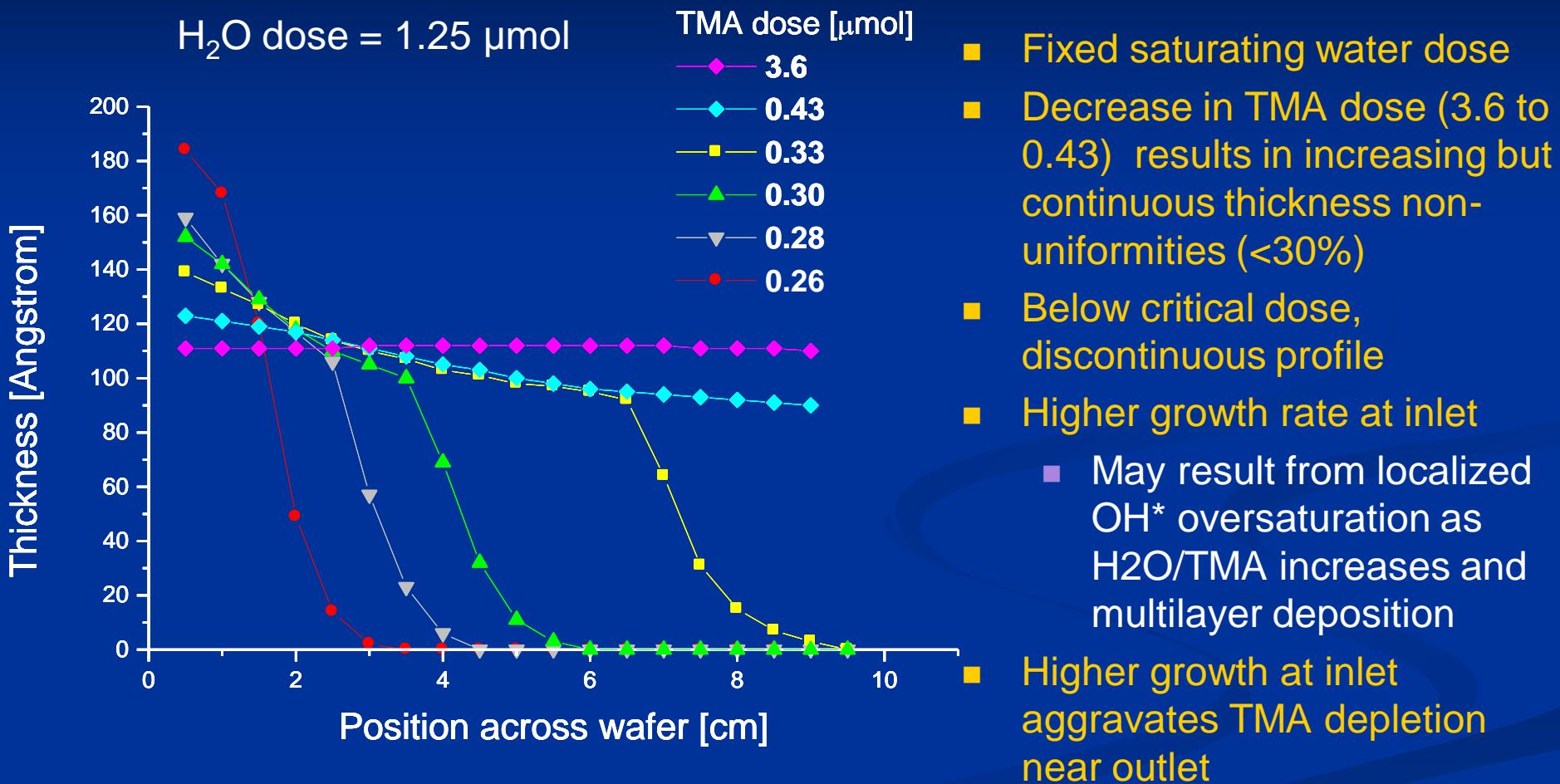
TEMAH half cycle



H_2O half cycle



Effect of TMA dose on thickness profiles



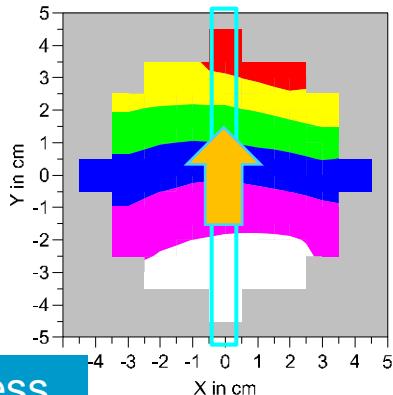
ALD cross-flow arrangements makes it easy to see when reaction is unbalanced as it amplifies consequent non uniformities

Ex-situ wafer characterization under water-starved conditions (0.8 μ mol)

Spect. ellipsometry

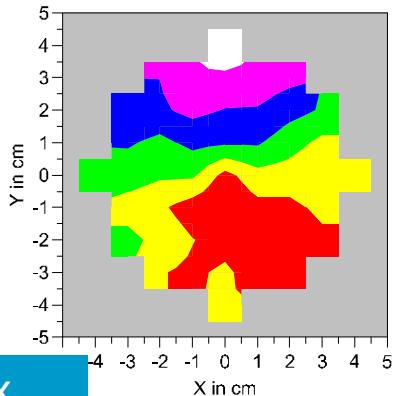
Thickness

Mean = 142.77
Min = 88.007
Max = 183.32
Std Dev = 24.103
Uniformity = 16.883 %

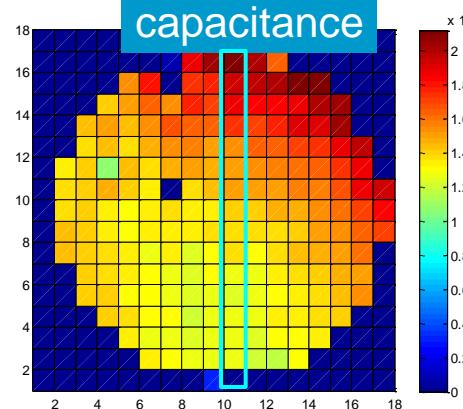


Index

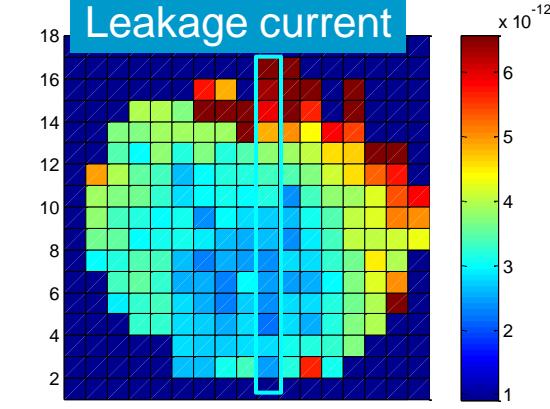
Mean = 1.6756
Min = 1.6398
Max = 1.7378
Std Dev = 0.024258
Uniformity = 1.4477 %



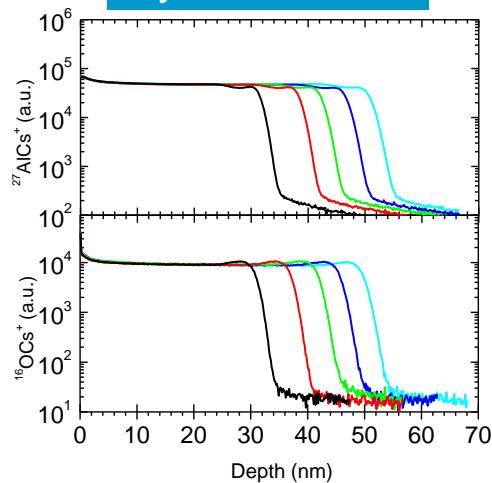
capacitance



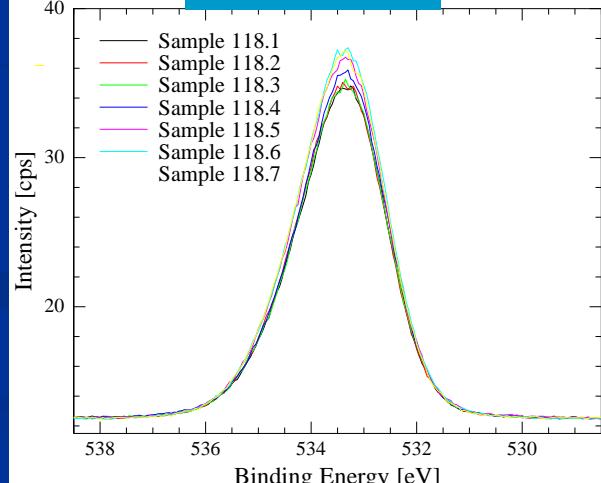
Leakage current



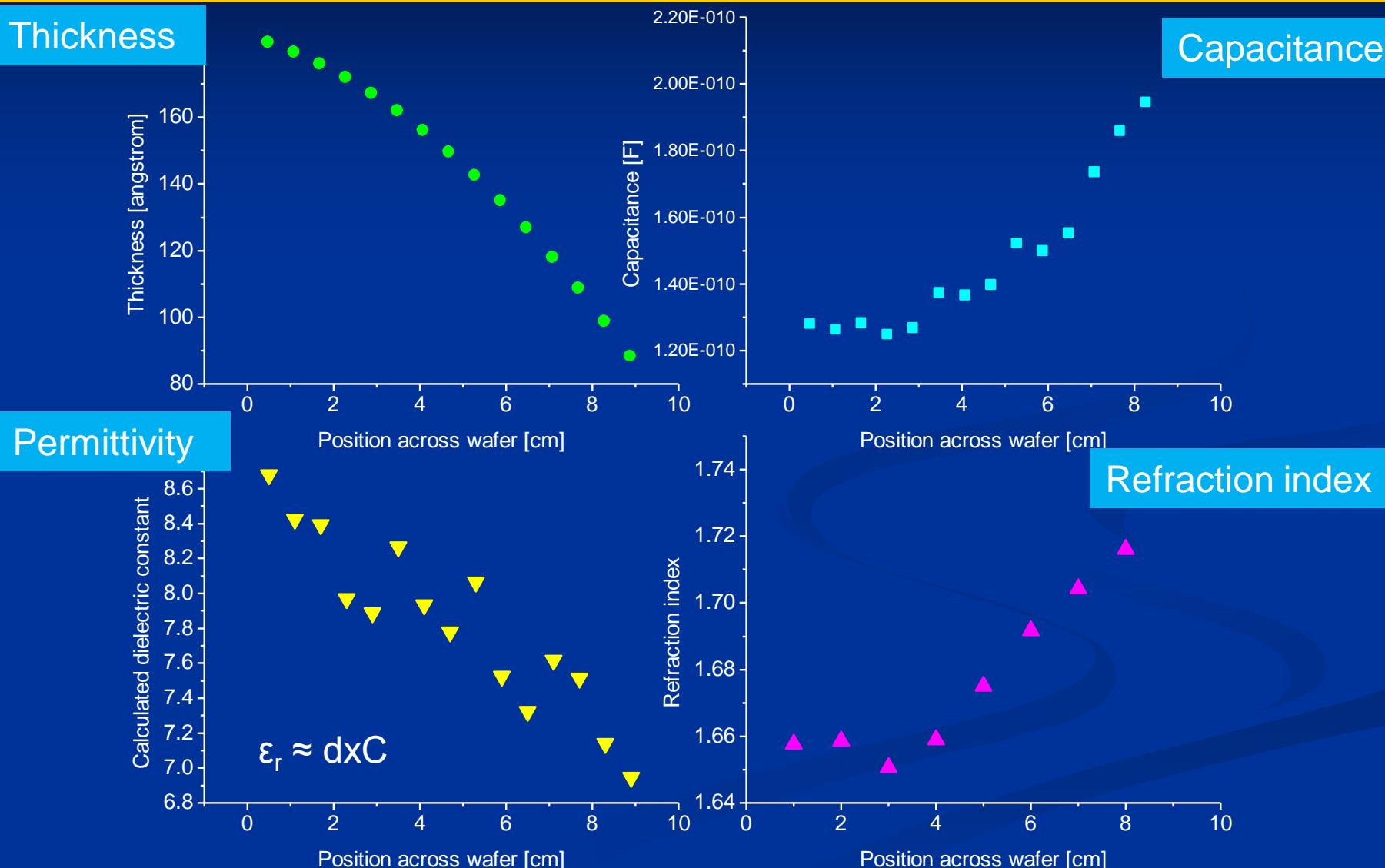
Dynamic SIMS



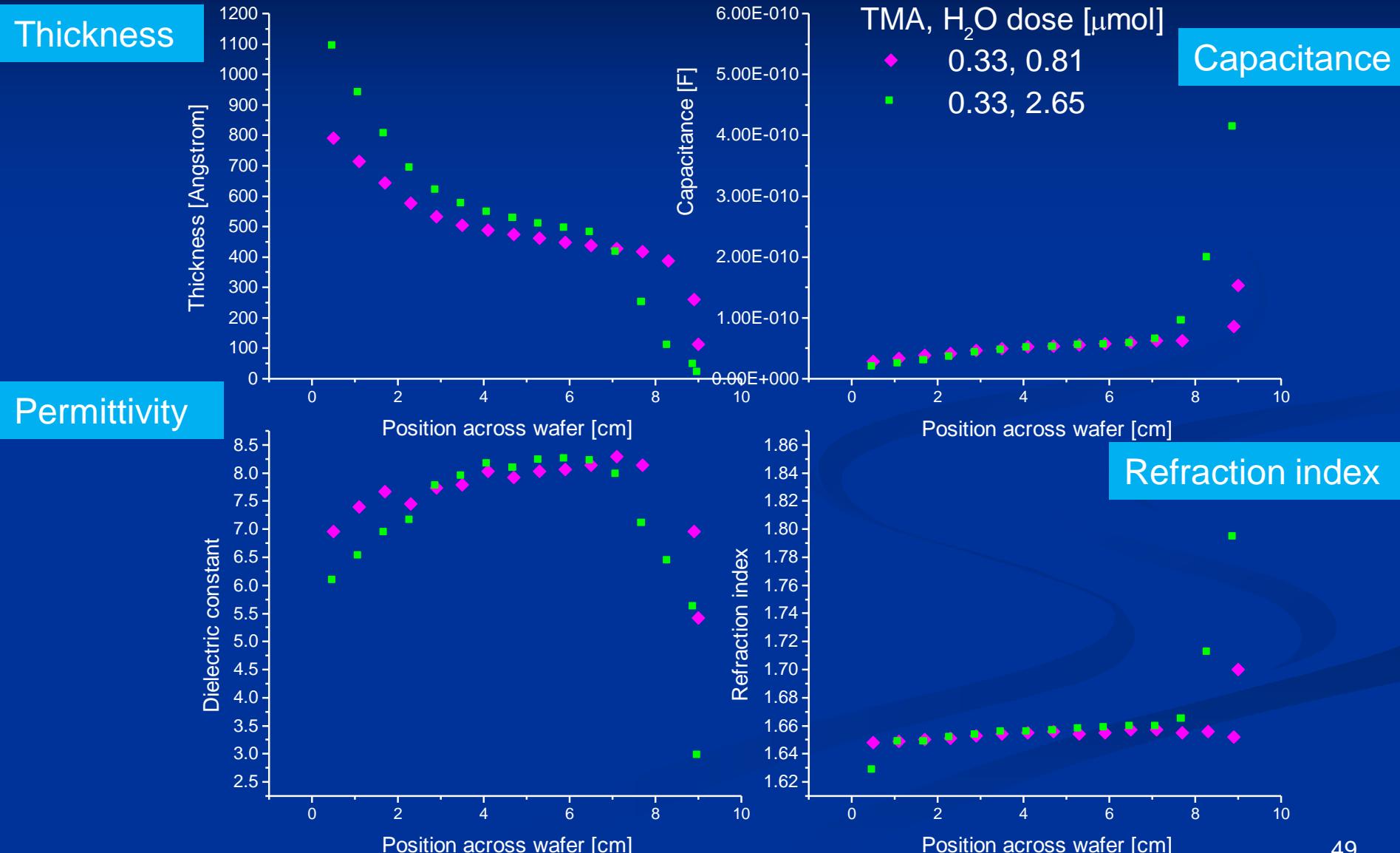
XPS



Film properties across wafer under water-starved conditions (0.8 μmol)

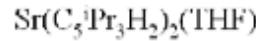
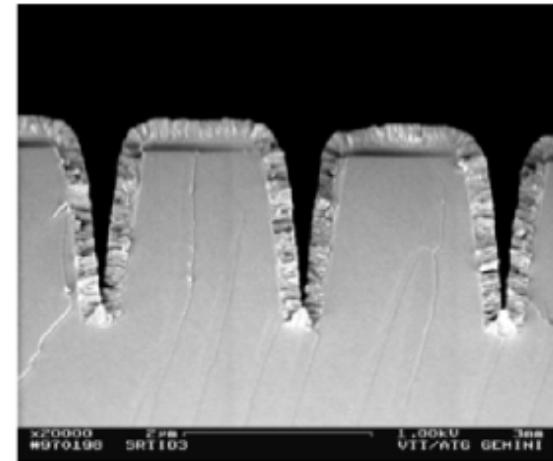
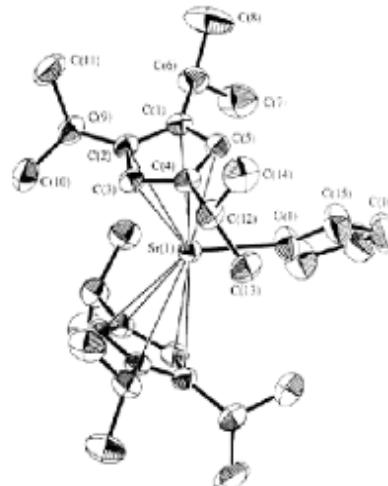
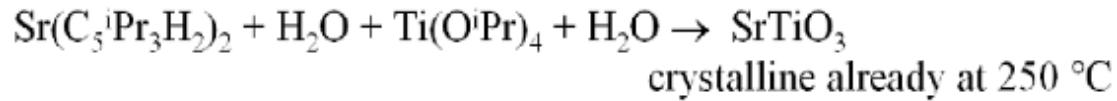
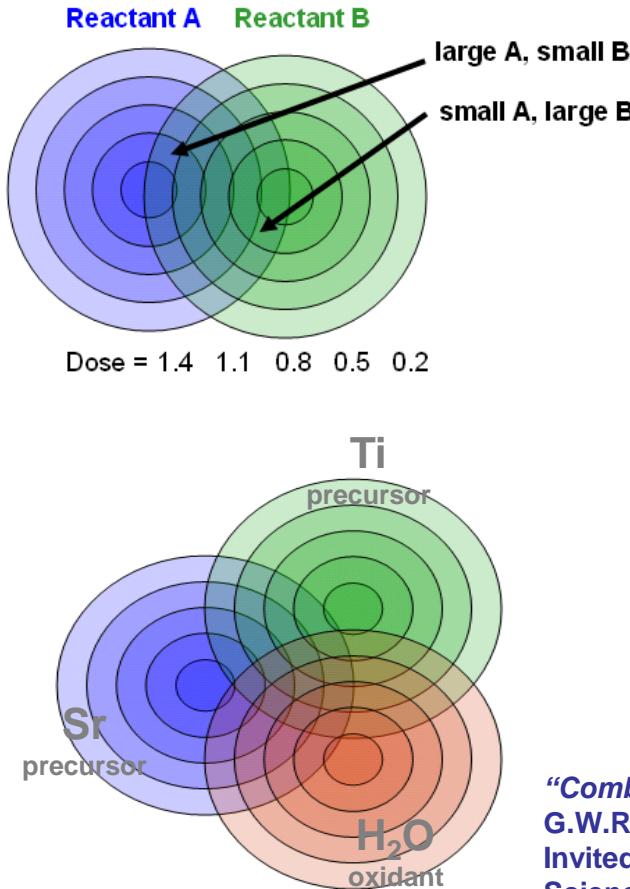


Film properties across wafer for TMA-underdosed films



Combinatorial ALD

Numerous process recipe permutations in real-world ALD



(Vehkamäki et al., *Electrochim. Solid State Lett.* 2 (1999) 505.)

"Combinatorial Chemical Synthesis of Thin Film Materials: CVD and ALD"
G.W.Rubloff, R.A.Adomaitis, L.Henn-Lecordier and M.Anderle
Invited to 4th International Workshop on Combinatorial Materials
Science and Technology, December 4-6, 2006, San Juan, Puerto Rico

Phleum pratense
Codolina



NONE

LEI

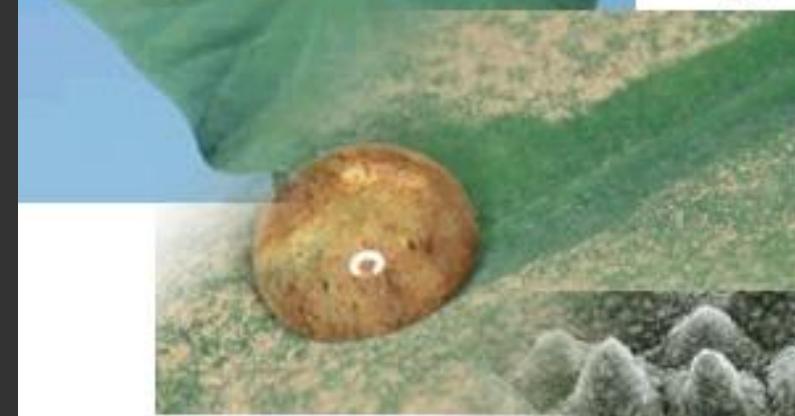
2.0kV

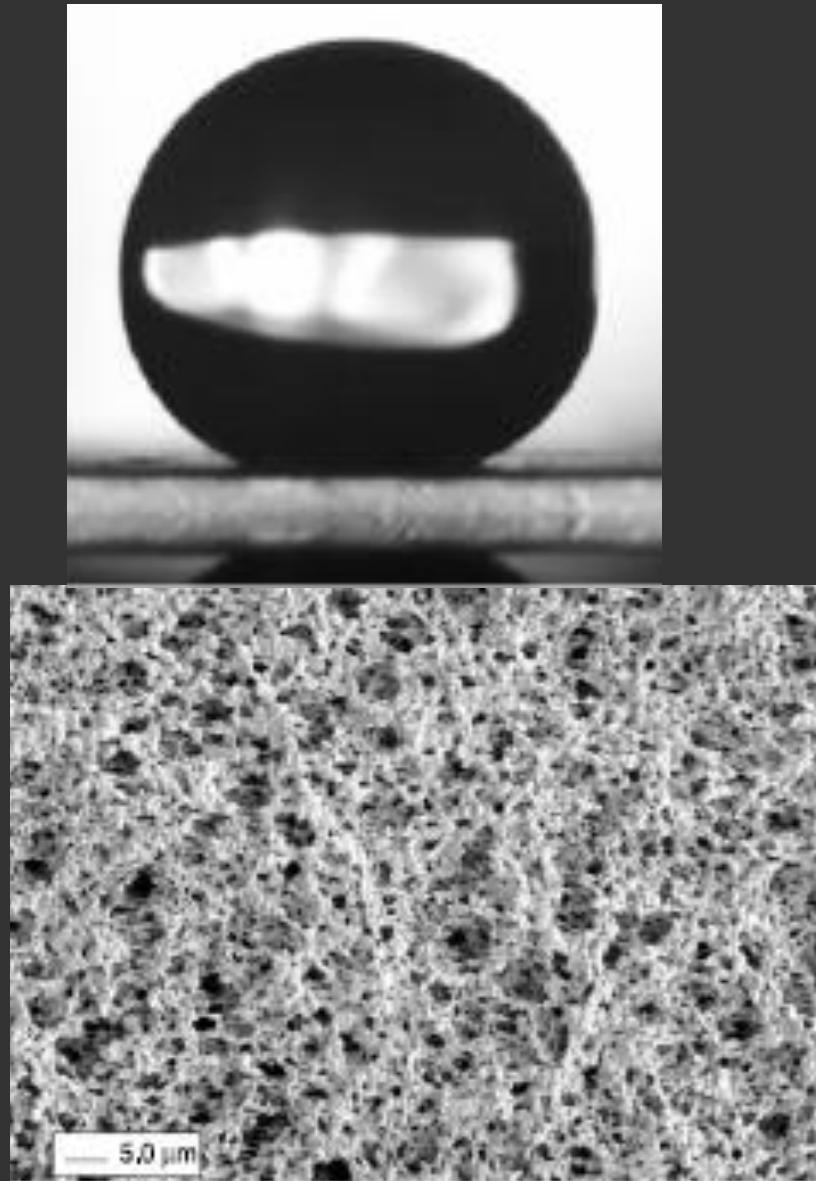
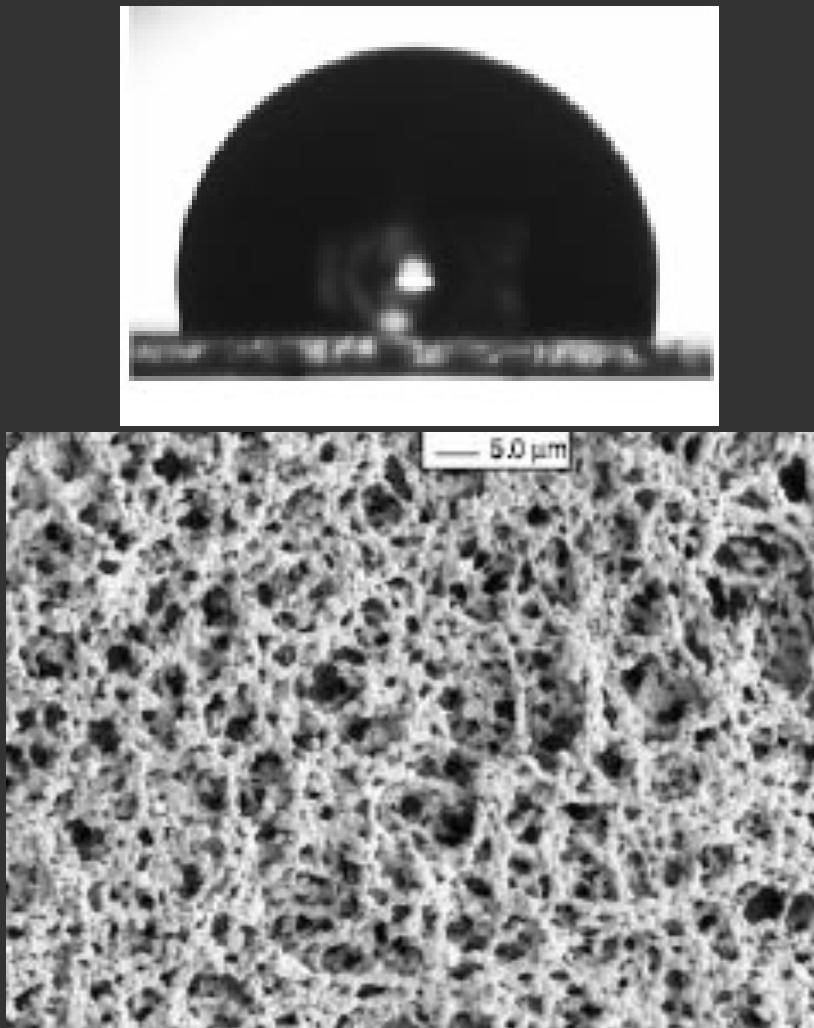
$\times 2,300$

10 μm

WD 24.3mm

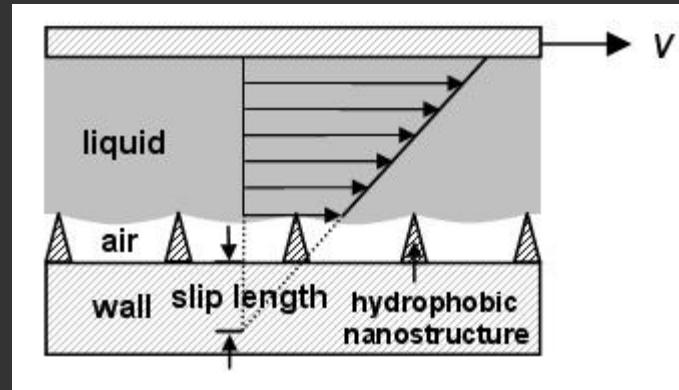
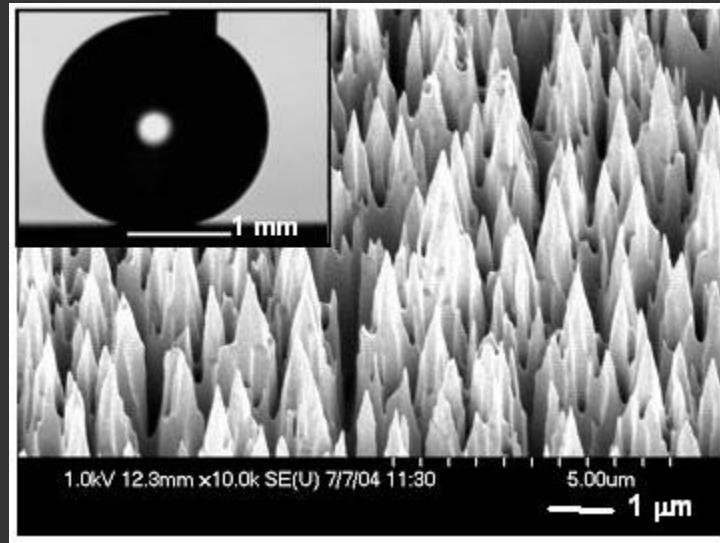
Loto



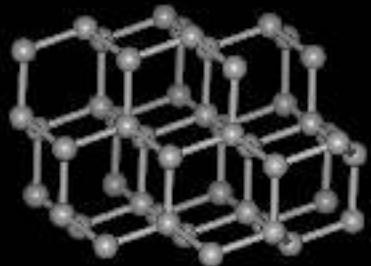


H.Y. Erbil et al,
“Transformation of a Simple Plastic into a Superhydrophobic Surface”
Science 299, 1377 (2003)

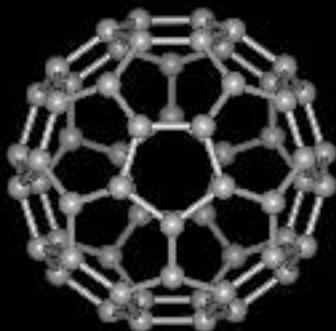
Superhydrophobic Surface



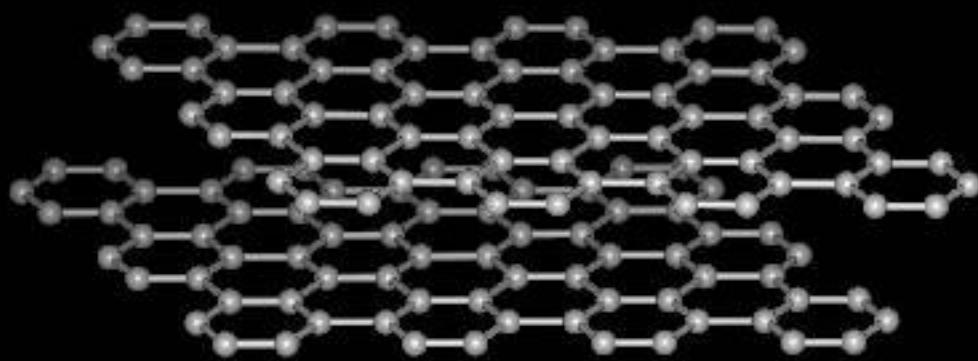
Choi and Kim, Physical Review Letters 2006



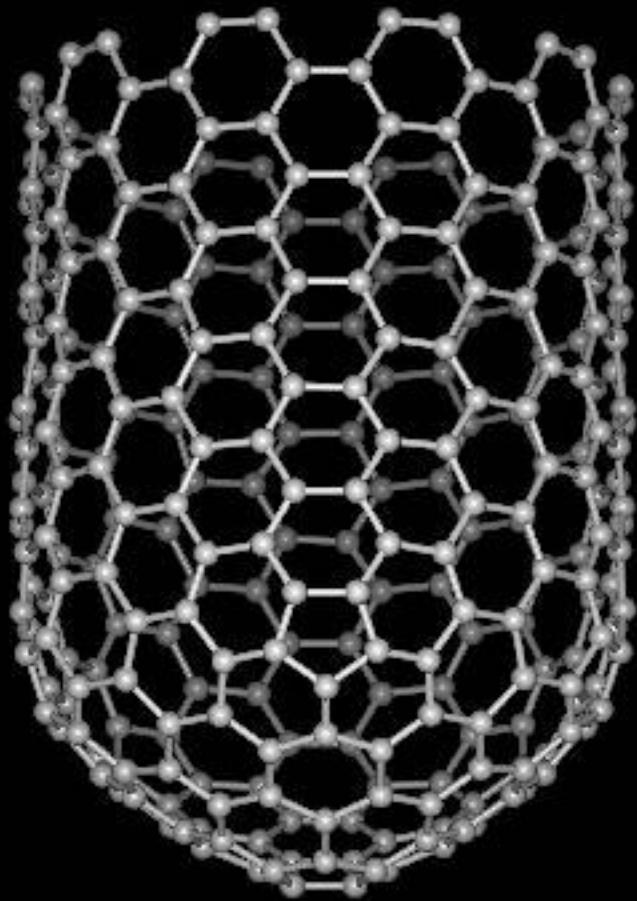
diamante



fullerene

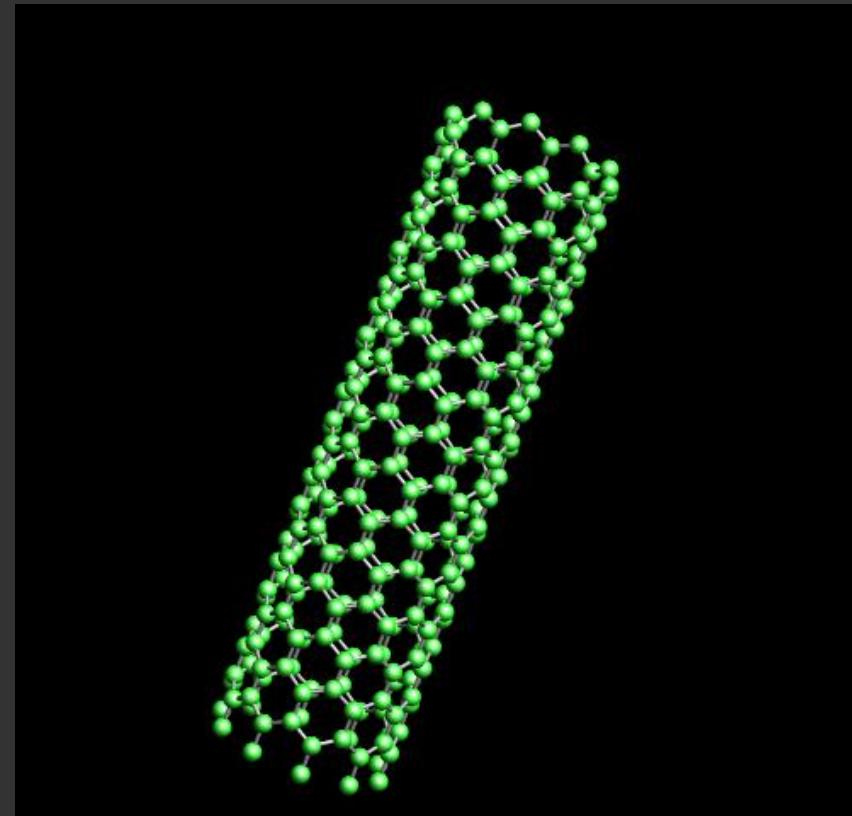
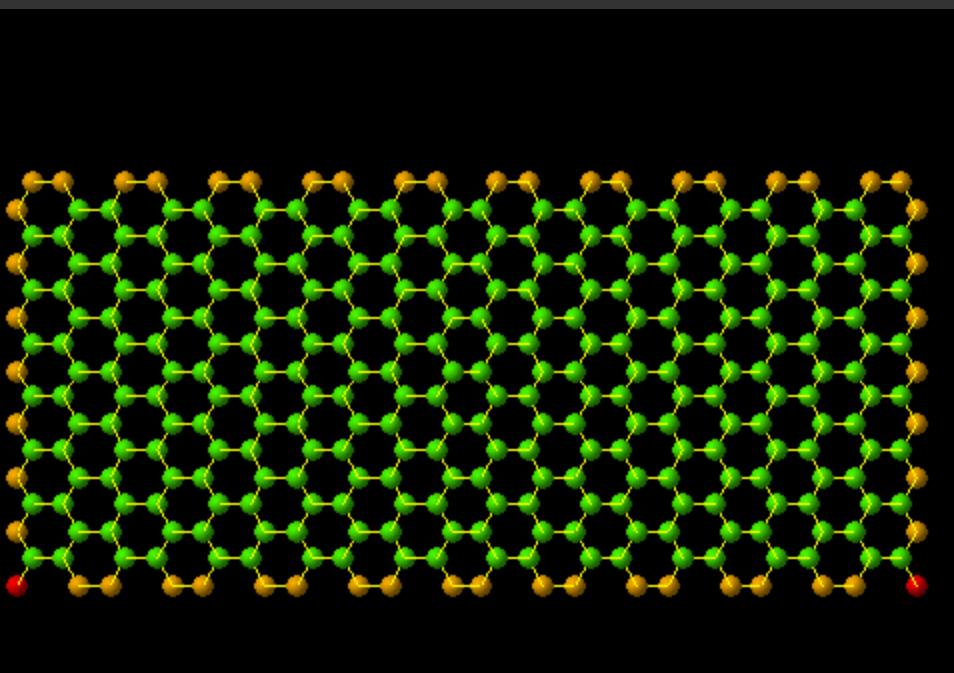


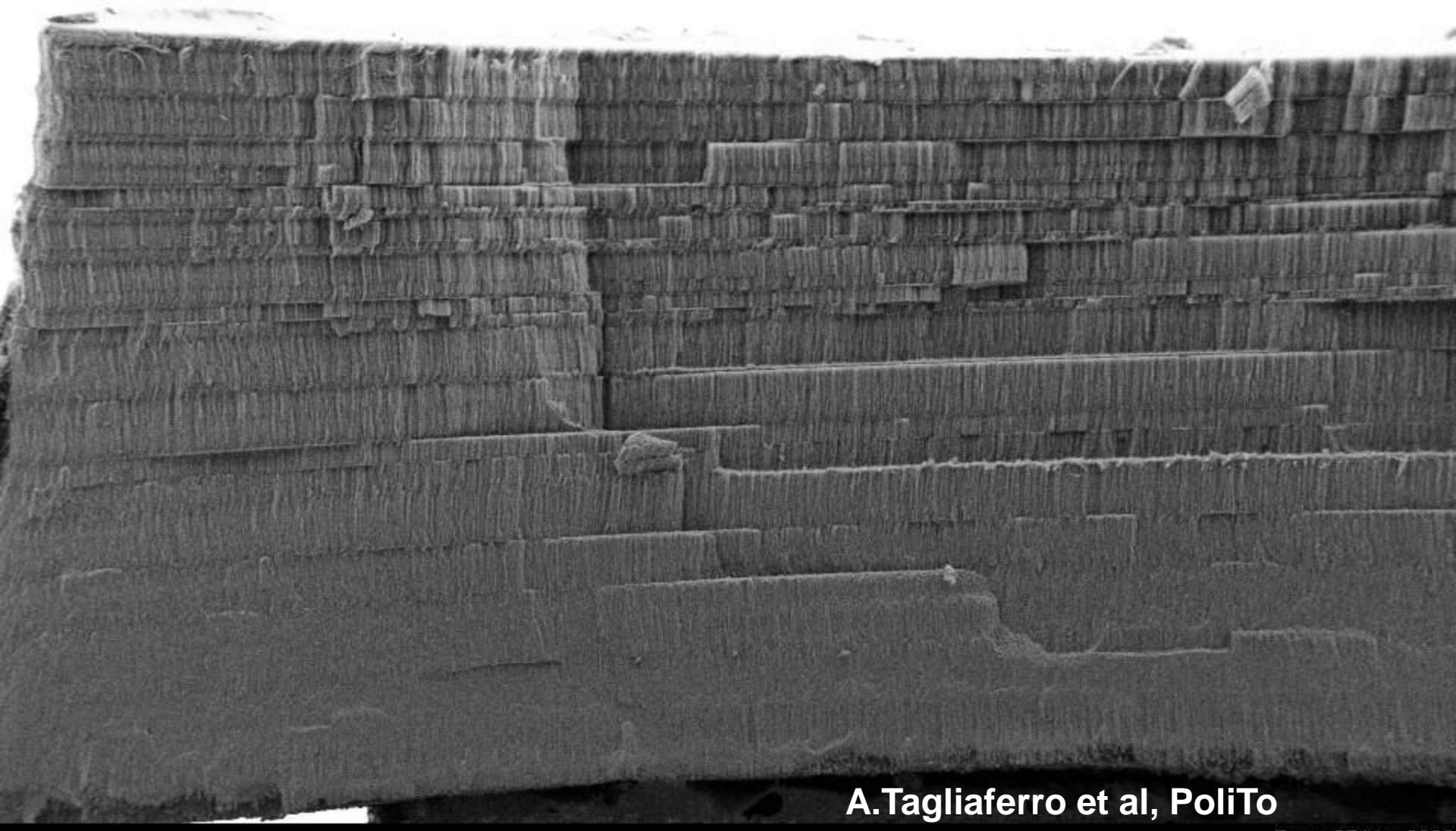
grafite



nanotubo

Nanotubi di carbonio





A.Tagliaferro et al, PoliT

200µm
H

WD = 5 mm

Aperture Size = 20.00 µm

Signal A = SE2

Date : 27 Mar 2006

Mag = 87 X

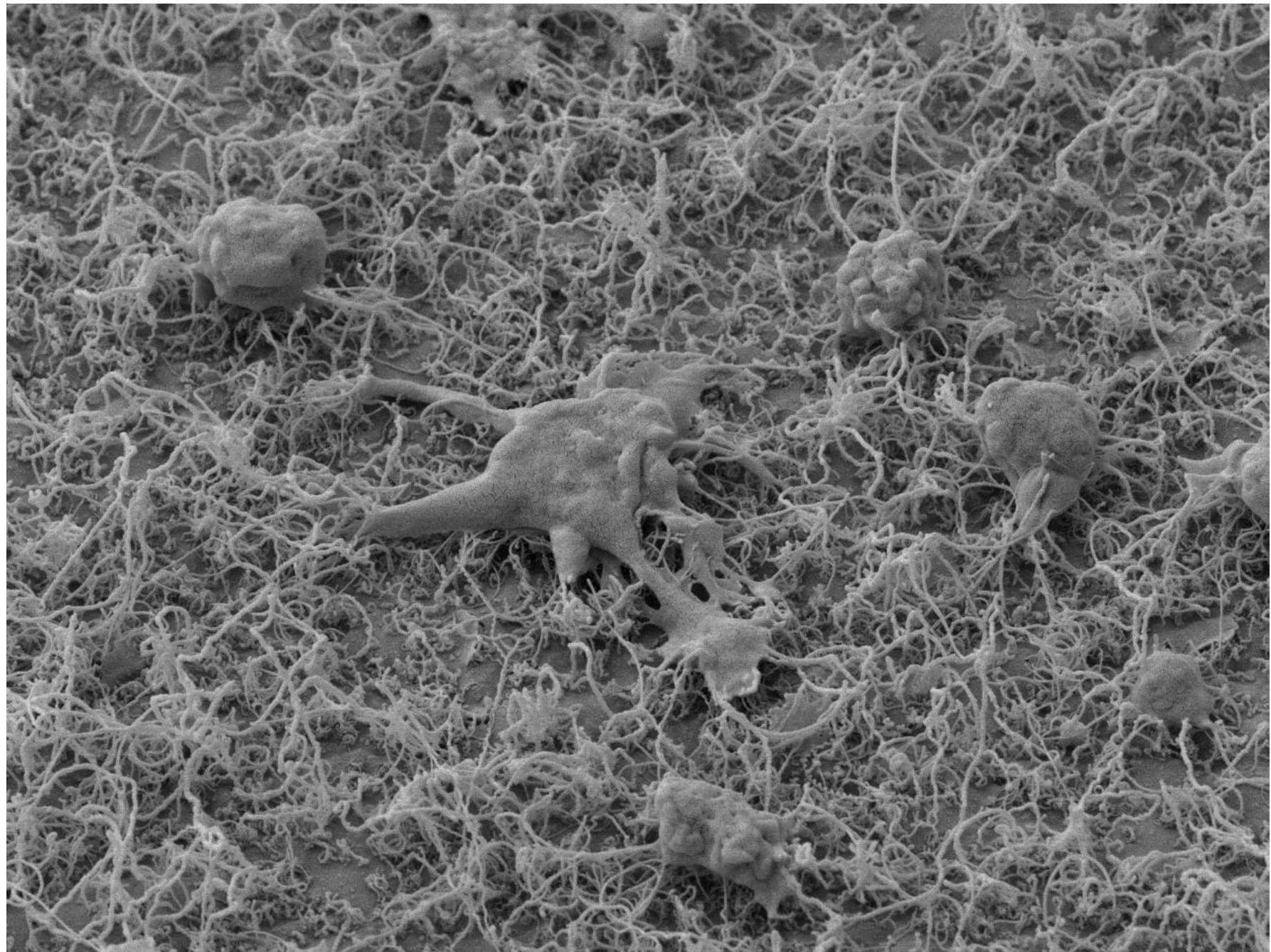
EHT = 5.00 kV

Stage at T = 45.0 °

Time : 11:51:03

User Name = ANGELI





NONE

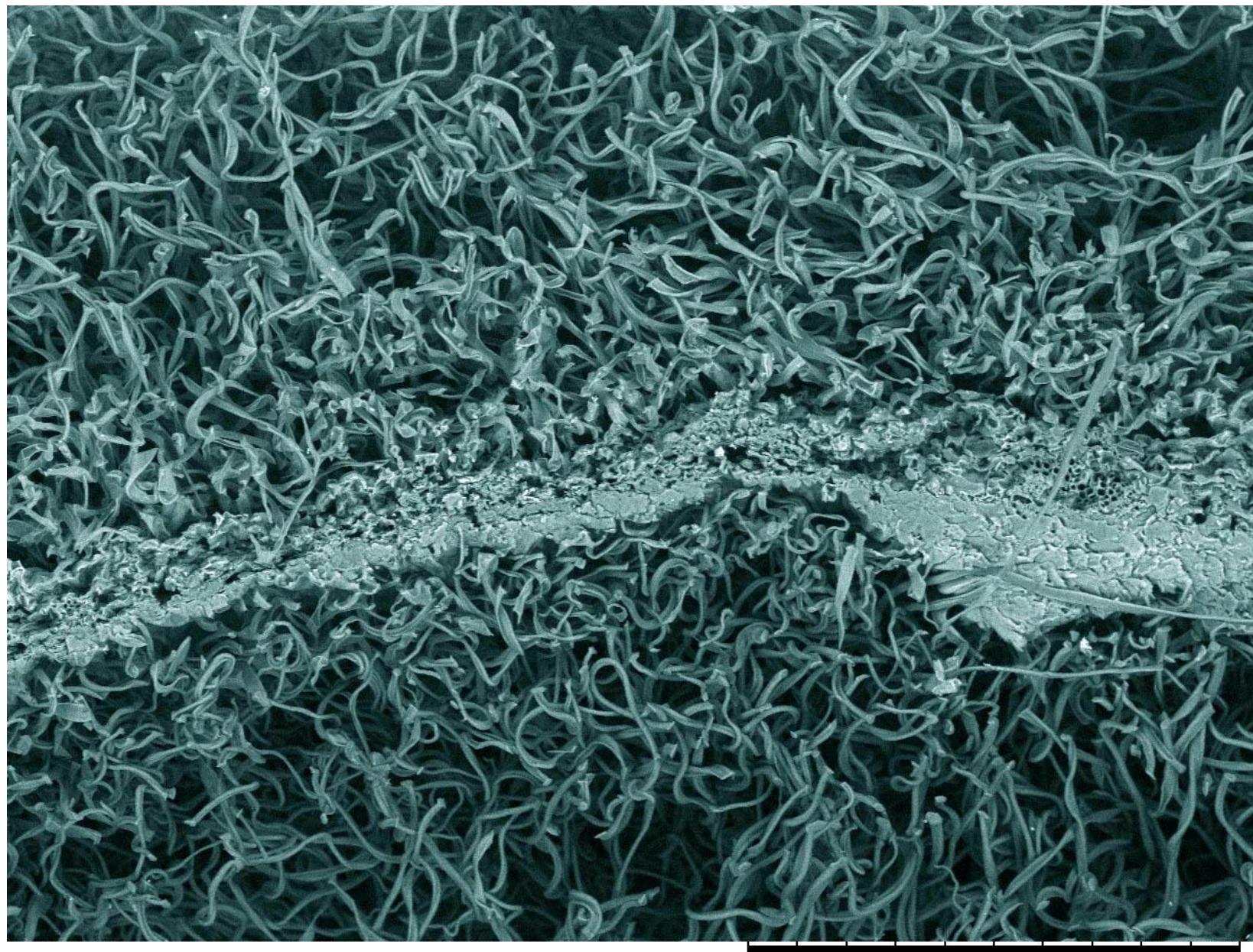
LEI

5.0kV

X6,000

1 μ m

WD 14.2mm

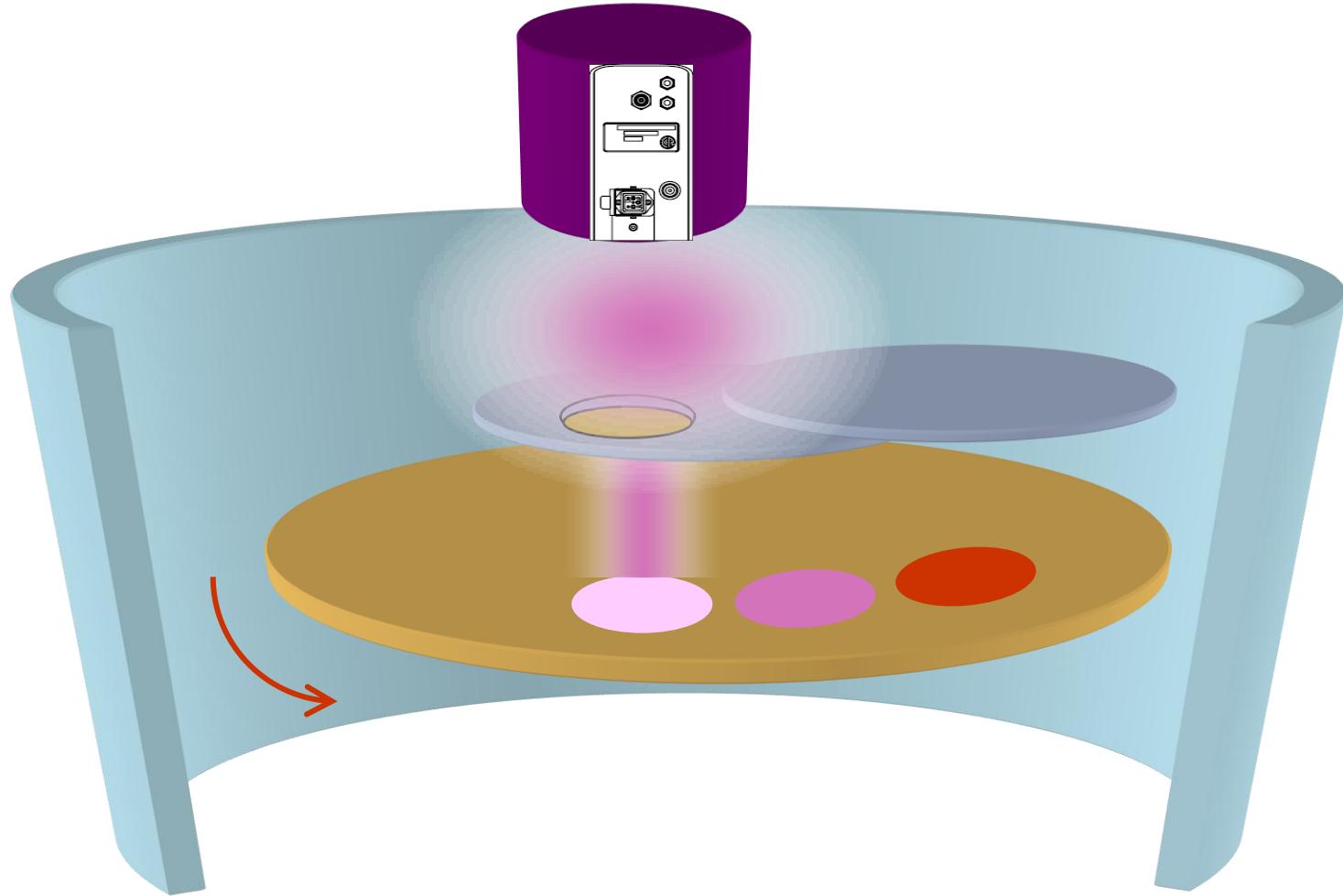


CEALP_1709

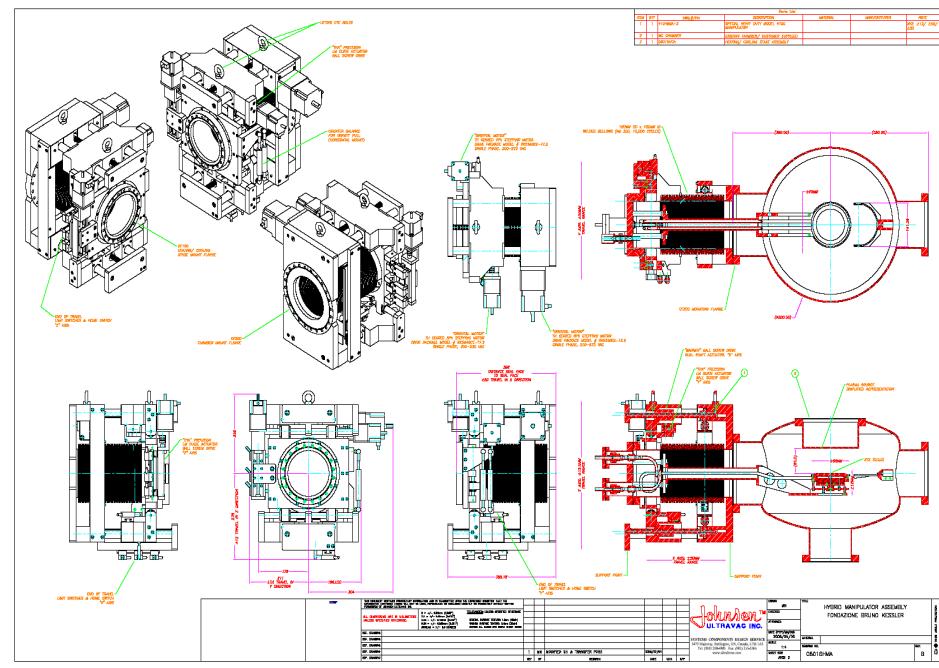
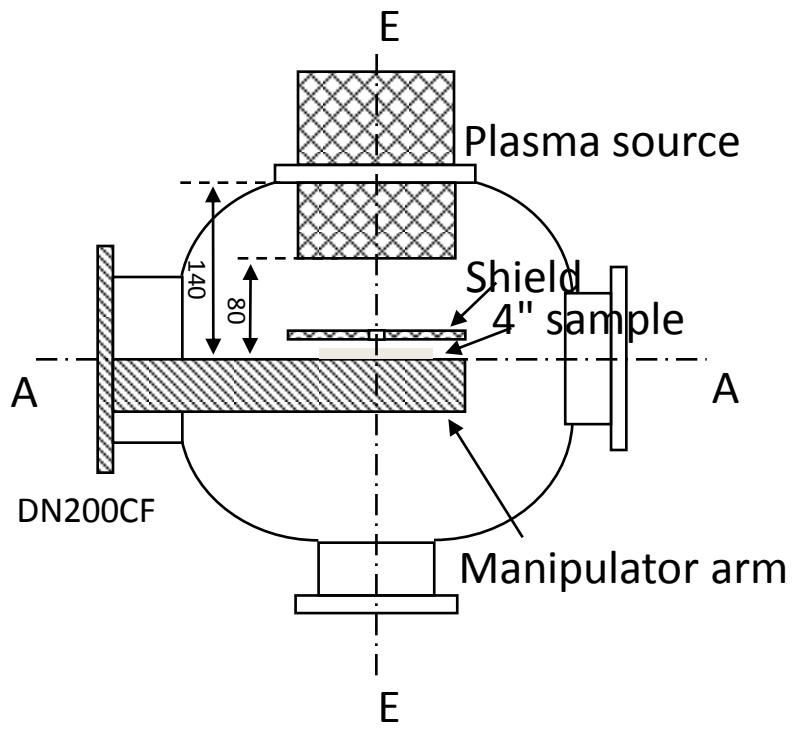
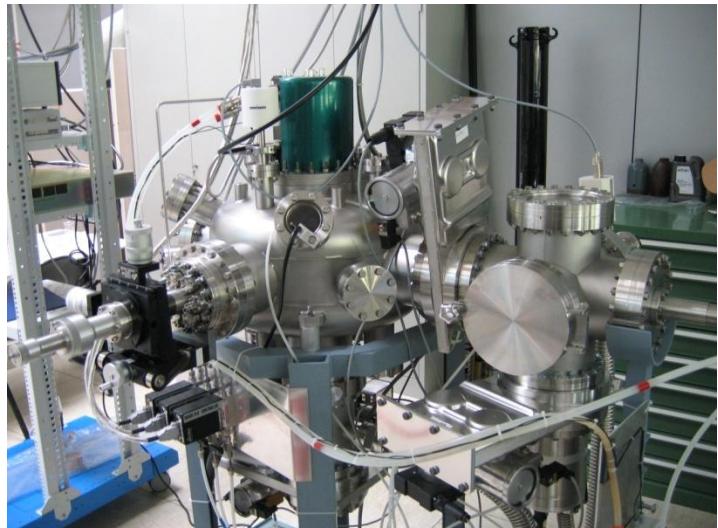
L x250

300 um

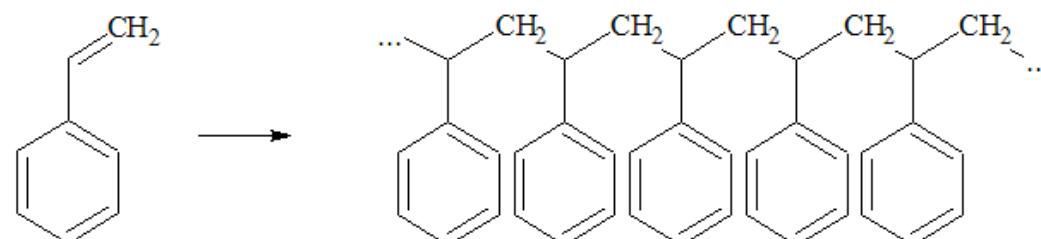
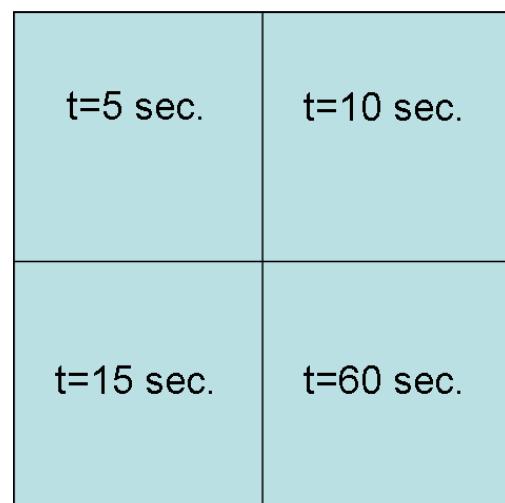
Dolomiti Invisibili 2006



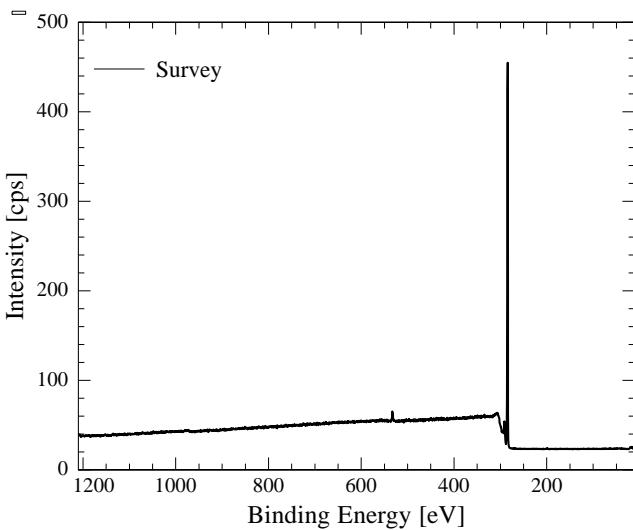
Combi Approach to Biosurfaces



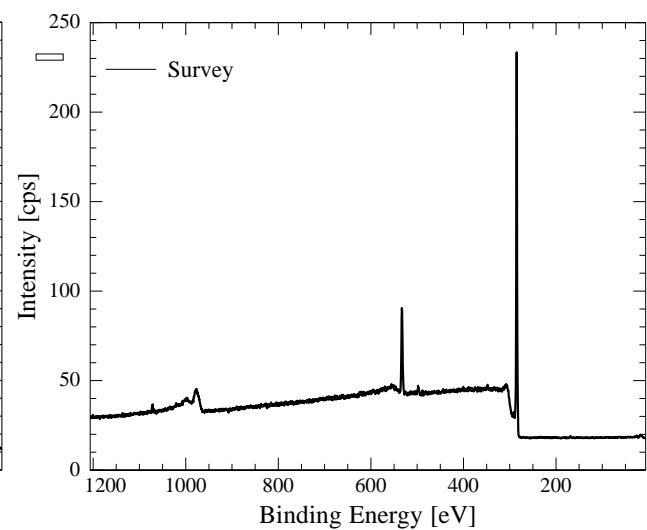
Substrate Size	Ø 50 mm (recommended)
	2x10-4 to 5x10-2 mbar
Plasma Density	10⁹ to 10¹² cm⁻³
Ion Current Density	0,01 to 1,0 mA/cm²
Ion Energy	30 to 200 eV
Frequency	13,56 MHz
Max Power	600 W
Process Gas	O₂, N₂, CO₂, NH₃, Ar and other
Uniformity	± 10%



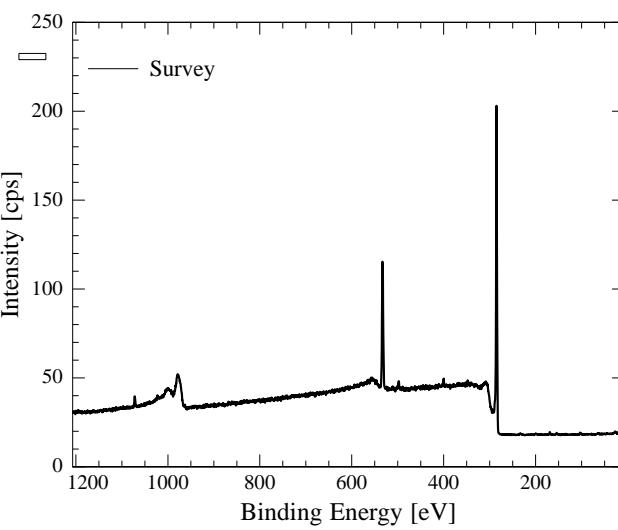
Filename: COMBIREF



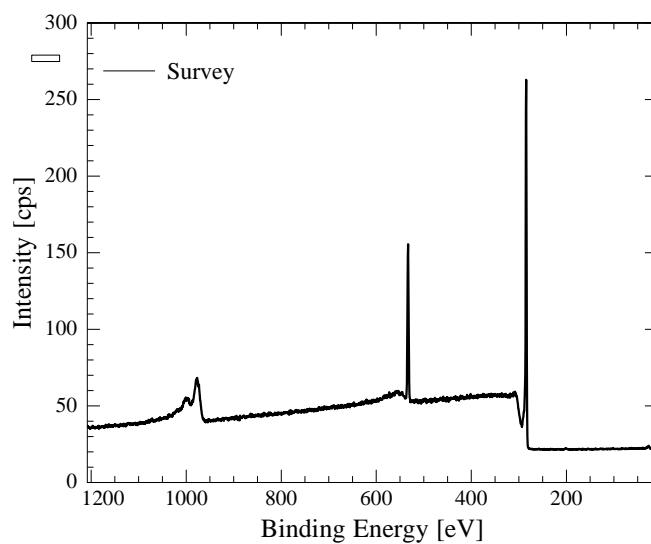
Filename: COMBI44



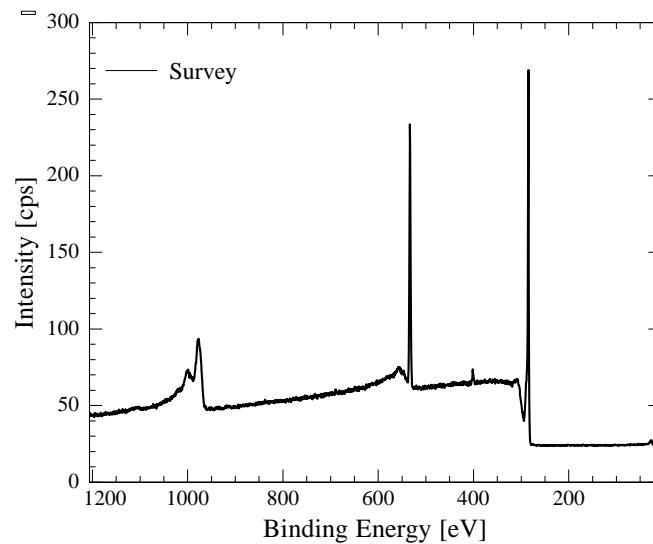
Filename: COMBI41



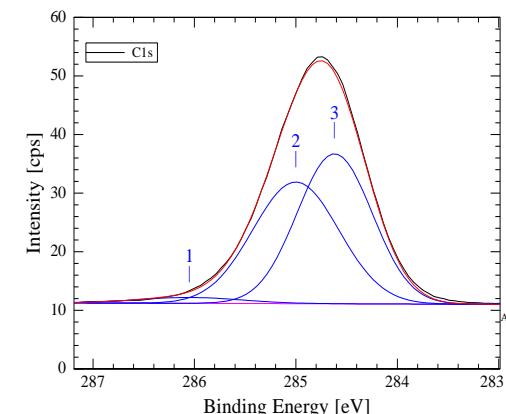
Filename: COMBI21



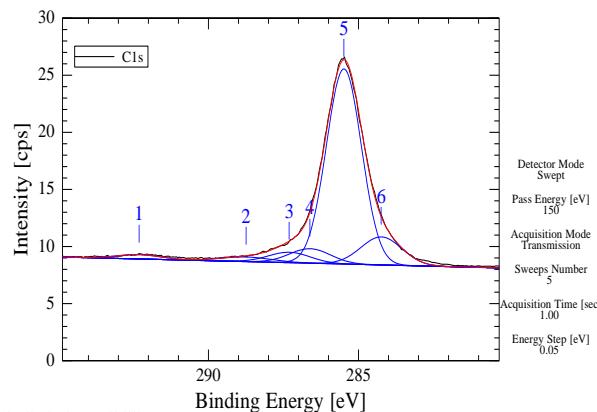
Filename: COMBI1



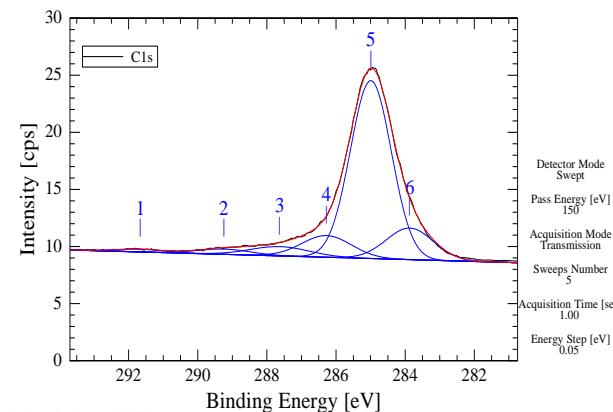
FileName: COMBIREF.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente : C1s001	25.161	286.05	0.996	1.20	1.00	0.00	2.62
Componente : C1s002	441.683	285.00	20.47	1.00	1.00	0.00	46.04
Componente : C1s003	474.260	284.62	25.606	0.87	1.00	0.00	49.43



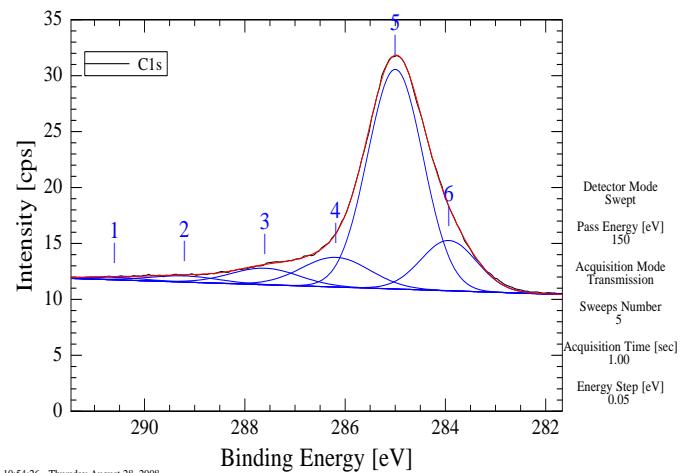
FileName: COMBI44.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente : C1s001	13.497	292.31	0.373	1.70	1.00	0.00	1.91
Componente : C1s002	17.007	288.74	0.399	2.00	1.00	0.00	2.40
Componente : C1s003	31.613	287.30	0.911	1.63	1.00	0.00	4.47
Componente : C1s004	43.127	286.61	1.266	1.60	1.00	0.00	6.10
Componente : C1s005	499.122	285.49	17.088	1.37	1.00	0.00	70.55
Componente : C1s006	83.625	284.24	2.455	1.60	1.00	0.00	11.82



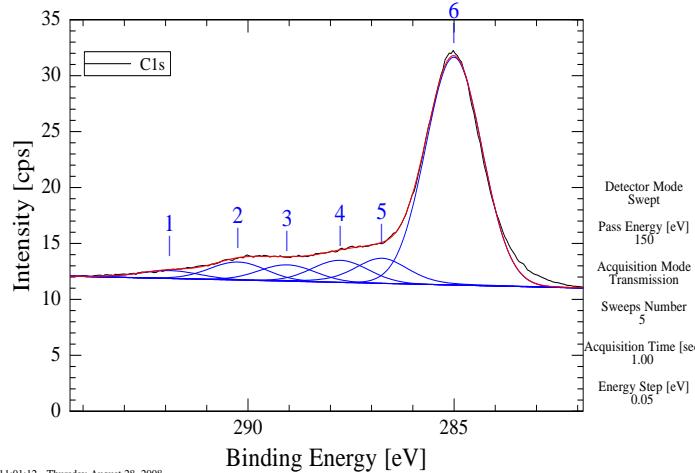
FileName: COMBI41.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente : C1s001	8.300	291.66	0.244	1.60	1.00	0.00	1.20
Componente : C1s002	17.698	289.24	0.460	1.81	1.00	0.00	2.56
Componente : C1s003	34.637	287.63	0.814	2.00	1.00	0.00	5.02
Componente : C1s004	68.018	286.27	1.896	1.69	1.00	0.00	9.86
Componente : C1s005	462.535	284.99	15.577	1.39	1.00	0.00	67.02
Componente : C1s006	94.103	283.87	2.763	1.60	1.00	0.00	13.63



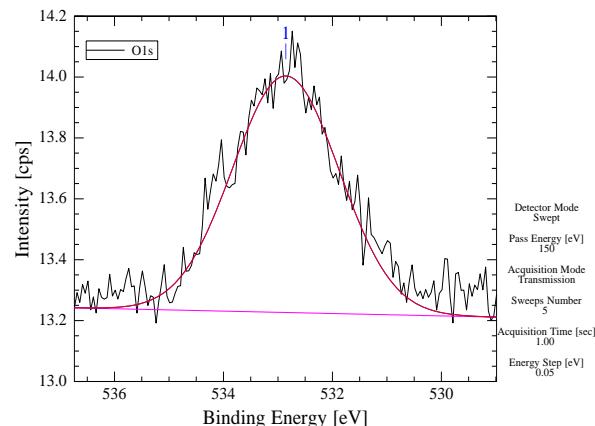
FileName: COMBI21.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente : C1s001	6.319	290.60	0.226	1.41	1.00	0.00	0.76
Componente : C1s002	18.793	289.21	0.552	1.60	1.00	0.00	2.25
Componente : C1s003	50.335	287.60	1.478	1.60	1.00	0.00	6.02
Componente : C1s004	90.482	286.19	2.656	1.60	1.00	0.00	10.81
Componente : C1s005	543.707	285.00	19.645	1.30	1.00	0.00	64.97
Componente : C1s006	124.446	283.93	4.497	1.30	1.00	0.00	14.87



FileName: COMBI11.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente : C1s001	25.433	291.90	0.703	1.70	1.00	0.00	2.57
Componente : C1s002	54.858	290.25	1.610	1.60	1.00	0.00	5.54
Componente : C1s003	49.713	289.05	1.459	1.60	1.00	0.00	5.02
Componente : C1s004	67.200	287.76	1.973	1.60	1.00	0.00	6.78
Componente : C1s005	79.468	286.75	2.247	1.50	0.70	0.00	8.02
Componente : C1s006	694.055	285.01	20.376	1.60	1.00	0.00	70.04

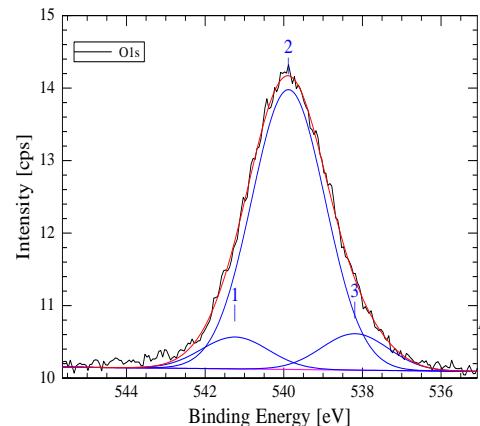


FileName:	COMBIREF.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente :	O1s001	41.431	532.85	0.777	2.30	1.00	0.00	93.96



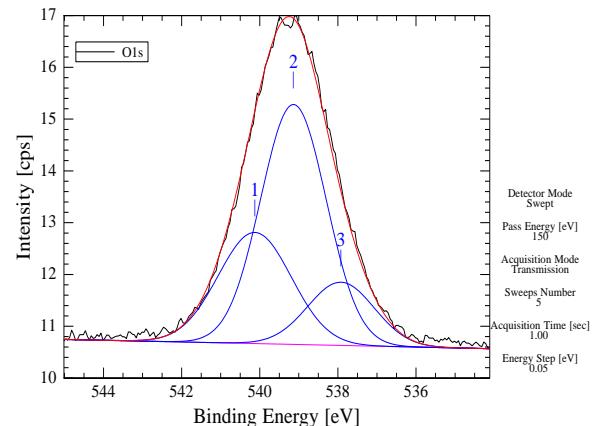
09:18:46 - Thursday August 28, 2008

FileName:	COMBI44.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente :	O1s001	18.695	541.24	0.439	2.00	1.00	0.00	8.23
Componente :	O1s002	180.766	539.88	3.860	2.20	1.00	0.00	79.53
Componente :	O1s003	21.385	538.19	0.502	2.00	1.00	0.00	9.41



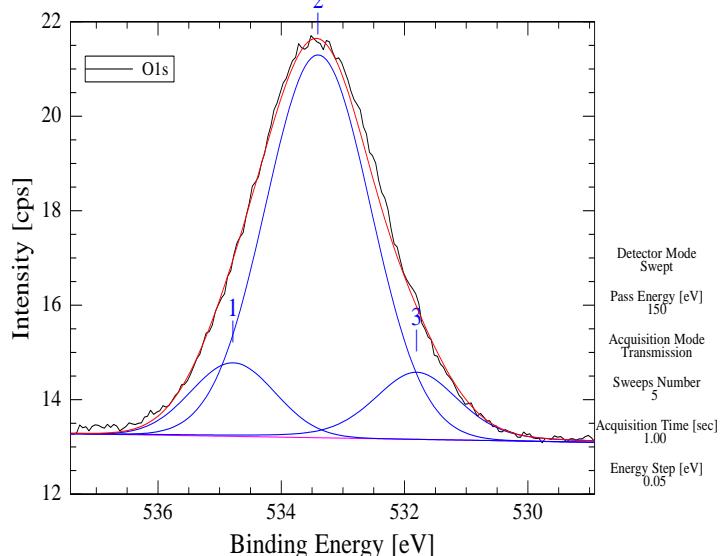
10:11:33 - Thursday August 28, 2008

FileName:	COMBI41.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente :	O1s001	100.472	540.13	2.145	2.20	1.00	0.00	27.96
Componente :	O1s002	194.617	539.14	4.632	1.97	1.00	0.00	54.16
Componente :	O1s003	51.962	537.92	1.220	2.00	1.00	0.00	14.46



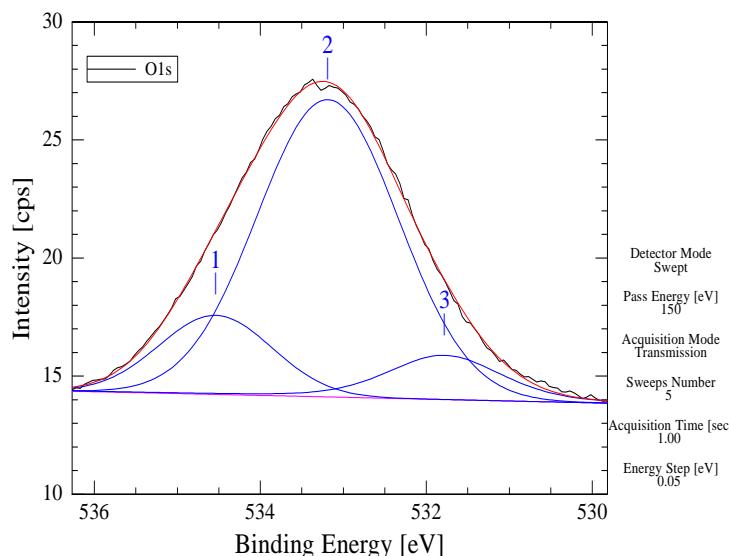
09:57:54 - Thursday August 28, 2008

FileName:	COMBI21.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente :	O1s001	53.031	534.78	1.557	1.60	1.00	0.00	11.49
Componente :	O1s002	345.128	533.40	8.106	2.00	1.00	0.00	74.79
Componente :	O1s003	52.532	531.81	1.421	1.60	0.70	0.00	11.38

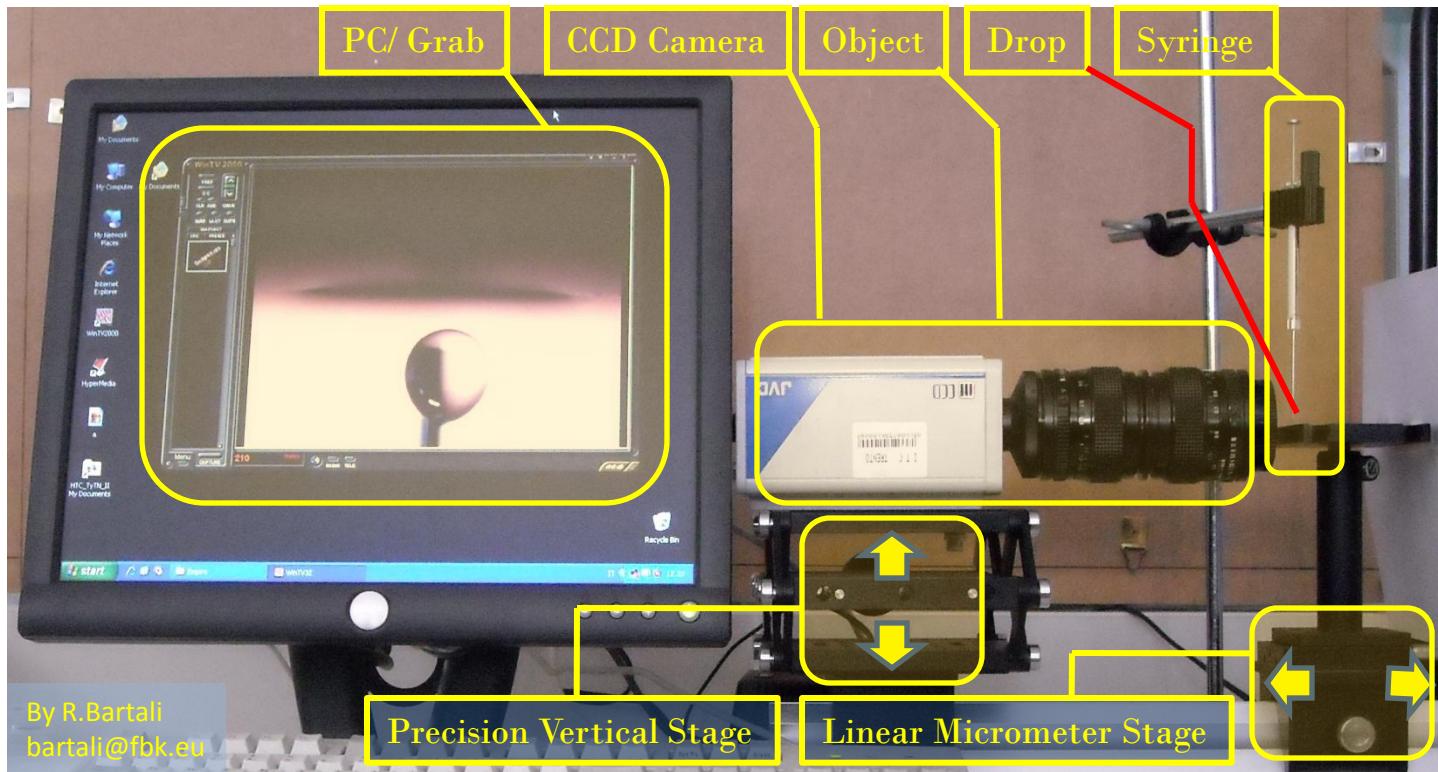


09:40:14 - Thursday August 28, 2008

FileName:	COMBI11.PXT	Area (cps)	BE(eV)	Height	Fwhm	MixGL	Asym	% Tot.
Componente :	O1s001	113.383	534.54	3.345	1.60	1.00	0.00	15.73
Componente :	O1s002	535.809	533.19	12.586	2.00	1.00	0.00	74.34
Componente :	O1s003	67.788	531.78	1.872	1.60	0.70	0.00	9.41

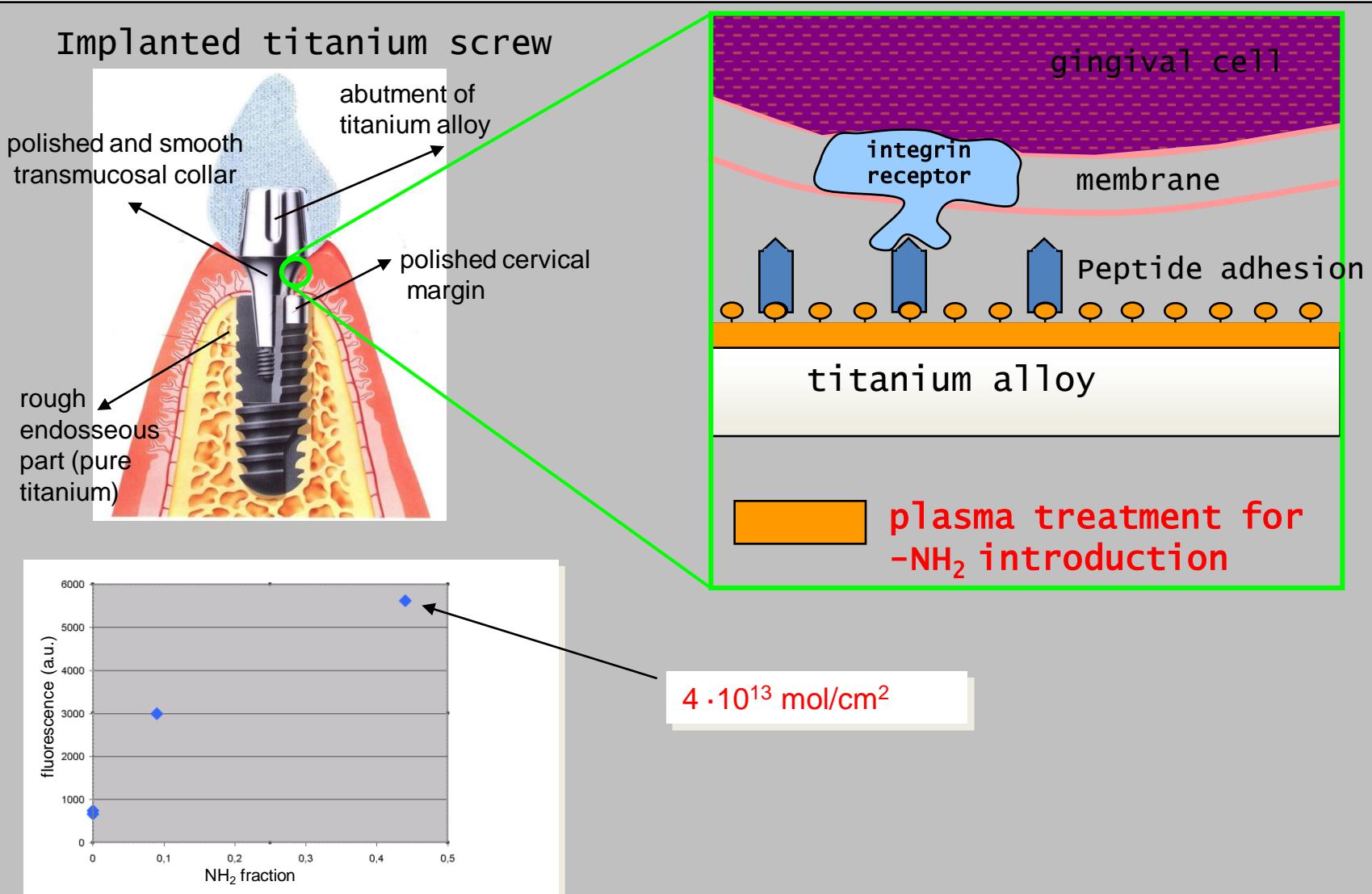


09:12:55 - Monday December 03, 2007





ADESIONE DELLA GENGIVA AD UN PROTESI DENTALE

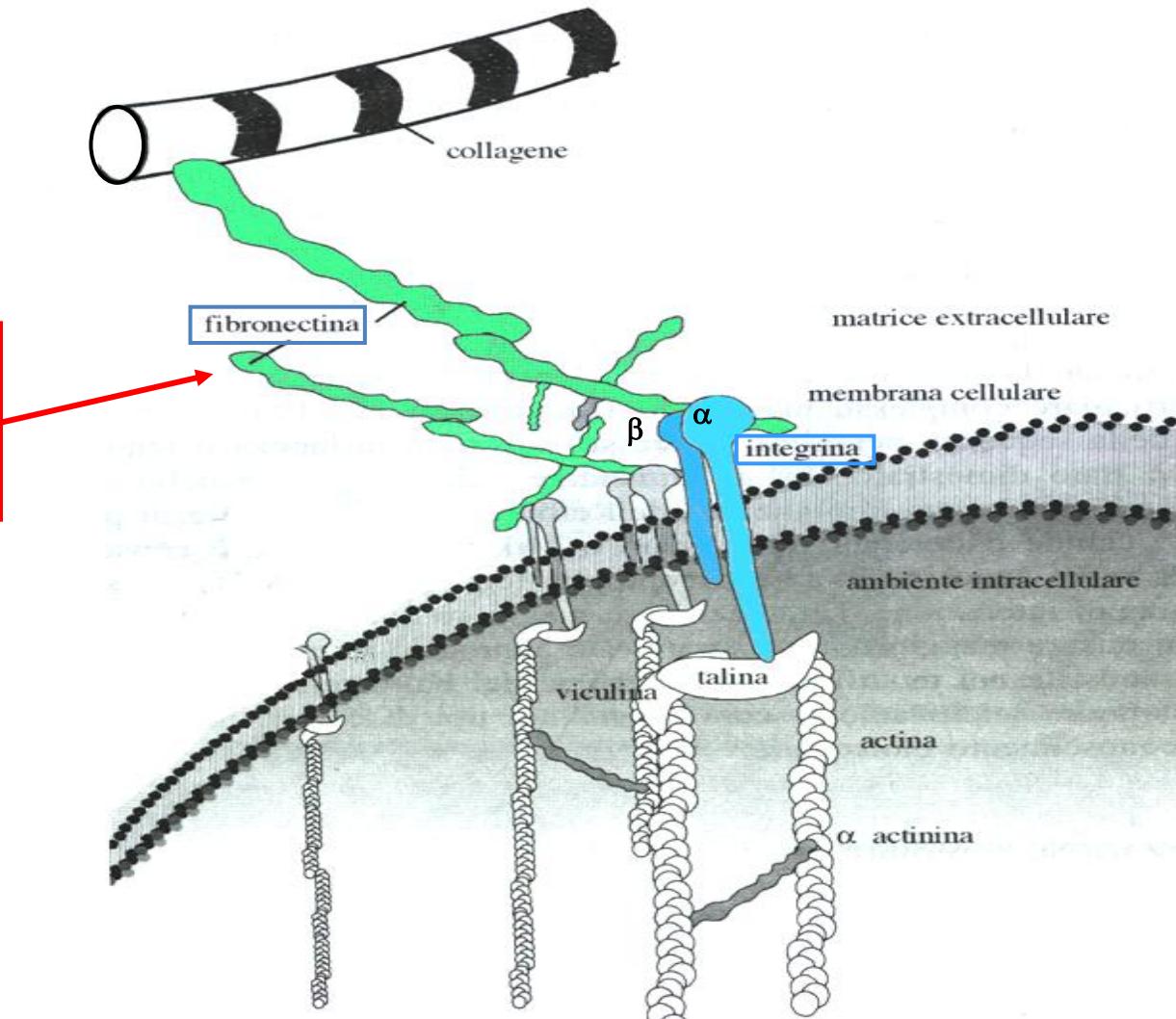


ECM proteins and integrin receptors

Fibronectin

**GLY-ARG-GLY-ASP-
-SER-TYR-CYS**

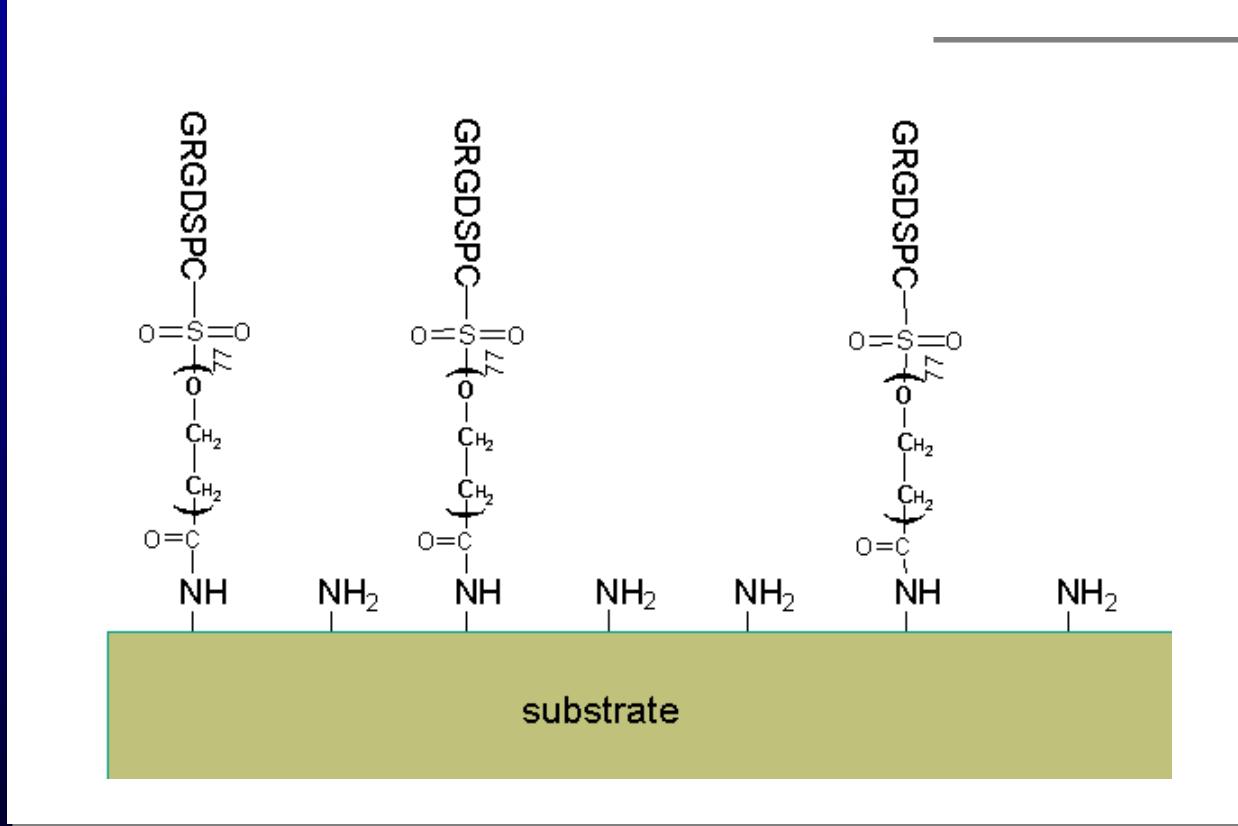
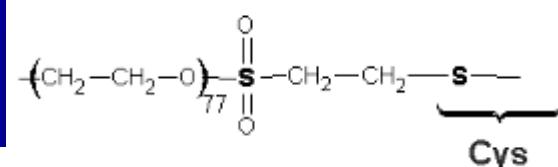
**RGD
Adhesion
Peptide**



Titanium alloy functionalization: Overview

Step 1: amide bond through the N-hydroxysuccinimide ester (*NHS*)

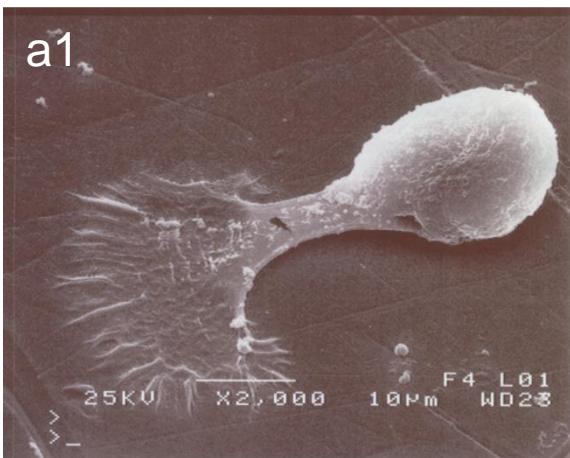
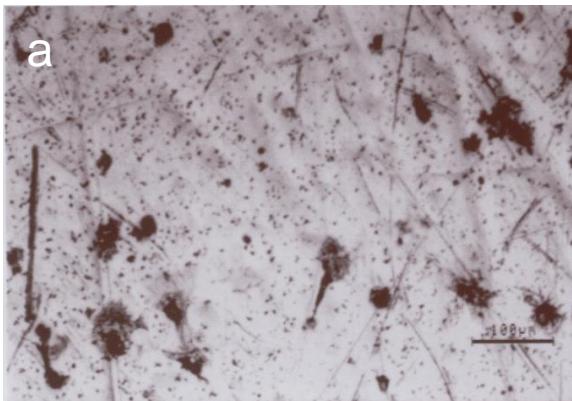
Step 2: thiol chemistry (*Vinylsulfone*)



Fluorescent derivative PEG: $5.5 \cdot 10^{13}$ molecules/cm²

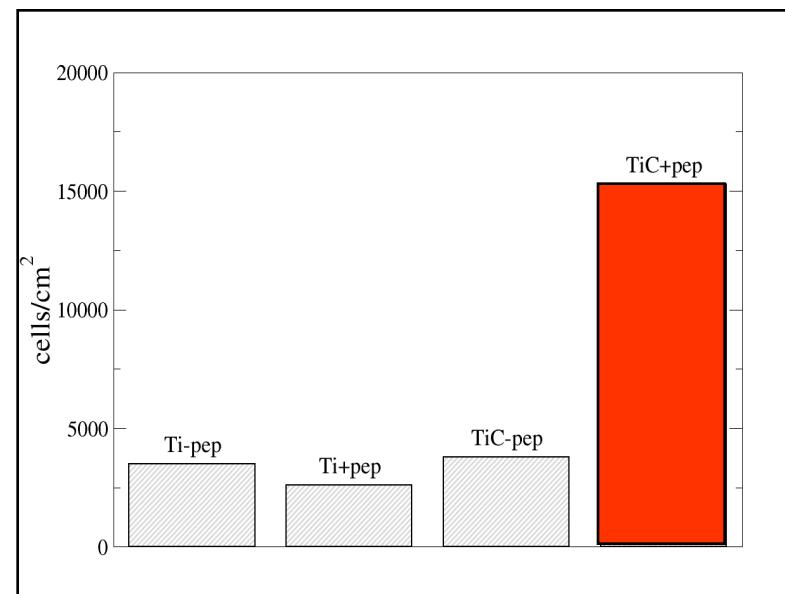
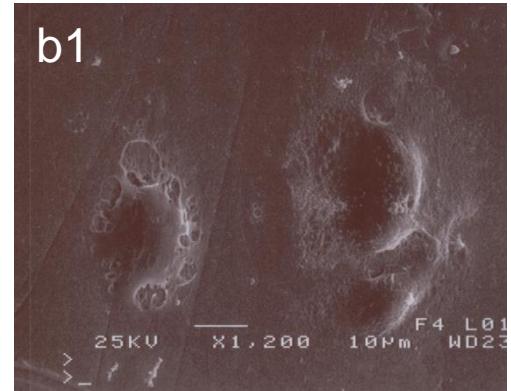
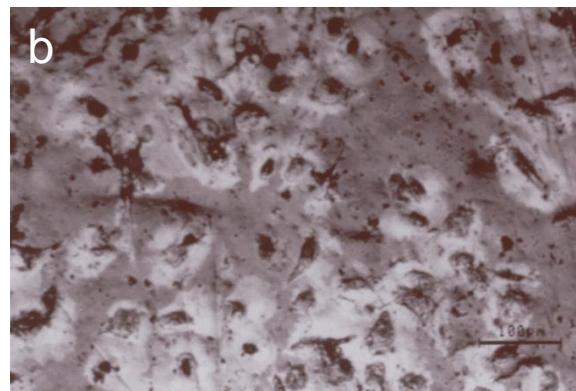
Titanium alloy functionalization 3: Human gingival cells (HGF-1) adhesion

titanium alloy



Cell images obtained with a laser scan microscope (a and b) and with a scanning electron microscope (a1 and b1)

RGD modified titanium alloy

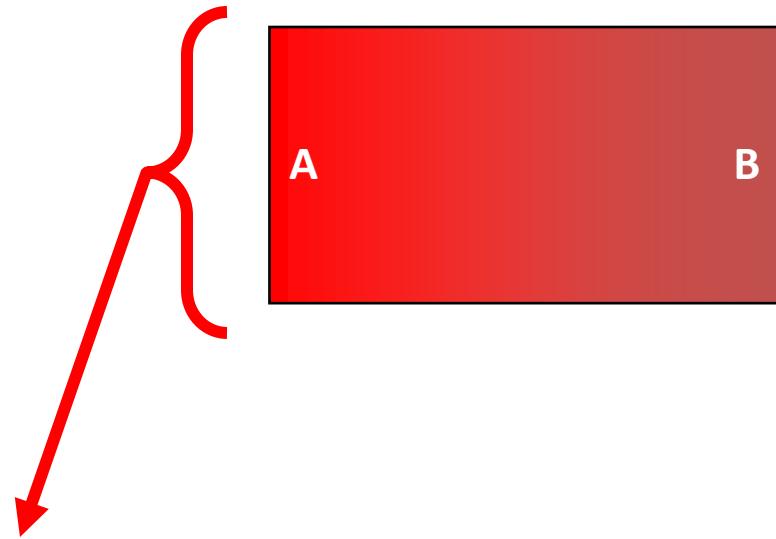
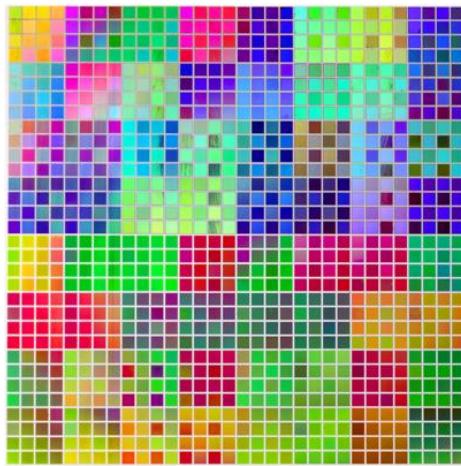


17,500 cell/cm²

Incubation 24 h in serum free medium plus cycloheximide (25 ug/ml)

Various combinatorial experimental designs

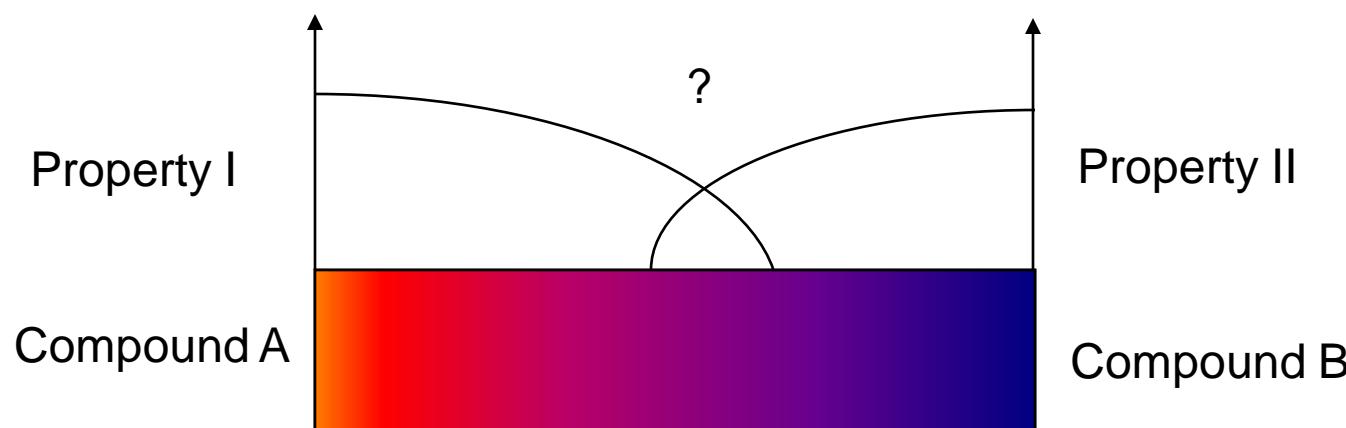
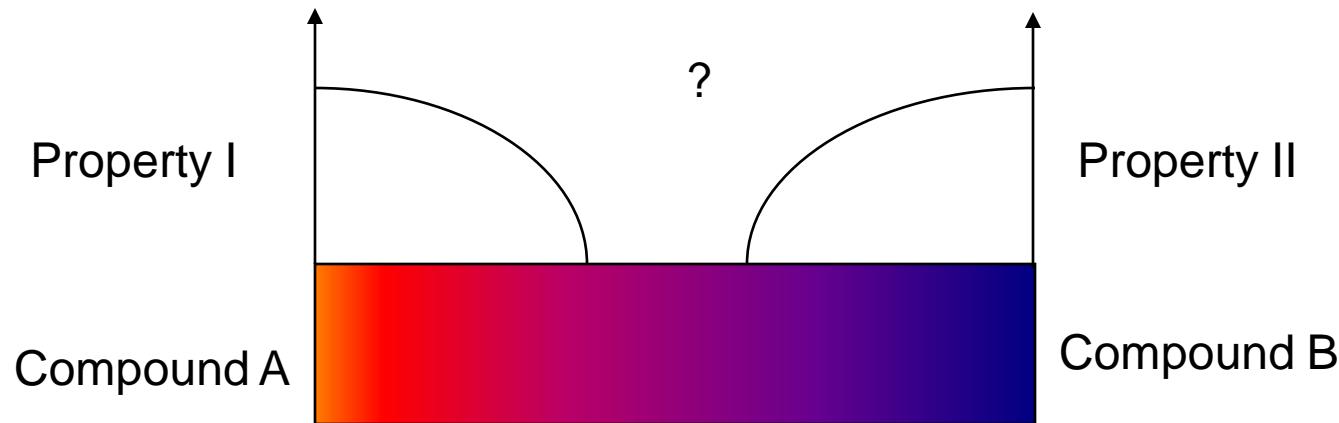
discrete libraries vs composition spreads



Composition spreads allow continuous mapping of physical properties and phase boundaries

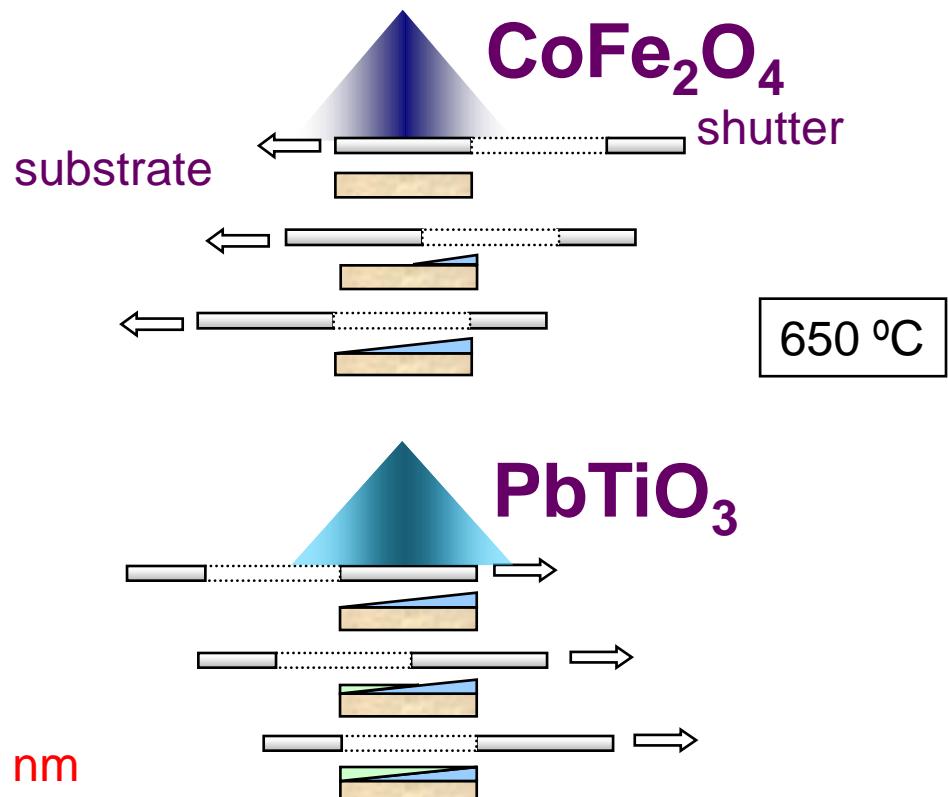
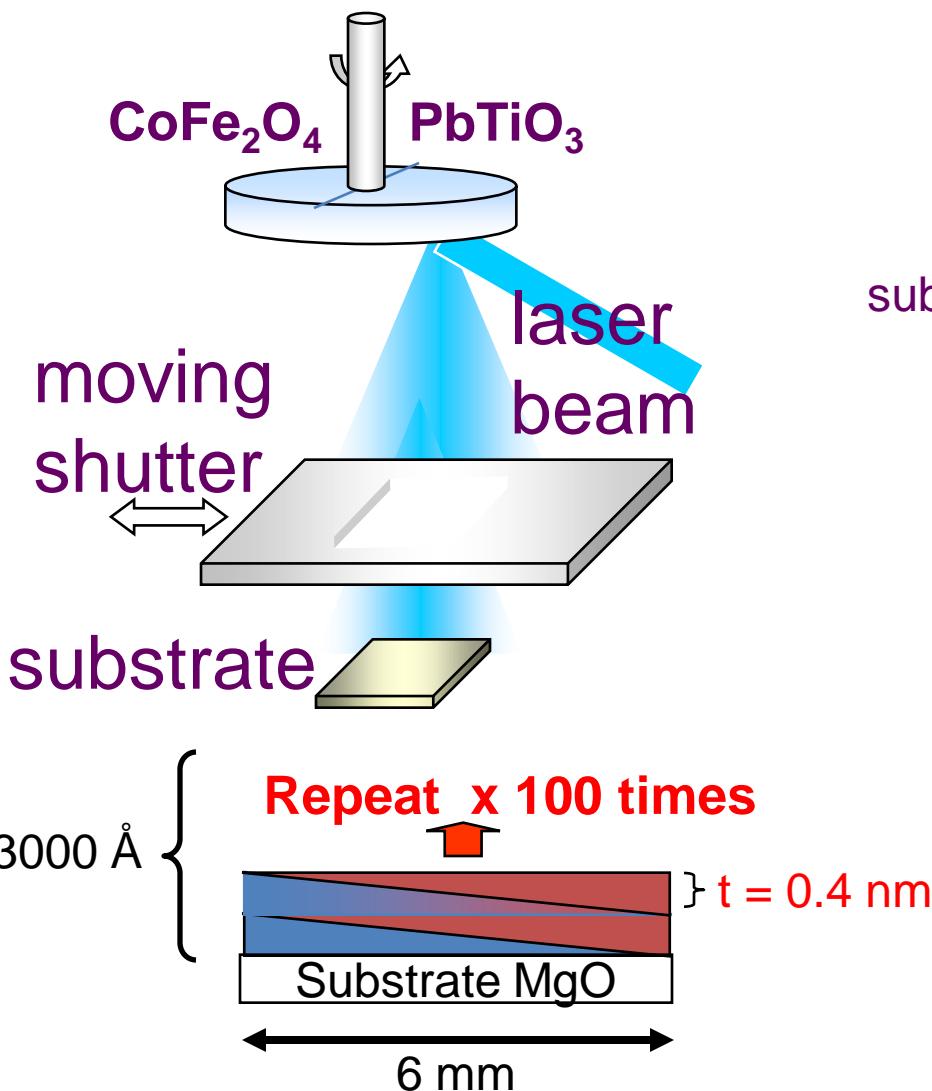
Using composition spreads to design multifunctional materials

Mixing properties: ferromagnetism, ferroelectricity, ferroelasticity, superconductivity, hydrogen storage capabilities, optical transparency, band gap, catalytic properties, various sensing, biological ..., ...



Microstructure: composites, nanocomposites, solid solutions

Fabrication of epitaxial continuous composition spread

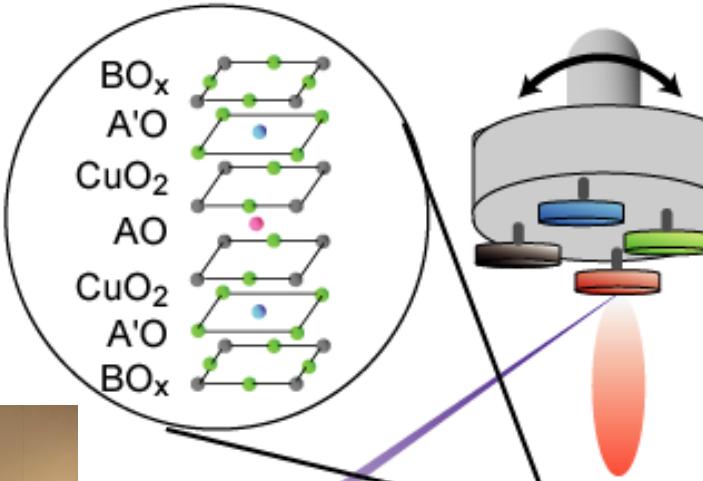


W. M. Keck Laboratory for Combinatorial Nanosynthesis and Multiscale Characterization

Ichiro Takeuchi, Gary W. Rubloff, Ellen D. Williams

Combinatorial
laser MBE for
creating arrays
of atomically
controlled crystal
structures

Atomic layer by layer deposition



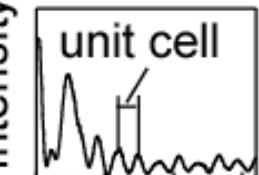
Targets

UV laser

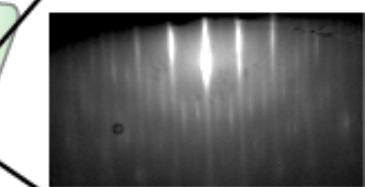
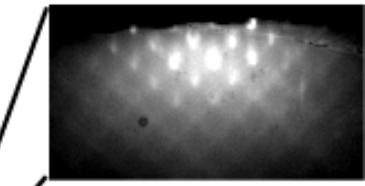
electron beam

Substrate
Shadow
masks

RHEED
oscillation



Time



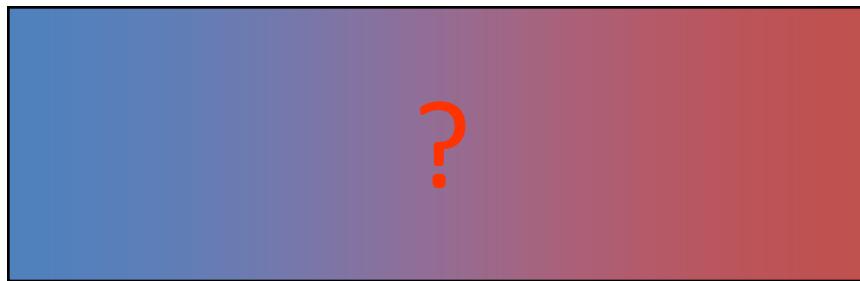
RHEED
patterns



Designing new functional materials using composition spreads

Ferroelectric-Ferromagnetic
composition Spread

Top view



Deposition on MgO substrate at 650C

Ferromagnetic
(CoFe_2O_4)

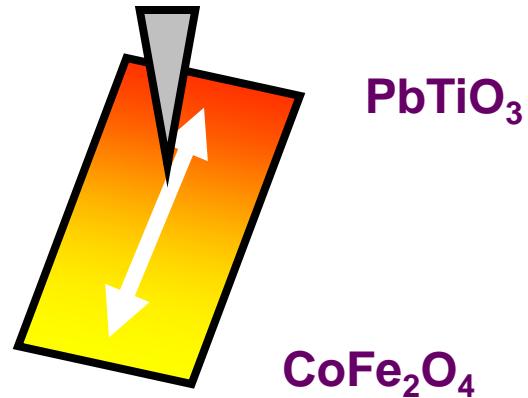
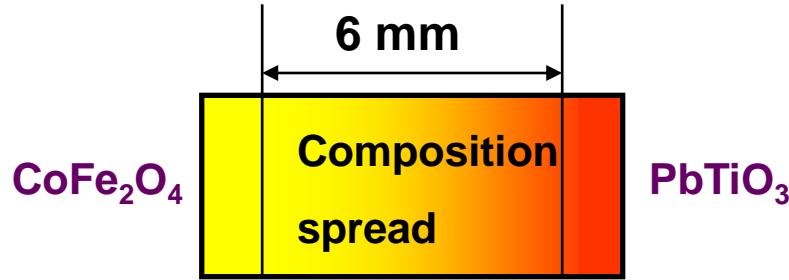
Ferroelectric
(PbTiO_3)

composition
variation

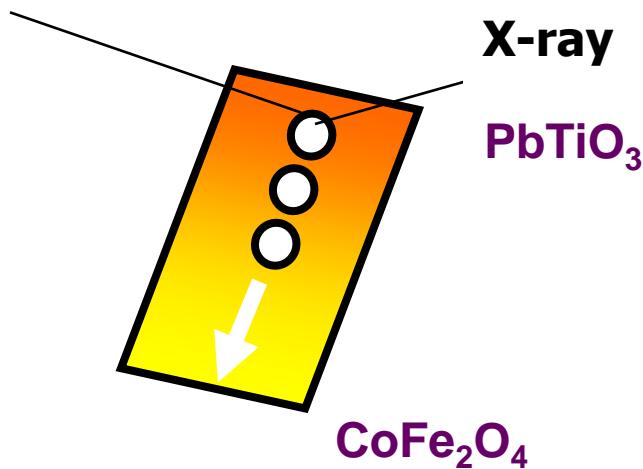
What do we get in the middle?

Composition spread characterization technique

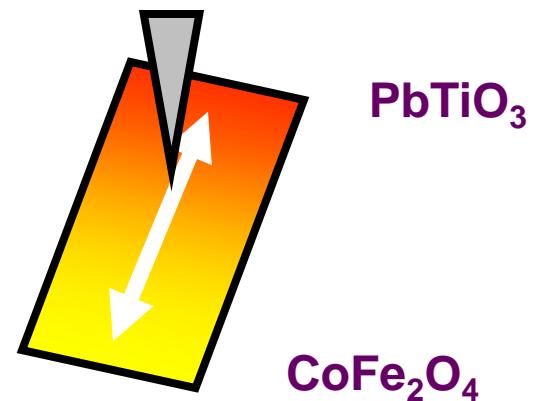
Scanning Microwave Microscope (SMM)



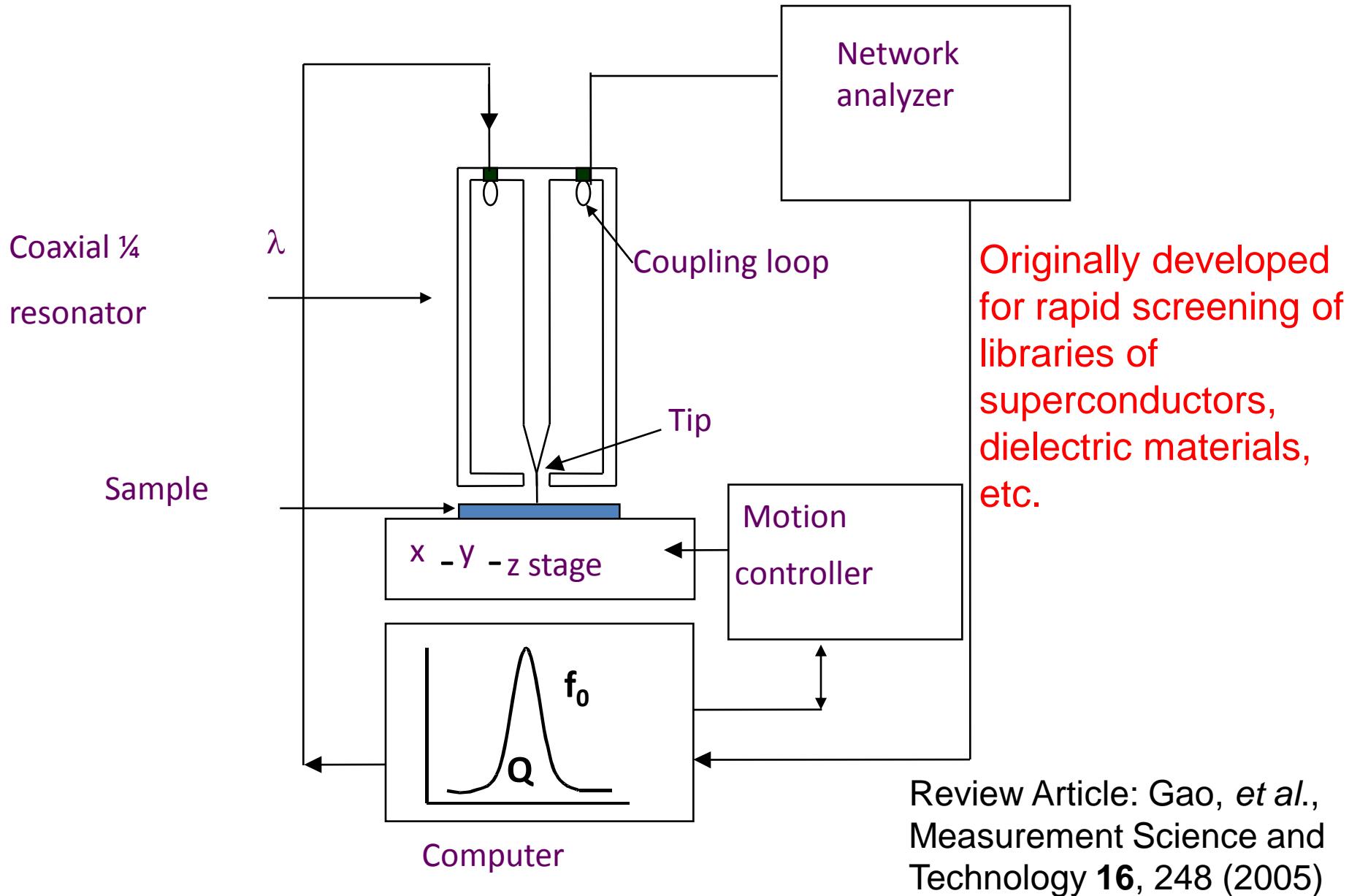
X-ray micro-diffraction



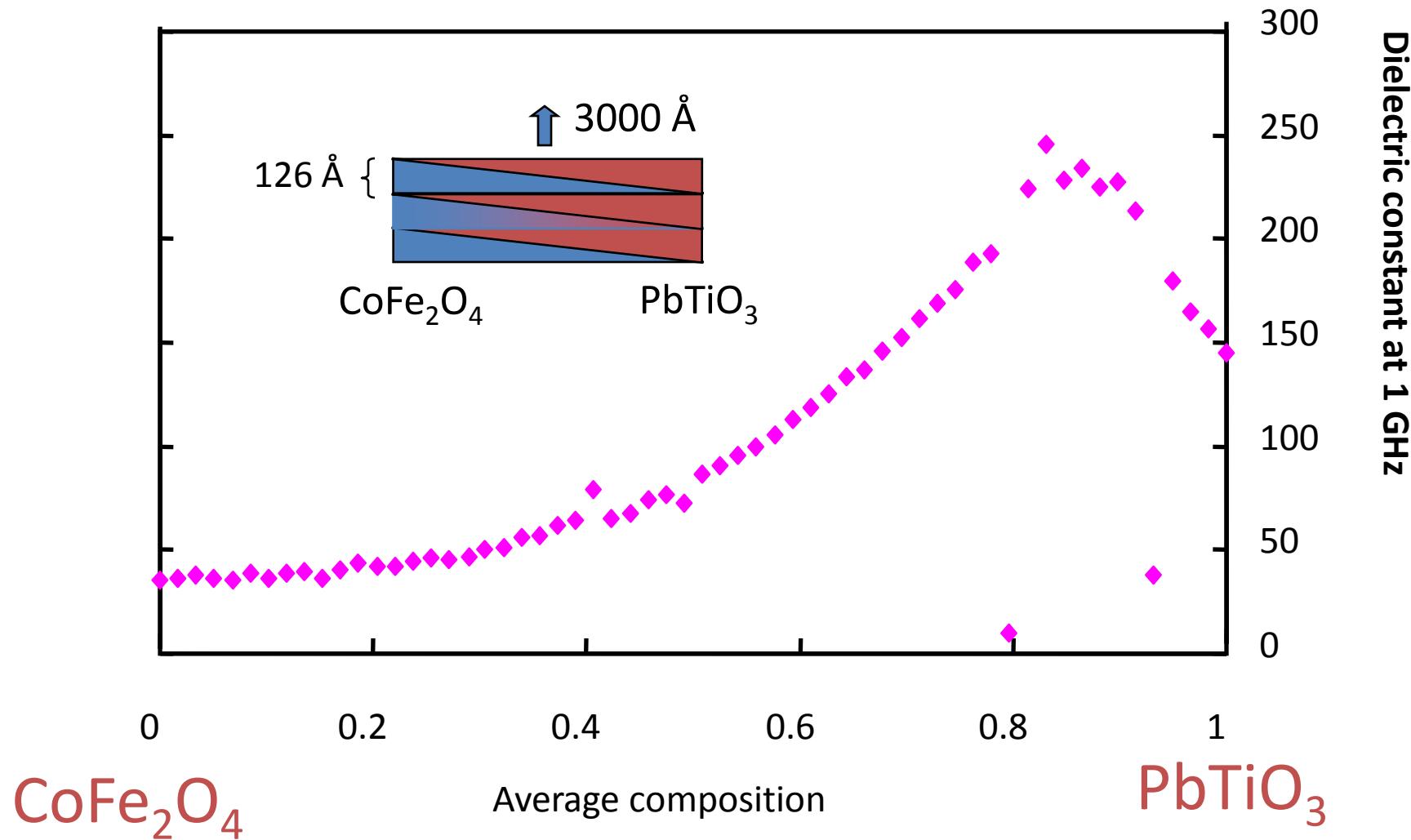
Scanning SQUID Microscope (SSM)



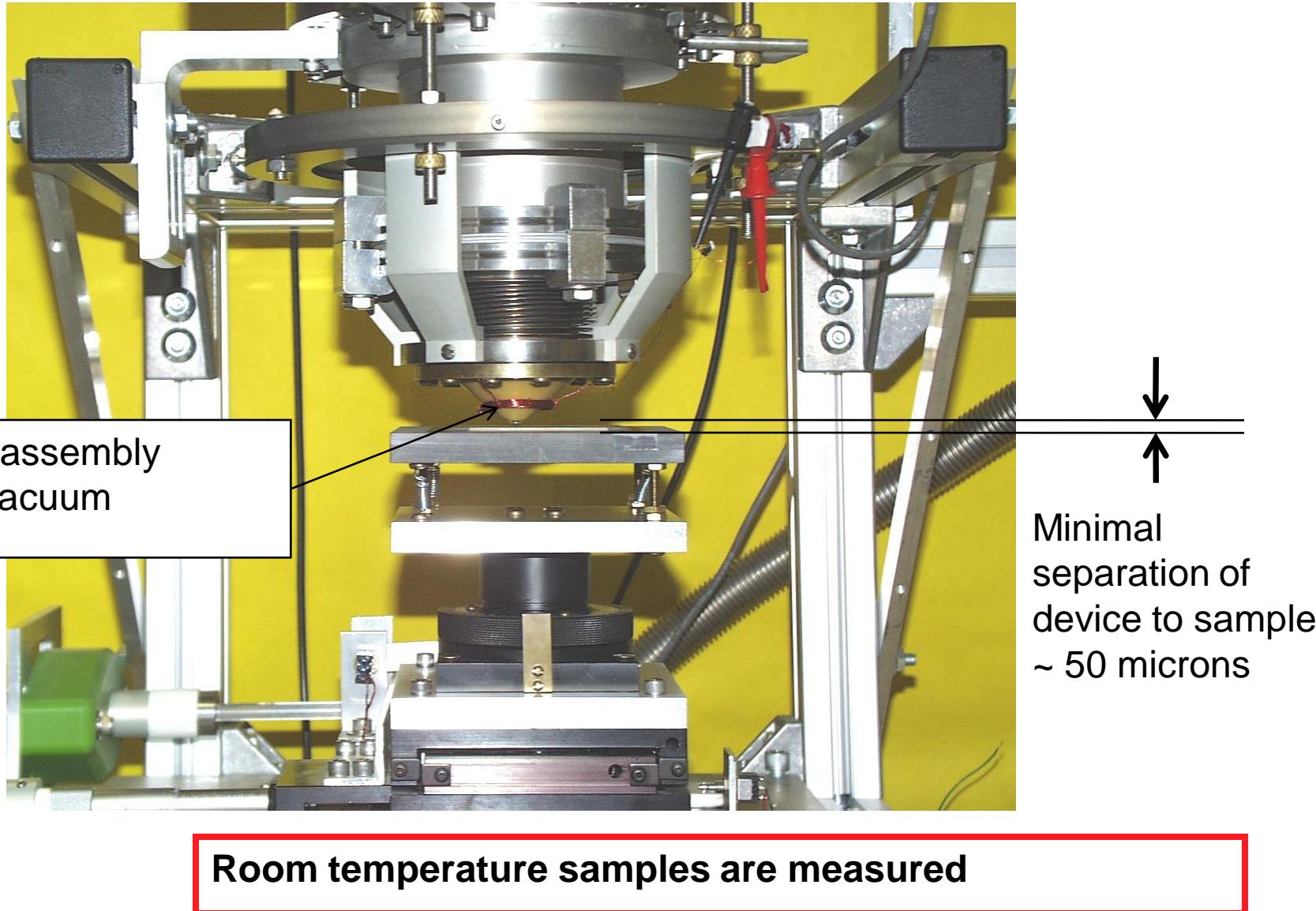
Scanning Microwave Microscope



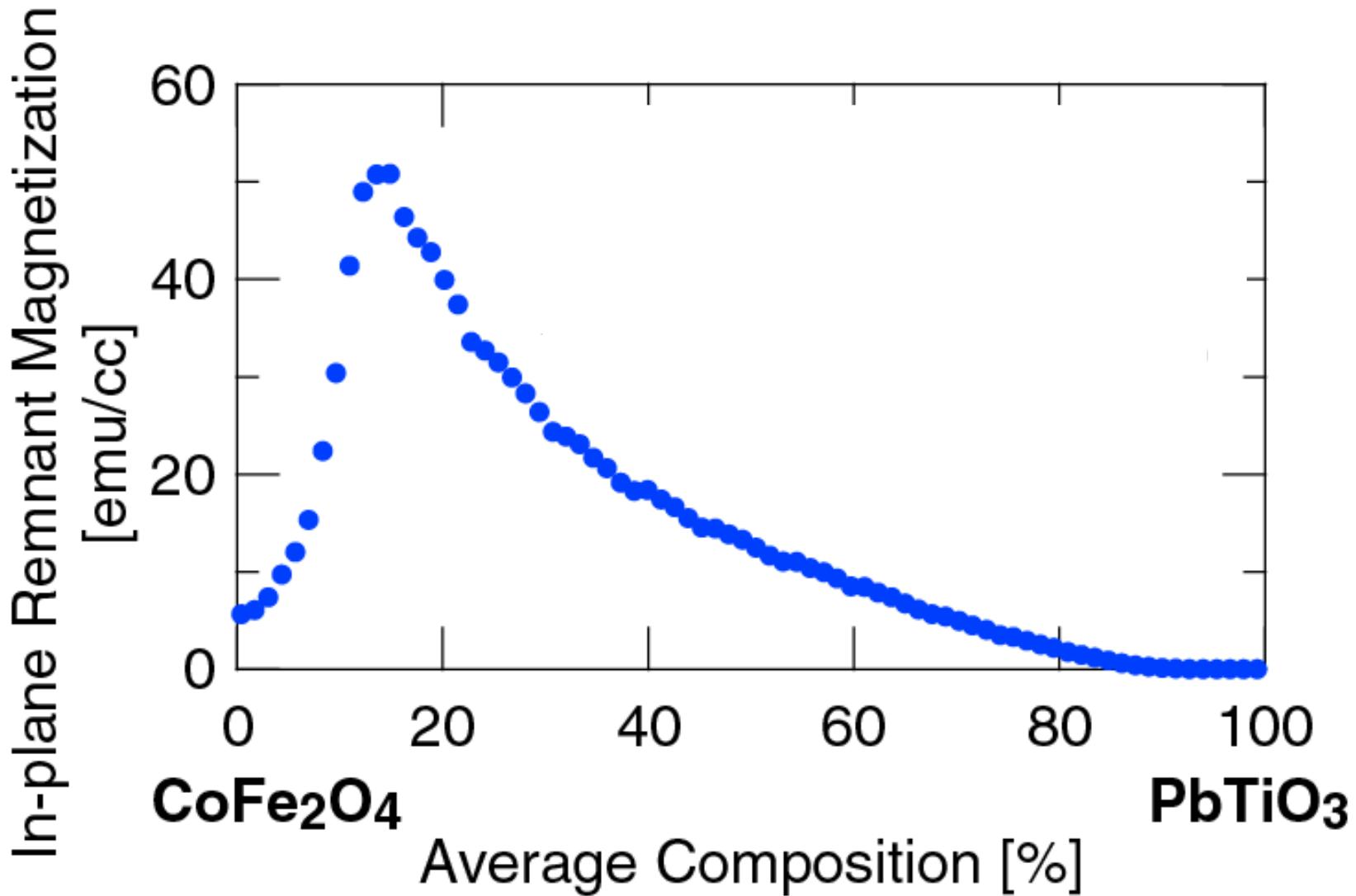
Dielectric constant characterization using microwave microscope



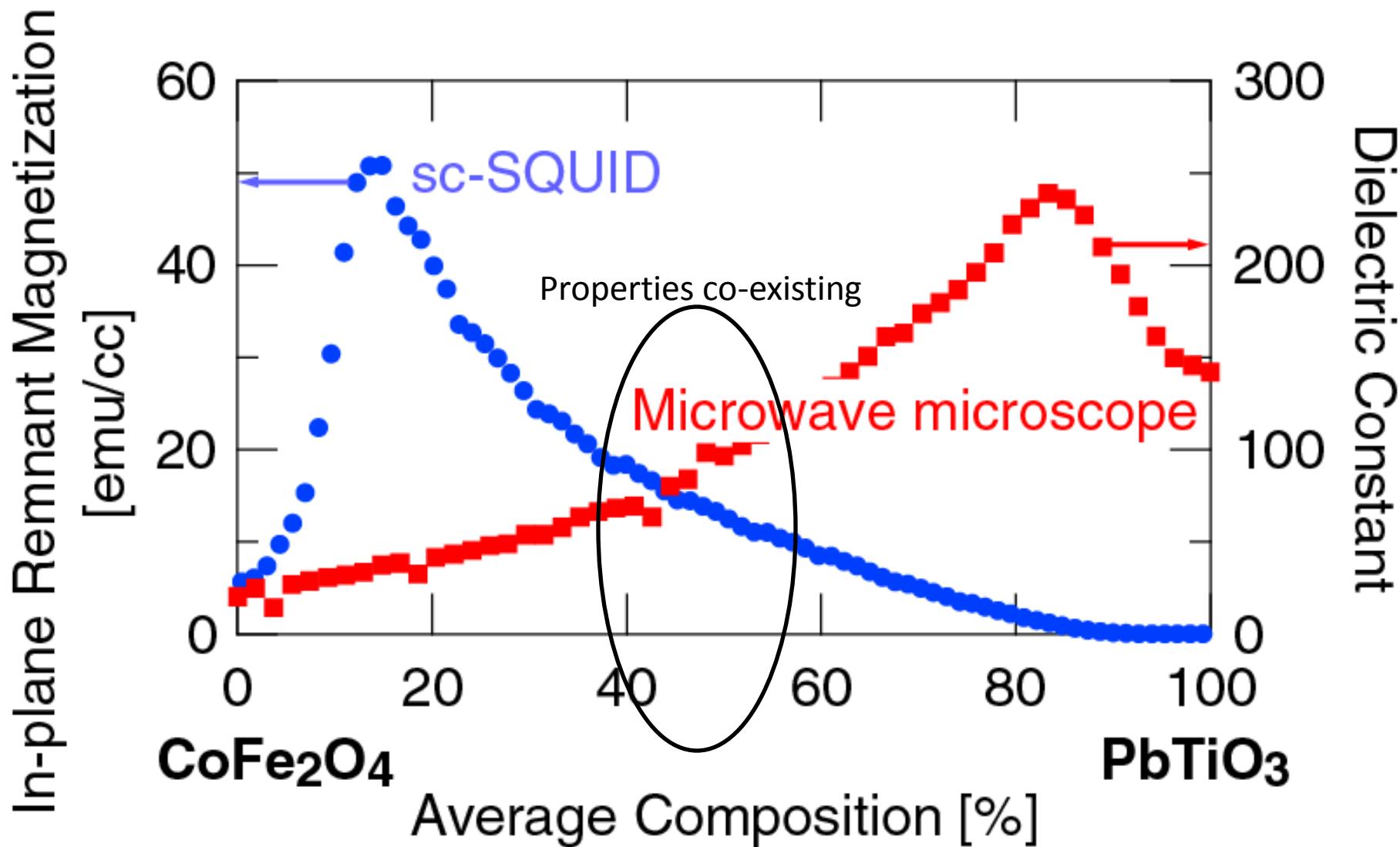
Scanning SQUID microscope based on $\text{YBa}_2\text{Cu}_3\text{O}_7$ thin film (F. Wellstood, UMD)



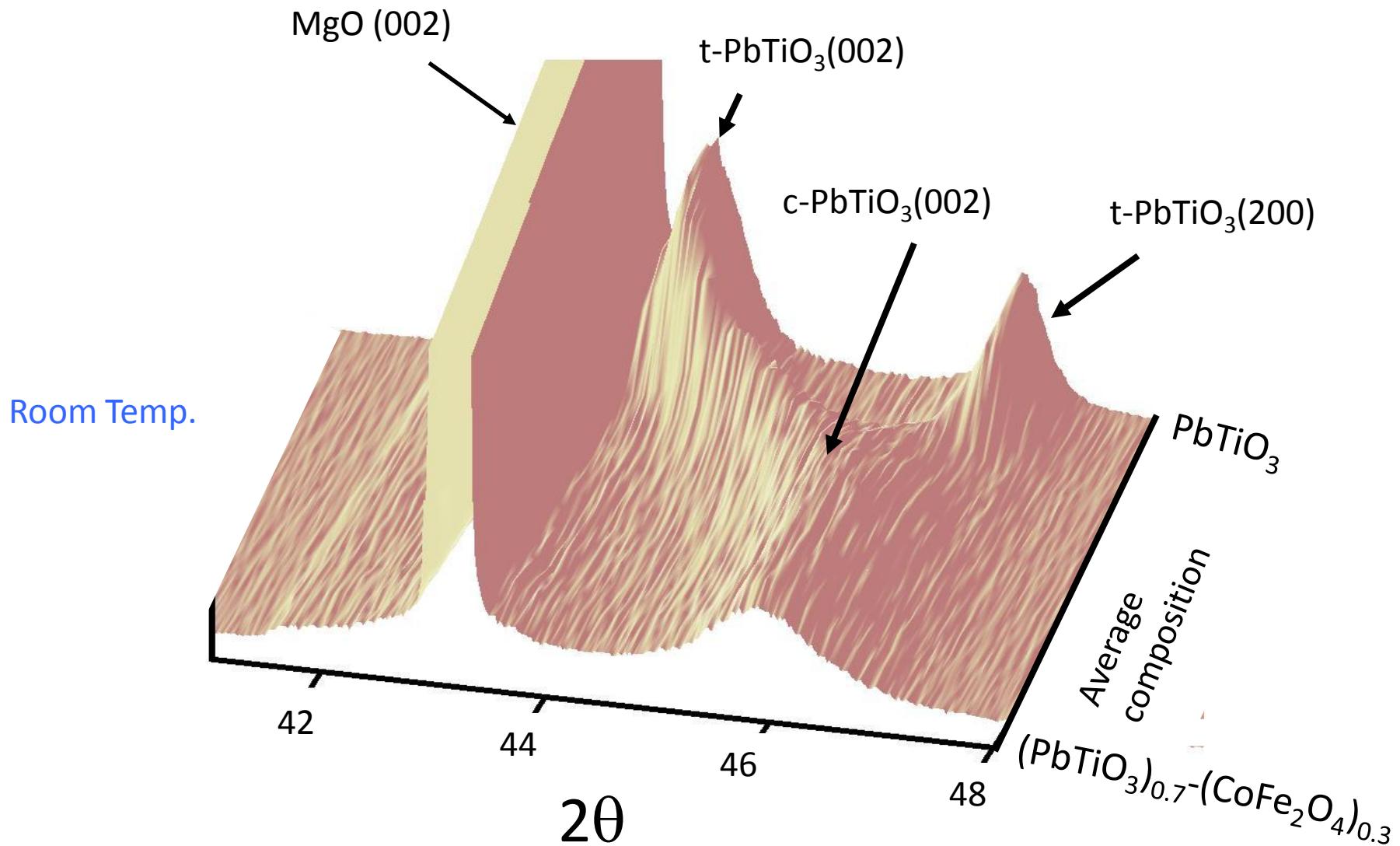
Magnetic property characterization using a scanning SQUID microscope



Magnetic and dielectric properties of $\text{PbTiO}_3\text{-CoFe}_2\text{O}_4$ system



Scanning X-ray microdiffraction



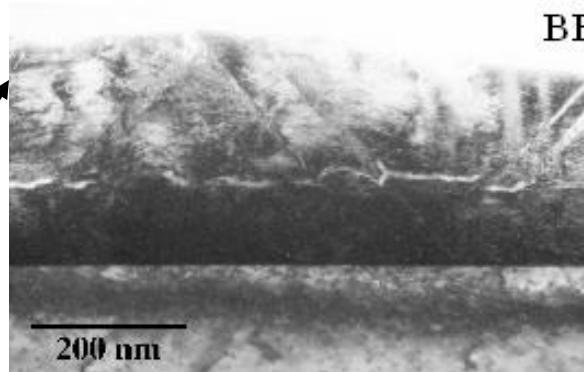
Continuous change in nanostructure across the nanocomposite spread

TEM micrographs of different compositions of $x\text{PbTiO}_3-(1-x)\text{CoFe}_2\text{O}_4$

$X=0.9$

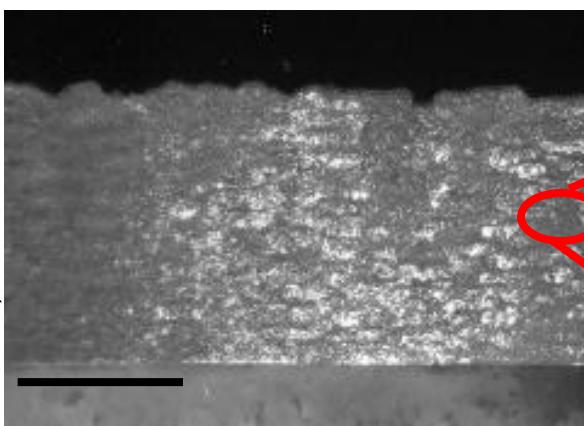
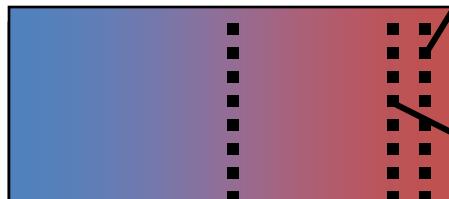
$X=0.8$

$X=0.5$



Solid solution

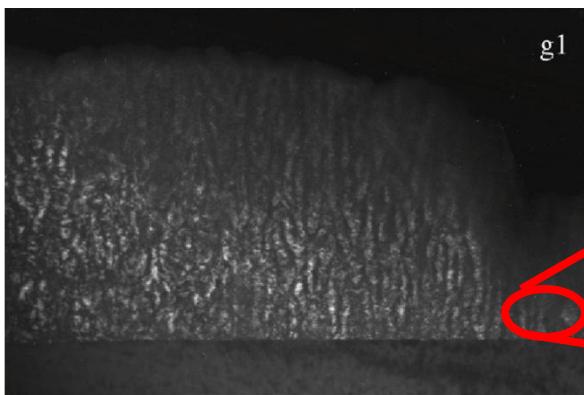
$\text{PbTiO}_3:\text{Co,Fe}$



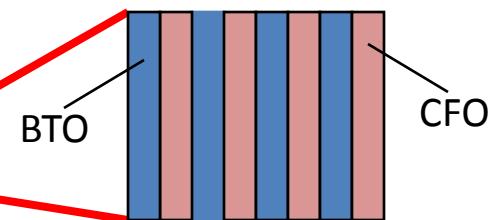
Nanopancakes

PTO

CFO

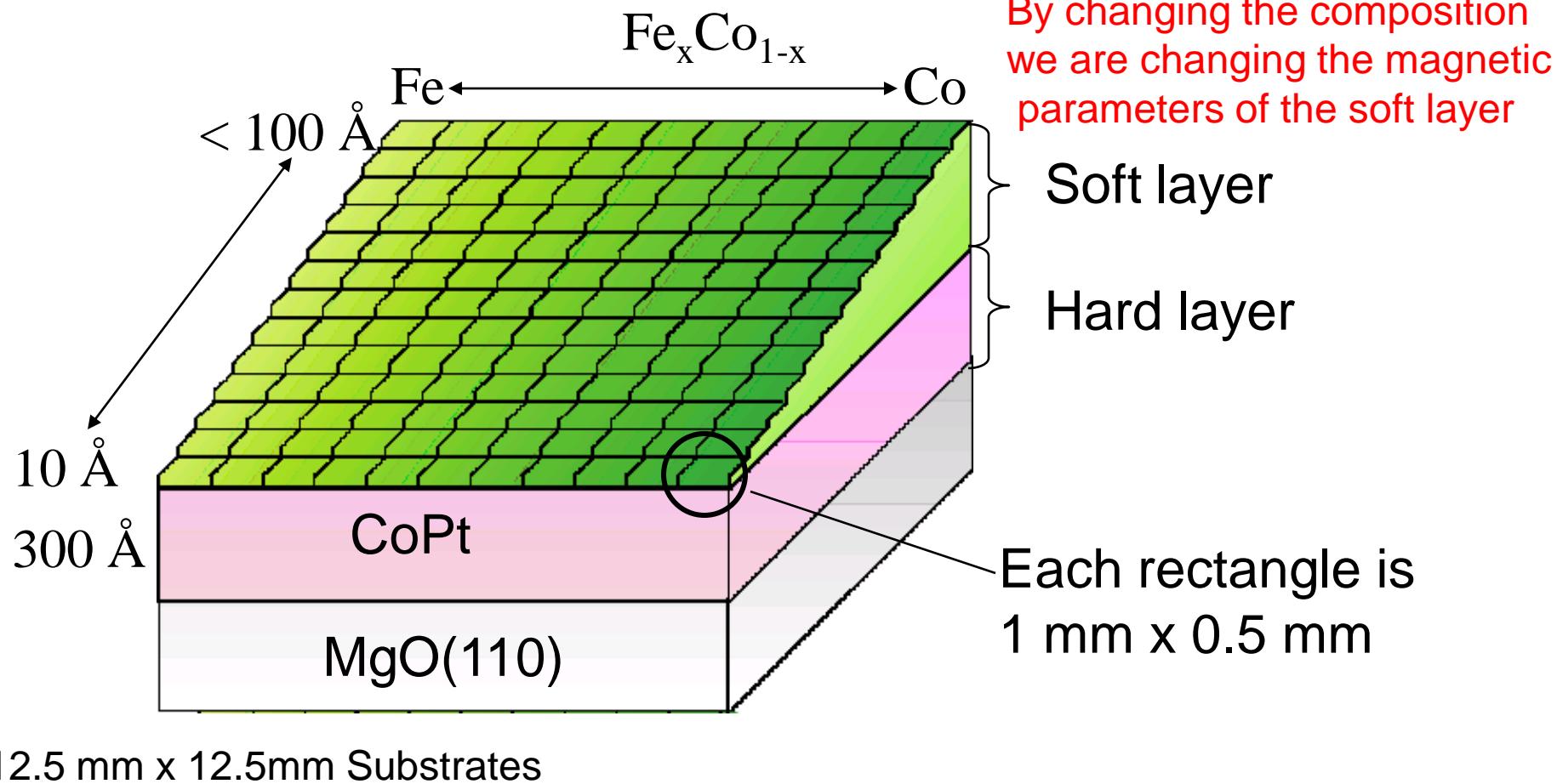


Nanopillars



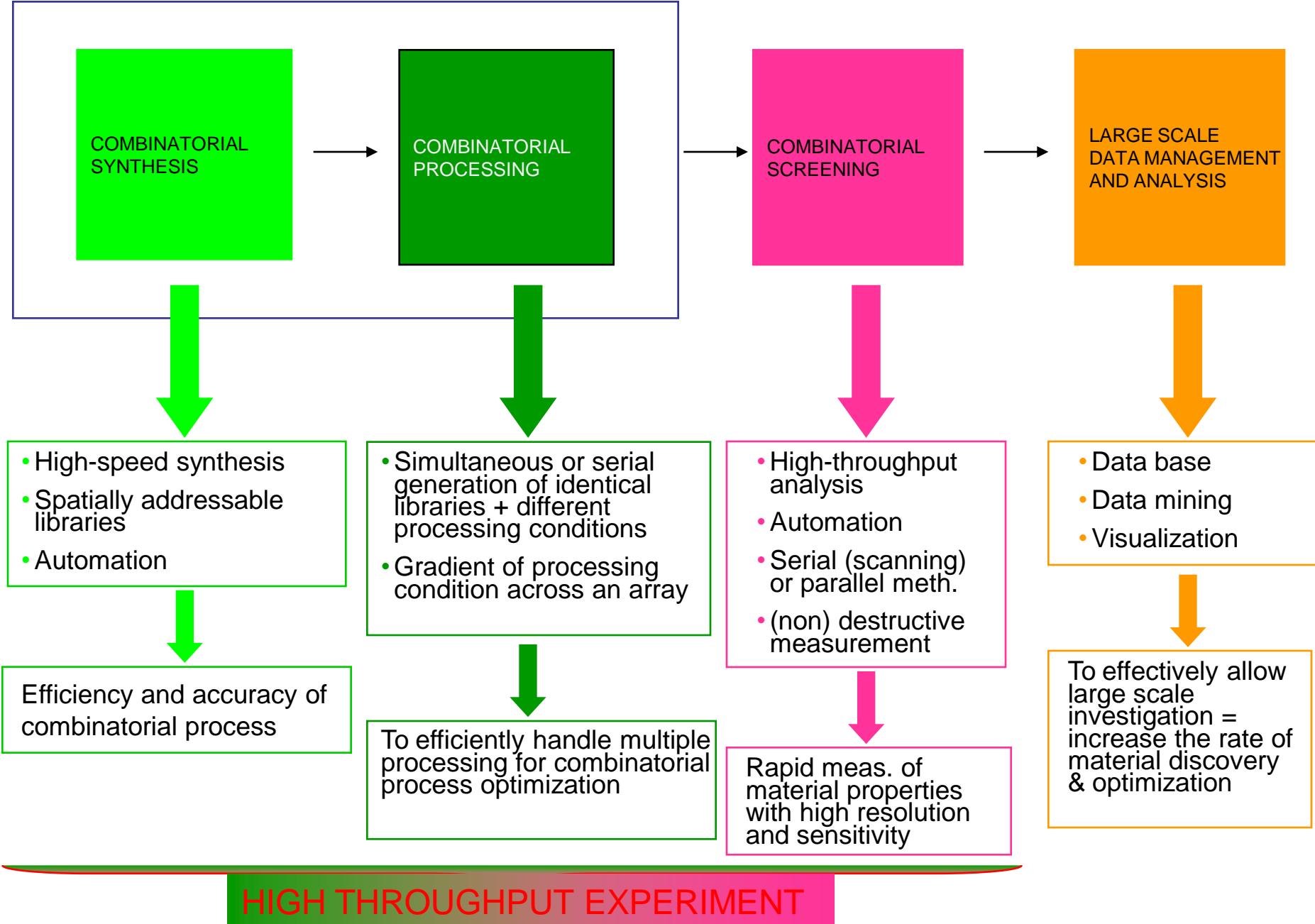
Combinatorial Library

Systematic study of the exchange coupling with small changes of the magnetic parameters



Ming-hui Yu, Jason Hattrick-Simpers, Ichiro Takeuchi, Jing Li, Z. L. Wang, J.P. Liu,
S.E. Lofland, Somdev Tyagi, J. W. Freeland, D. Giubertoni, M. Bersani, M. Anderle,
“Inter-phase Exchange Coupling in $\text{Fe}/\text{Sm-Co}$ Bilayers with Gradient Fe Thickness”
Journal of Applied Physics 98, 063908 (2005).

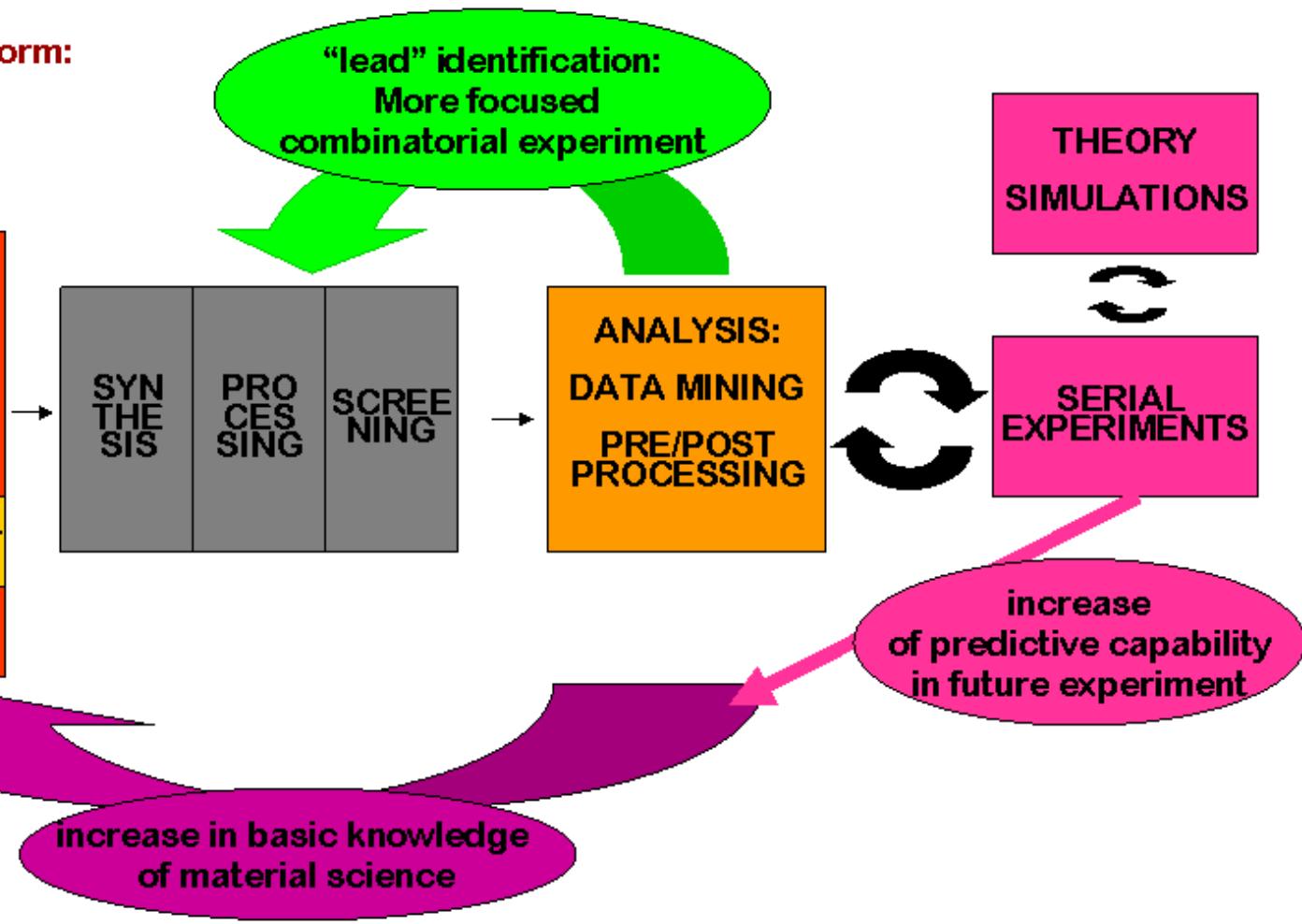
Combinatorial material experiment & analysis



The material discovery/optimization PROCESS

Technological platform:

- Experiments
- Data management
- Visualization



"Data analysis in combinatorial experiments: applying supervised principal components to predict thin film ternary composition spreads from TOF-SIMS spectra"

R. Dell'Anna, P. Lazzeri, R. Canteri, C.J. Long, J. Hattrick-Simpers, I. Takeuchi and M. Anderle

QSAR & Combinatorial Science 2008, 27, 171-178

Ringraziamenti



FONDAZIONE
BRUNO KESSLER

Giorgio Speranza, Luca Minati,

Alberto Lui, Cecilia Pederzolli

Rossana Dell'Anna, Ruben Bartali

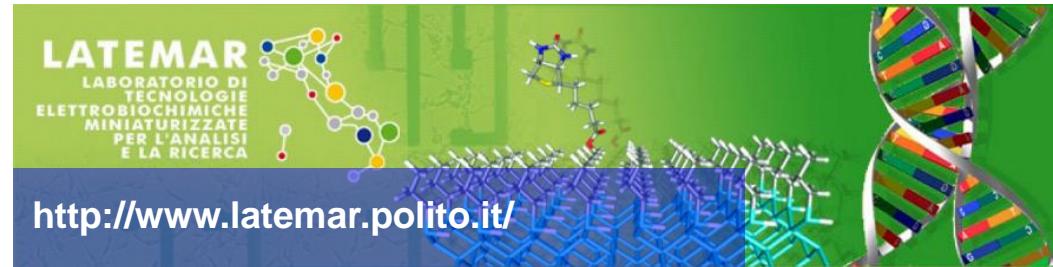
Nadhira Laidani



Elisa Molinari



Ichiro Takeuchi, Gary Rubloff, Laurent Lecordier



<http://www.latemar.polito.it/>

Fabrizio Pirri



Alberto Tagliaferro

A. JAMES CLARK
SCHOOL OF ENGINEERING